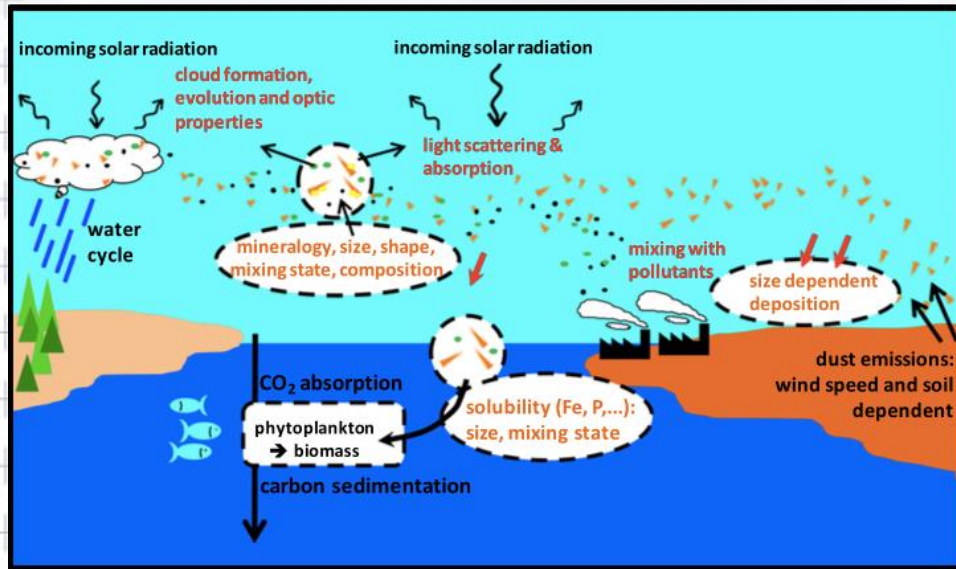




**Training workshop on Sand and Dust storms in the Arab region
10-12 Feb 2018, Cairo**

ground observation of airborne dust

Sergio Rodríguez
srodriguezg@aemet.es
AEMET, Spain



Dust and climate

- light scattering and absorption
- droplets and ice clouds formation
- clouds optical properties
- fertilization (P and Fe) of the ocean
implications on CO₂ budget



dust and health

dust

dust, aerosols and pollutants

in-situ observations:

PM_{10} and $PM_{2.5}$ levels

PM_{10} and $PM_{2.5}$ composition

complementary measurements

remote sensing observations:

column properties

altitude resolved properties

let's build our observation network !!!

dust

dust, aerosols and pollutants

in-situ observations:

- PM₁₀ and PM_{2.5} levels

- PM₁₀ and PM_{2.5} composition

- complementary measurements

remote sensing observations:

- column properties

- altitude resolved properties

let's build our observation network !!!

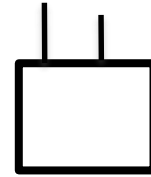


people live in cities and breath a cocktail dust + pollutants





people live in cities and breath a cocktail dust + pollutants



dust - air quality stations



parameters indicative of:

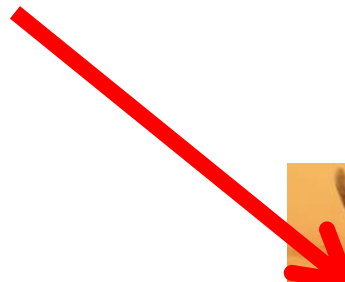
dust
ambient air quality



people live in cities and breath a cocktail dust + pollutants

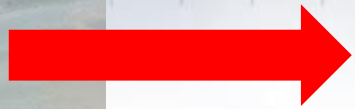
what is dust?

type of dust sources ?



types of dust sources:

desert dust

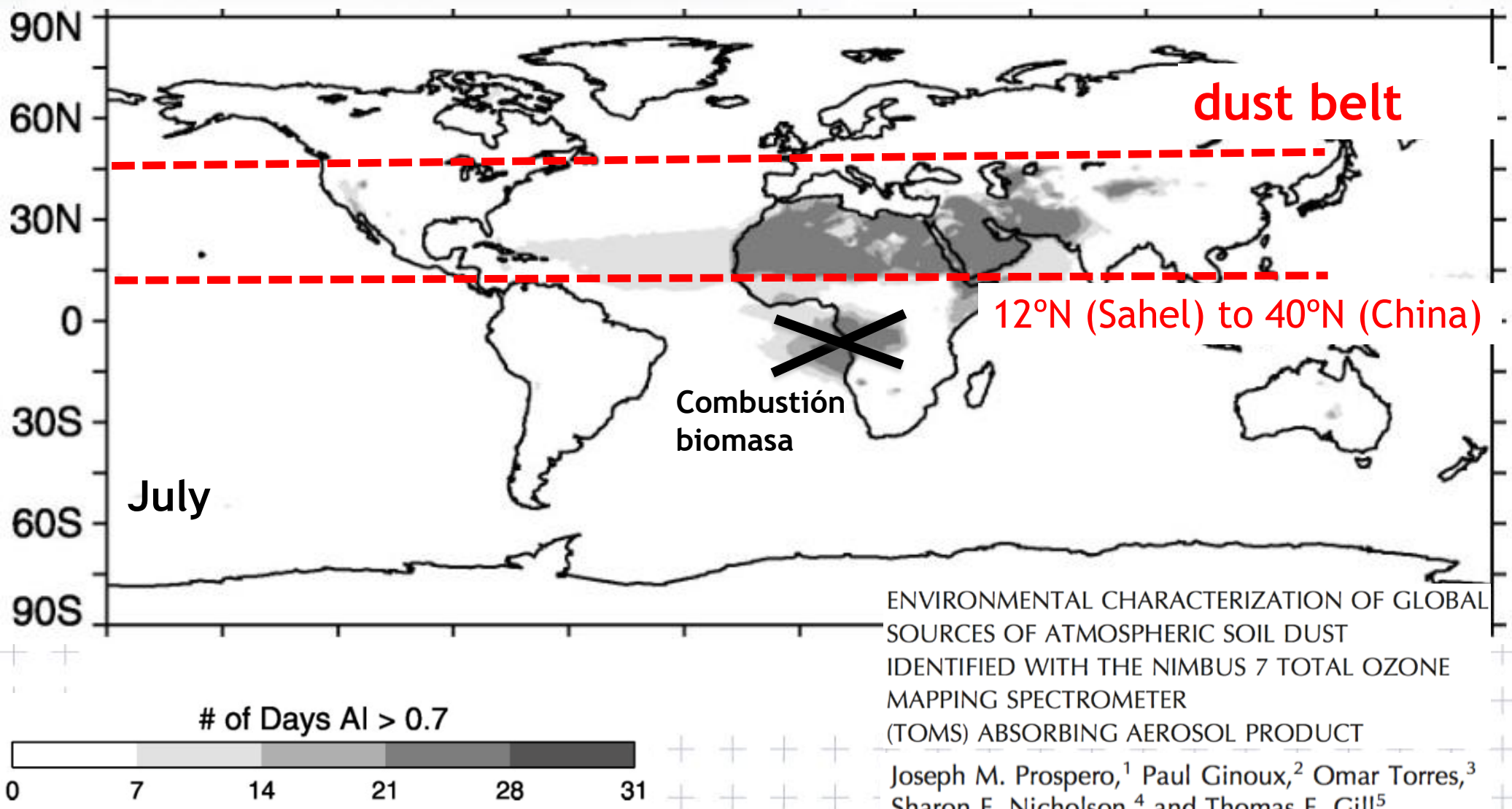


people live in cities and breath a cocktail dust + pollutants

what is this dust?
sources?

UV absorbing aerosols - dust Satellite

What is dust?



ENVIRONMENTAL CHARACTERIZATION OF GLOBAL SOURCES OF ATMOSPHERIC SOIL DUST IDENTIFIED WITH THE NIMBUS 7 TOTAL OZONE MAPPING SPECTROMETER (TOMS) ABSORBING AEROSOL PRODUCT

Joseph M. Prospero,¹ Paul Ginoux,² Omar Torres,³ Sharon E. Nicholson,⁴ and Thomas E. Gill⁵

desert dust

chotts, sabkhas, wadis, salares

1. what is dust ?

There are several types of sources, but the mayor dust sources are associate with dry lakes/rivers beds

2. chemistry and mineralogy

clays, feldspars, oxides, evaporites

Si, Al, Ca, Fe, Mg, Na, Cl, Mn....

Table 6. Density and Real Index of Refraction of Minerals Found in Saharan Dust^a

1. clays

illite	$K_{0.6}(H_3O)_{0.4}Al_{1.3}Mg_{0.3}Fe_{0.1}Si_{3.5}O_{10}(OH)_2 \cdot (H_2O)$
kaolinite	$Al_2Si_2O_5(OH)_4$
montmorillonite	$(Na,Ca)_{0.5}(Al,Mg,Fe)_4(Si,Al)_8O_{20}(OH)_4 \cdot n(H_2O)$
smectite	$(Na,Ca)Al_4(Si,Al)_8O_{20}(OH)_4 \cdot 2(H_2O)$
chlorite	$Na_{0.5}(Al,Mg)_6(Si,Al)_8O_{18}(OH)_{12} \cdot 5(H_2O)$

2. evaporites

calcite	$CaCO_3$
dolomite	$CaMg(CO_3)_2$
gypsum	$CaSO_4 \cdot 2(H_2O)$
anhydrite	$CaSO_4$
halite	$NaCl$

4. oxides

hematite	Fe_2O_3
goethite	$FeO(OH)$
rutile	TiO_2

3. feldspars

microcline	$KAlSi_3O_8$	Var oligoclase	$(Na,Ca)(Si,Al)_4O_8$
		Var albite	$NaAlSi_3O_8$
		Var anorthite	$CaAl_2Si_2O_8$

Characterization of African dust transported to Puerto Rico by individual particle and size segregated bulk analysis

Elizabeth A. Reid,^{1,2,3} Jeffrey S. Reid,³ Michael M. Meier,⁴ Michael R. Dunlap,⁴ Steven S. Cliff,⁴ Aaron Broumas,⁴ Kevin Perry,⁵ and Hal Maring⁶

desert dust

chotts, sabkhas, wadis, salares

1. what is dust ?

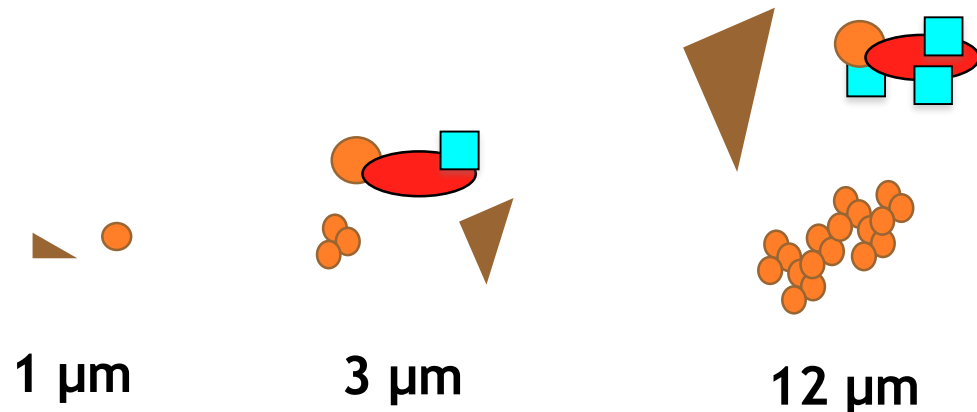
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clays, feldspars, oxides, evaporites

3. Size and morphology

1 and 20 μm
aggregates



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Chotts, Sabkhas



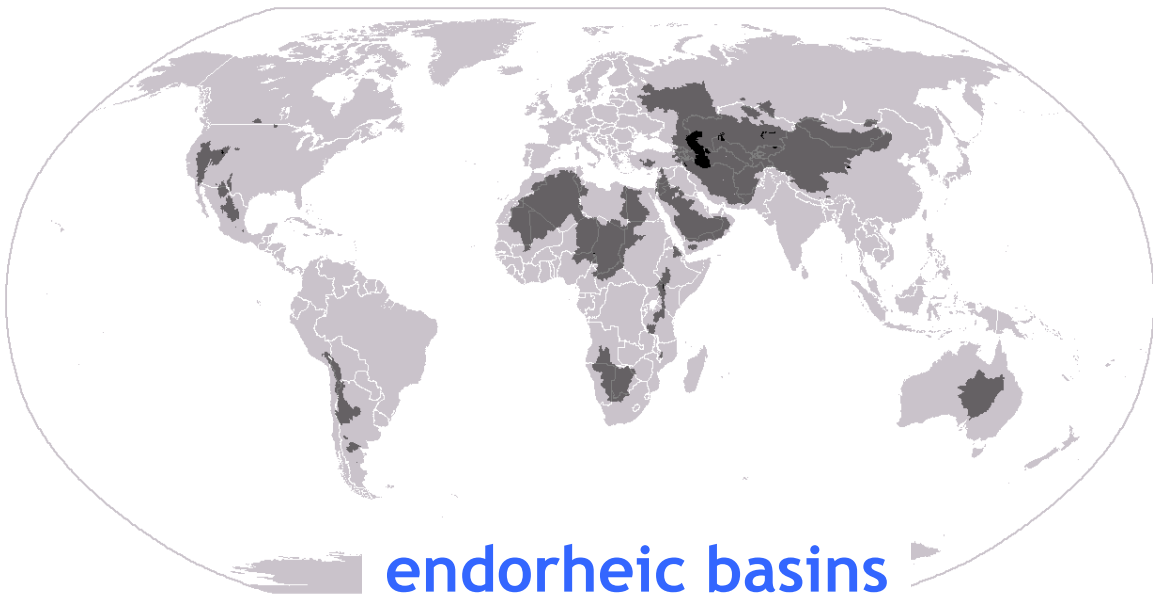
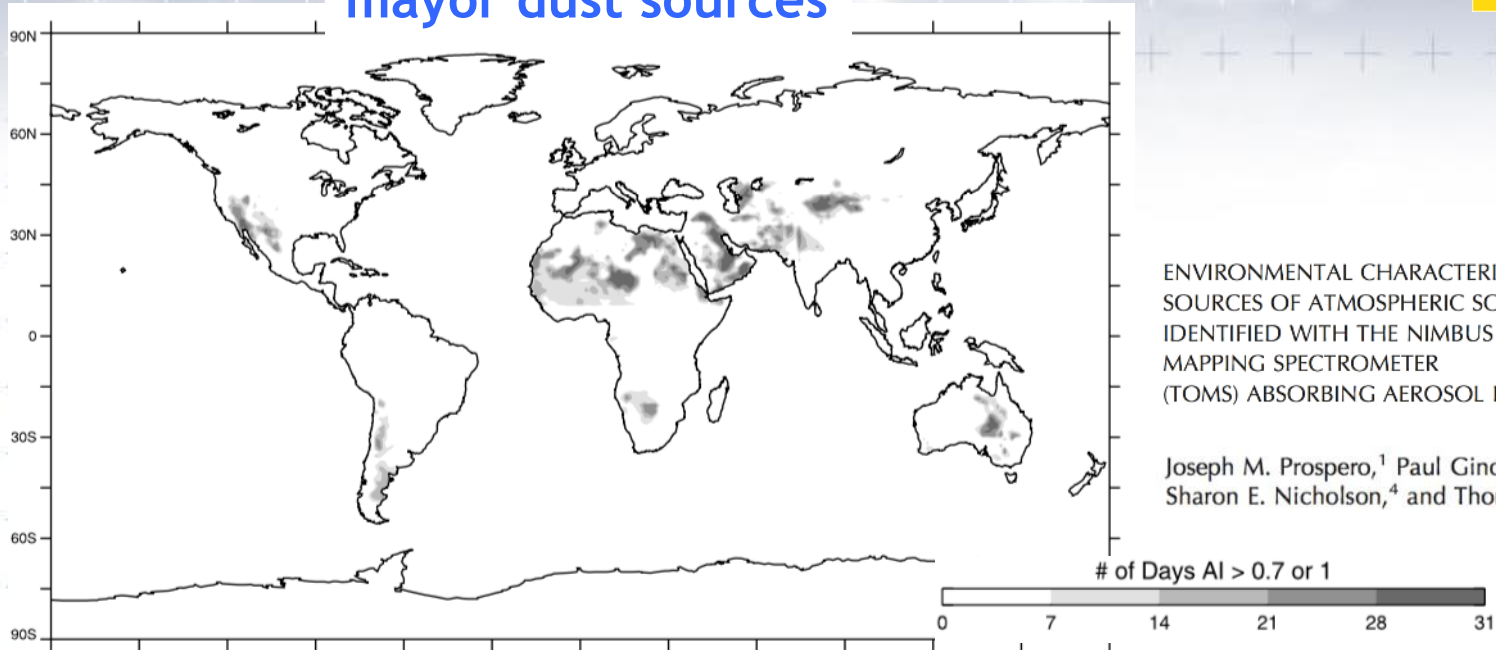
Dry lakes beds



wadis

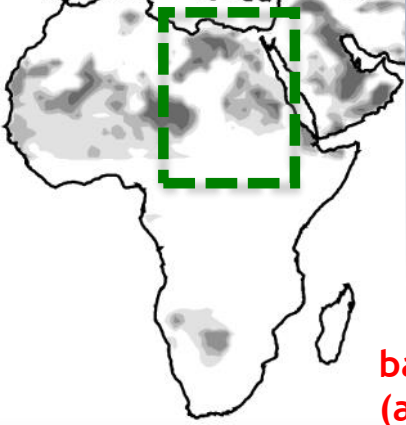
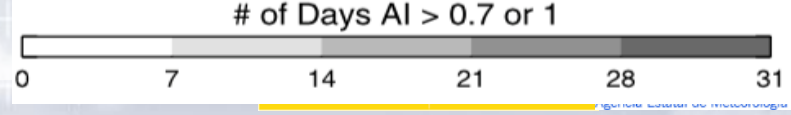


mayor dust sources



endorheic basins
no conection to sea

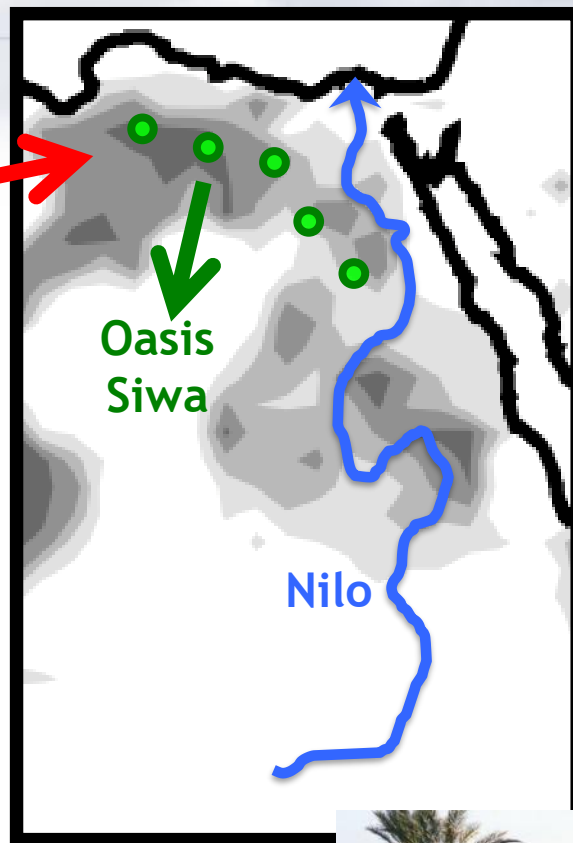
Detección satélite

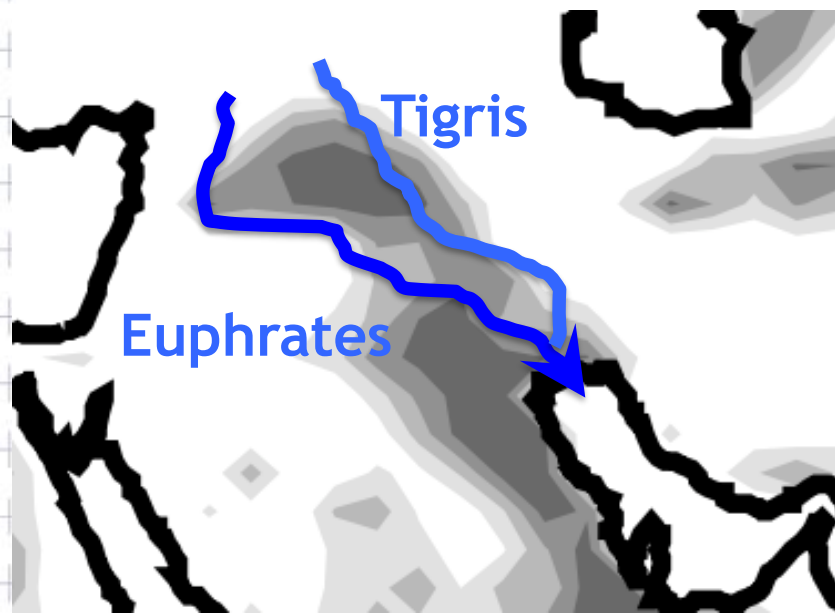
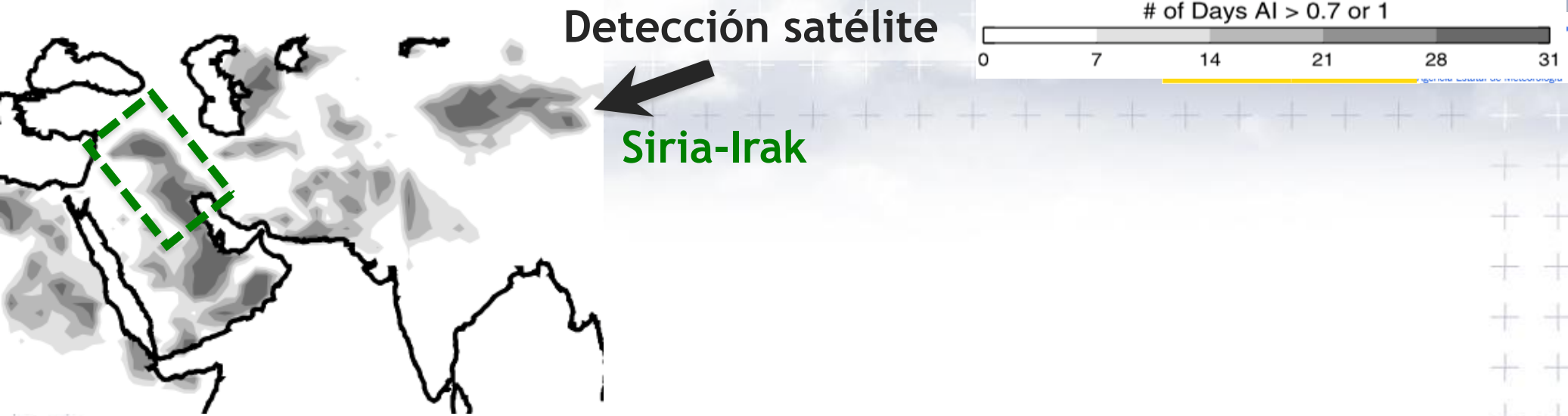


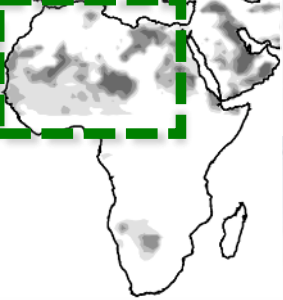
bajas topográficas
(altitudes -)

● lakes, oasis,
cultivation,
underground water,
Ancient rivers

Oasis Siwa
29°13'N, 25°31'E





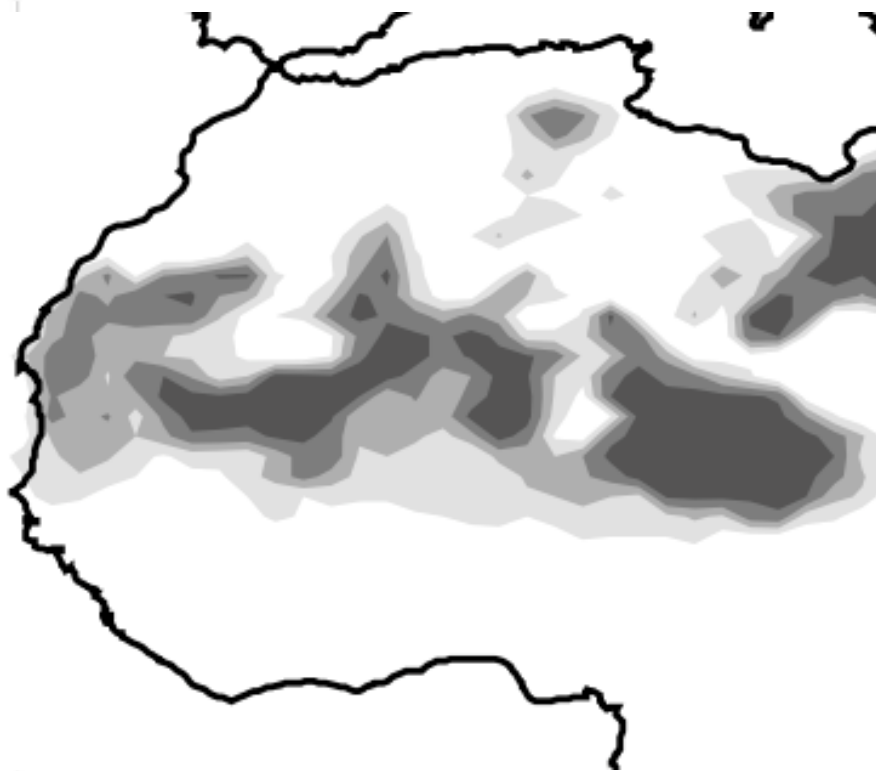


Sahara
Sahel



MINISTERIO
DE MEDIO AMBIENTE
Y MEDIO RURAL Y MARINO

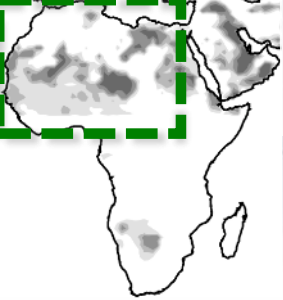
AEMet
Agencia Estatal de Meteorología



Detección satélite

of Days AI > 0.7 or 1



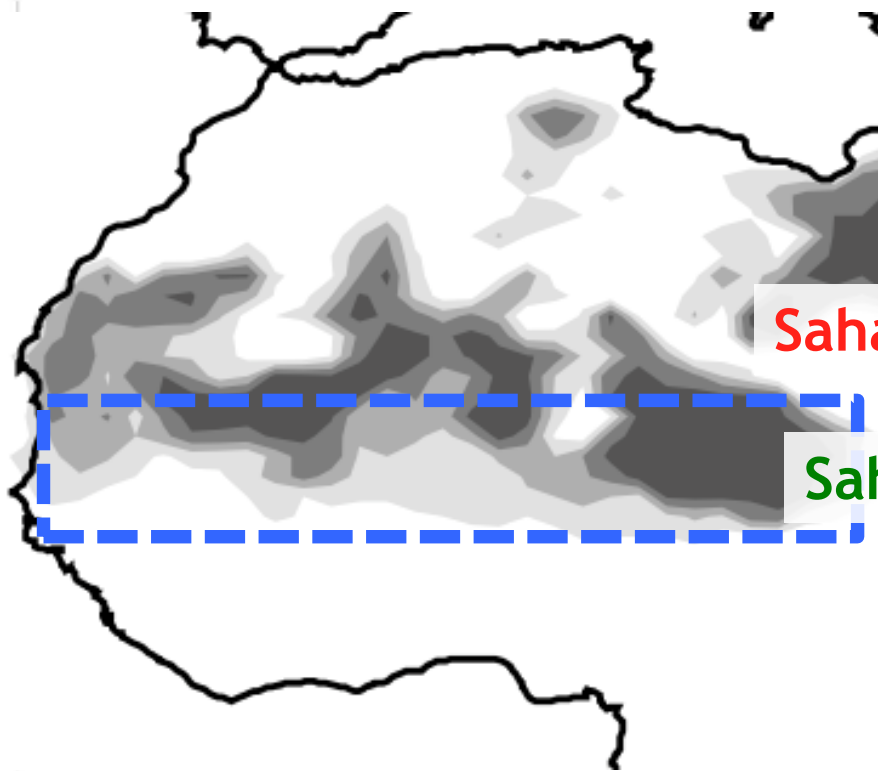


Sahara
Sahel



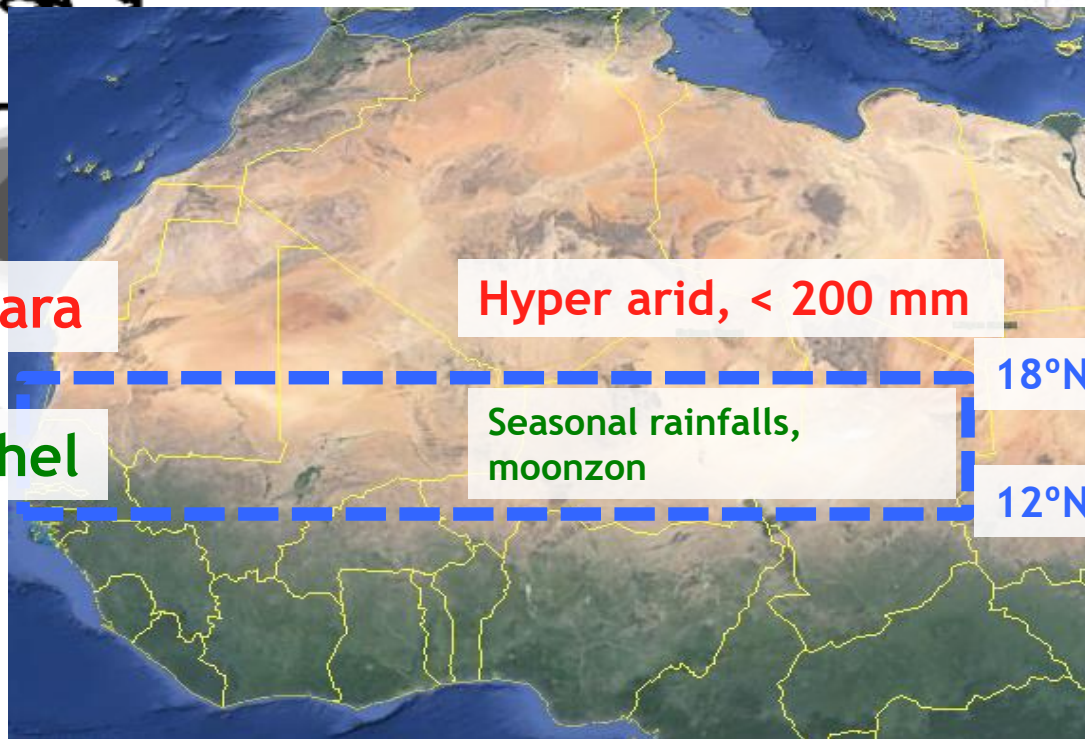
MINISTERIO
DE MEDIO AMBIENTE
Y MEDIO RURAL Y MARINO

AEMet
Agencia Estatal de Meteorología



Sahara

Sahel



Hyper arid, < 200 mm

Seasonal rainfalls,
monsoon

18°N

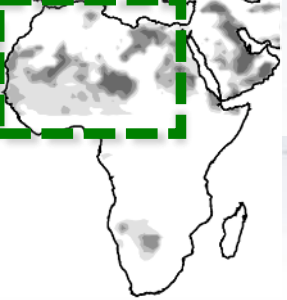
12°N



Detección satélite

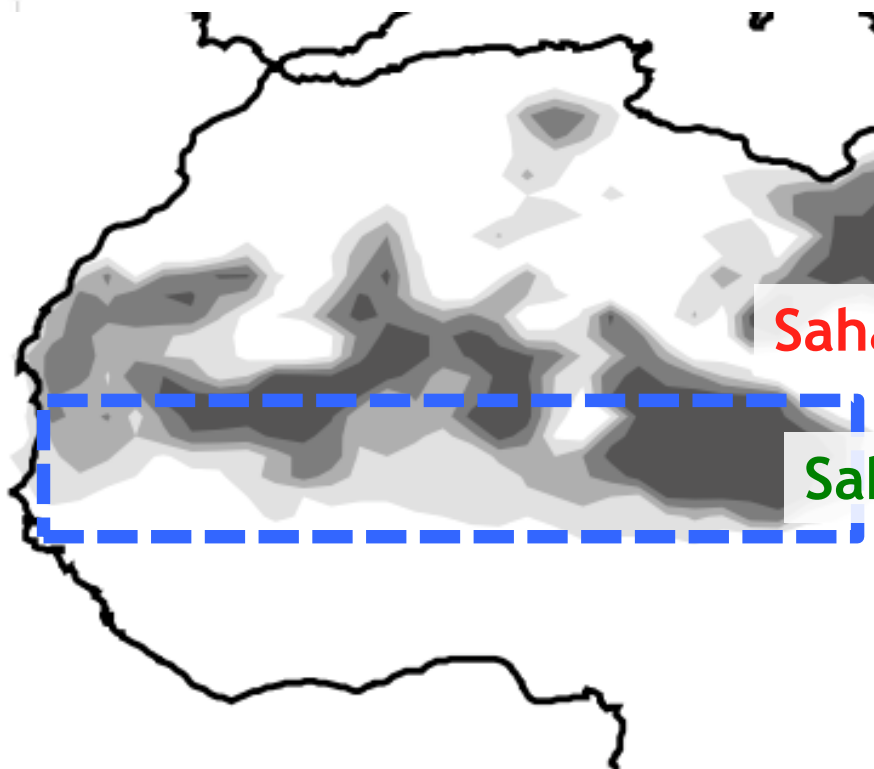
of Days AI > 0.7 or 1



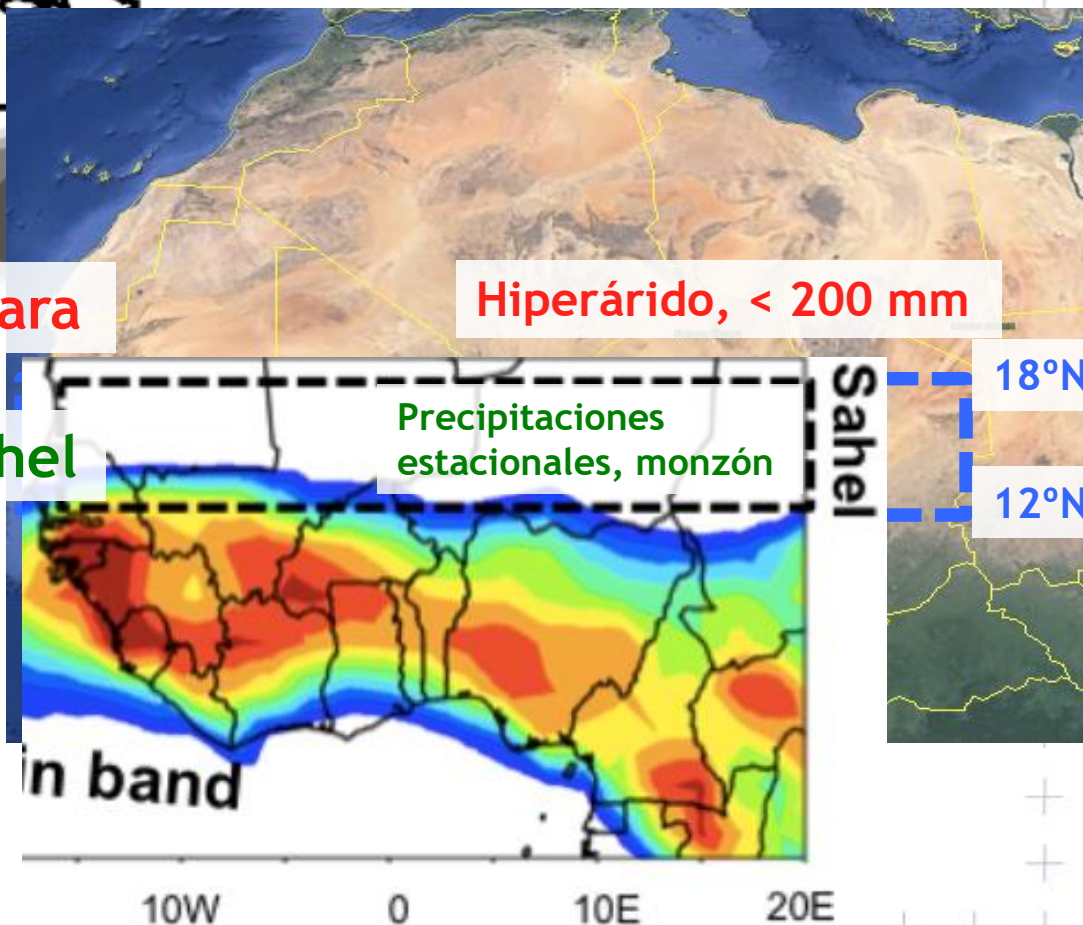
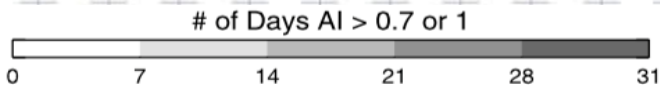


Sahara
Sahel

Sahel



Detección satélite



Sahara

Hiperárido, < 200 mm

Sahel

Precipitaciones estacionales, monzón

Sahel

18°N

12°N

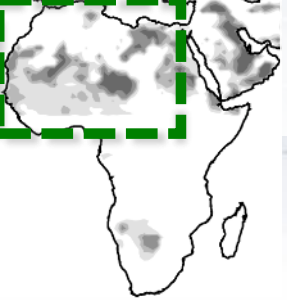
in band

10W

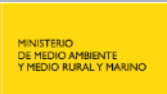
0

10E

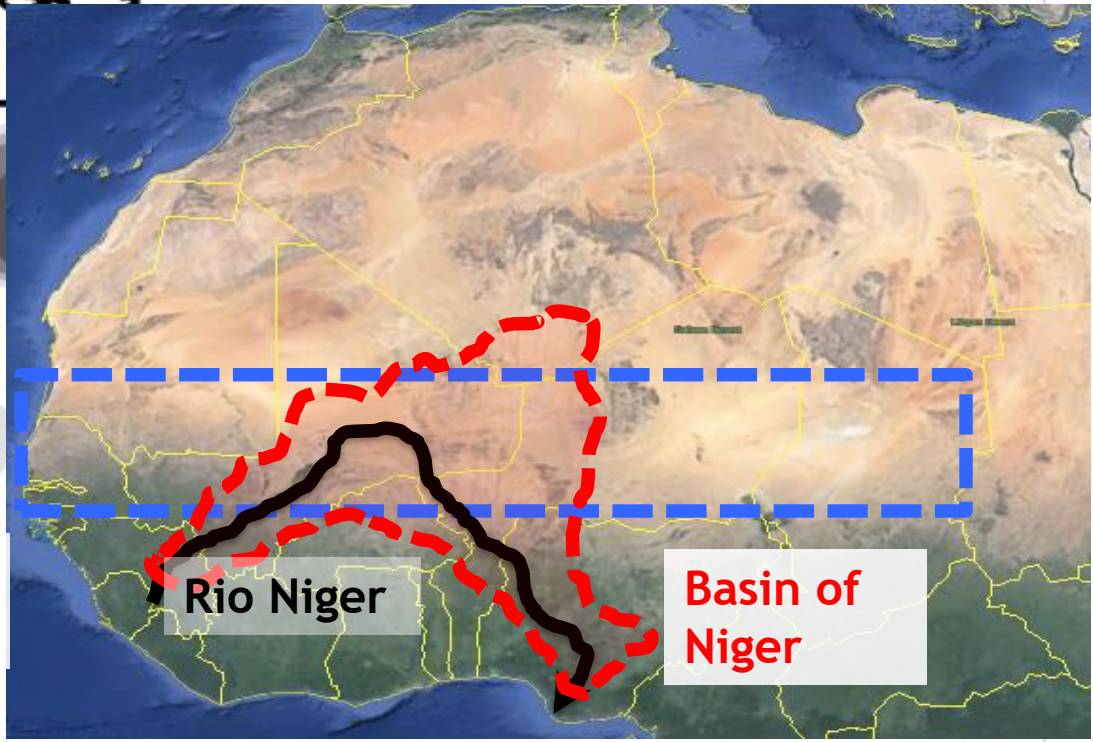
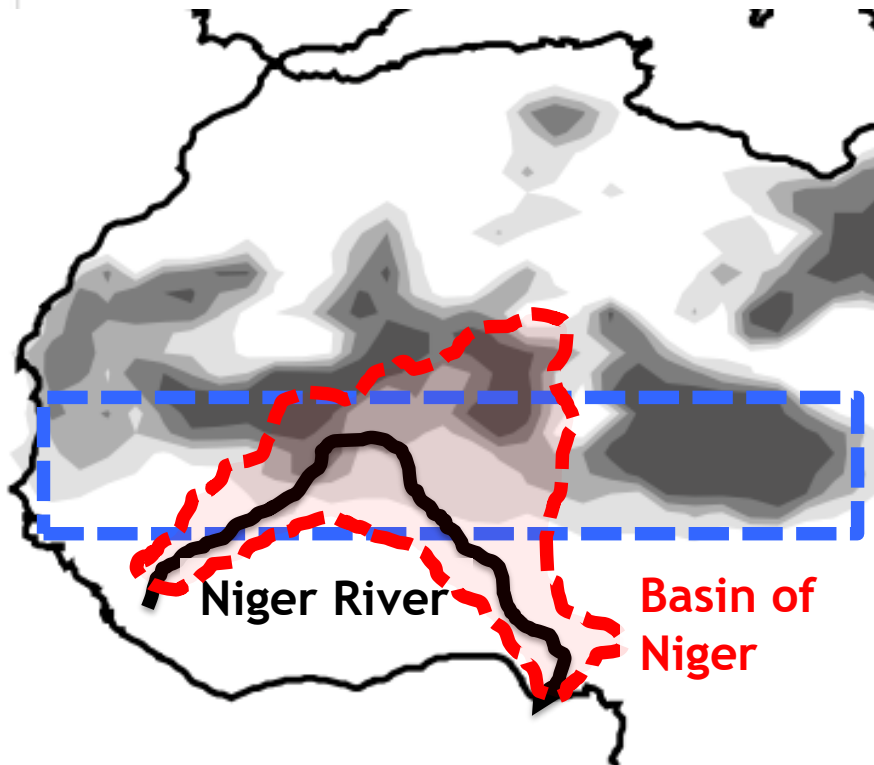
20E



Sahara
Sahel

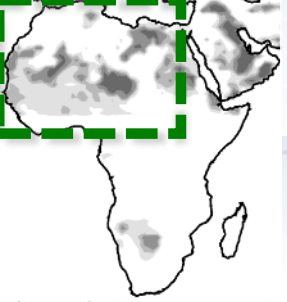


Sahel

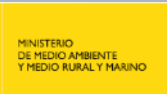


Remote Sensing
satellite

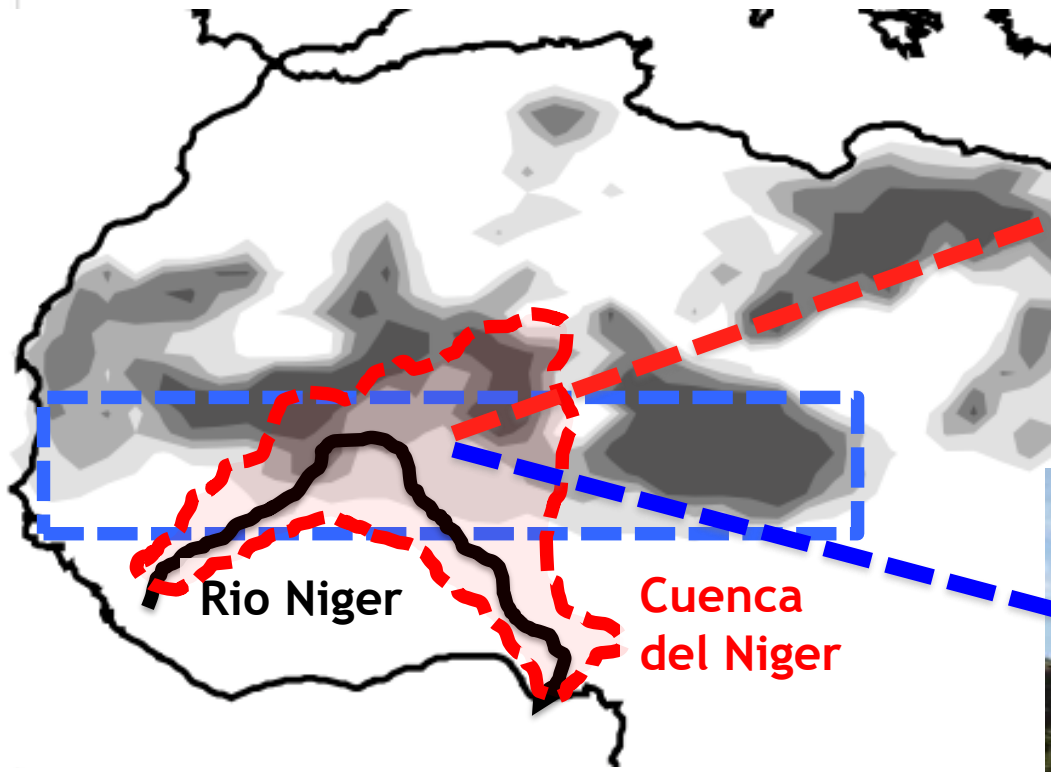




Sahara
Sahel



Sahel



Dry season

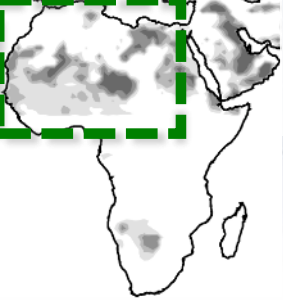


fluvial deposit, sediment

Wet Season

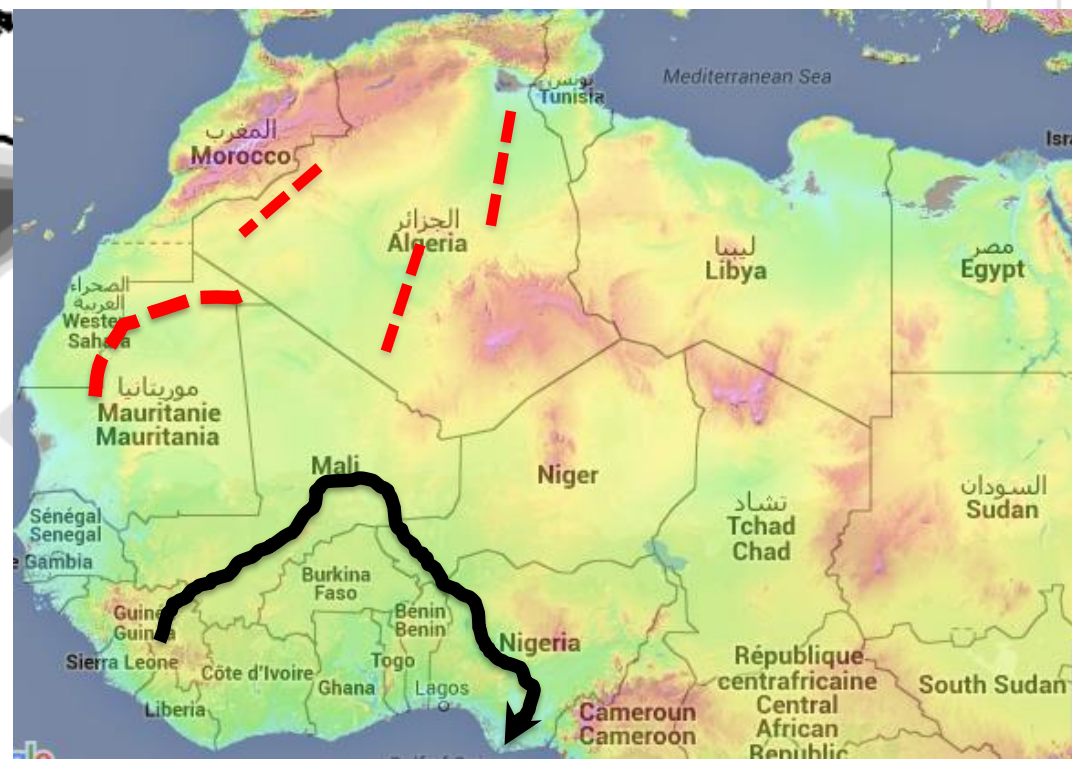
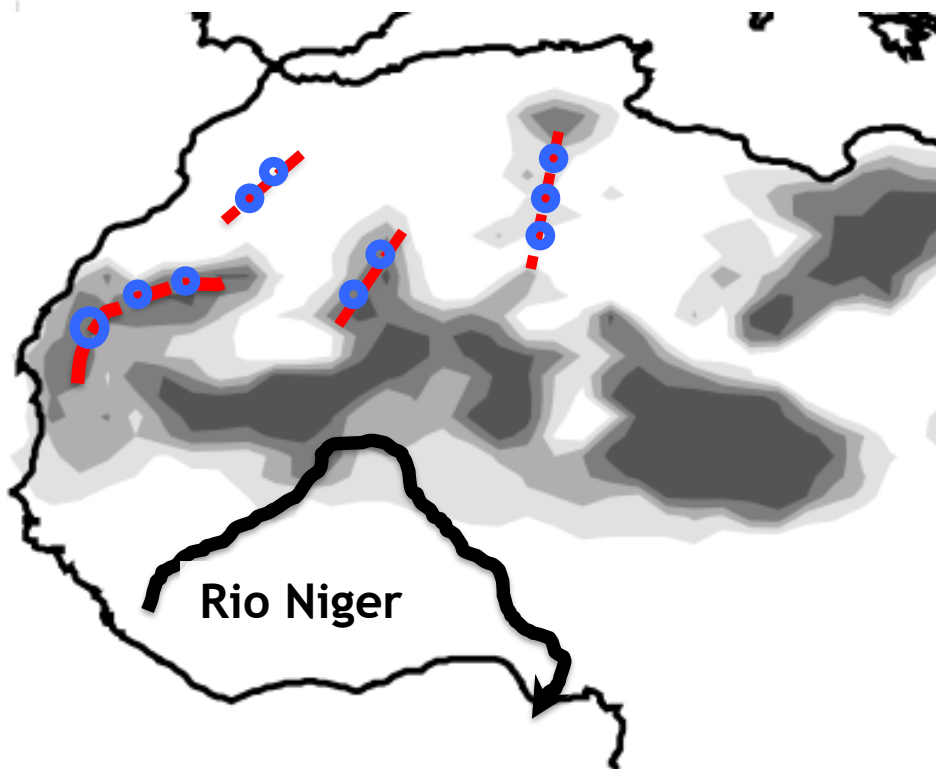


↑
Detección satélite



Sahara
Sahel

Sahara

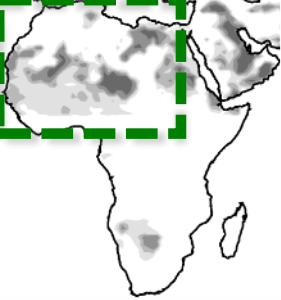


Detección satélite

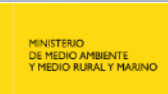
of Days AI > 0.7 or 1



- bajas topográficas
Wakis: barrancos con inundaciones estacionales
- chots, sabkas: lechos salados de lagos ecos



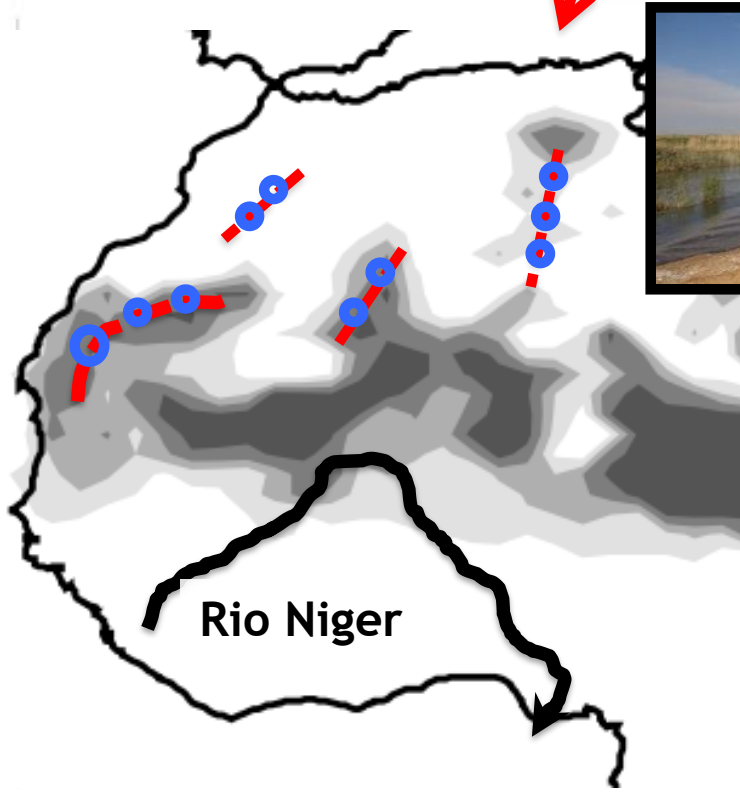
Sahara
Sahel



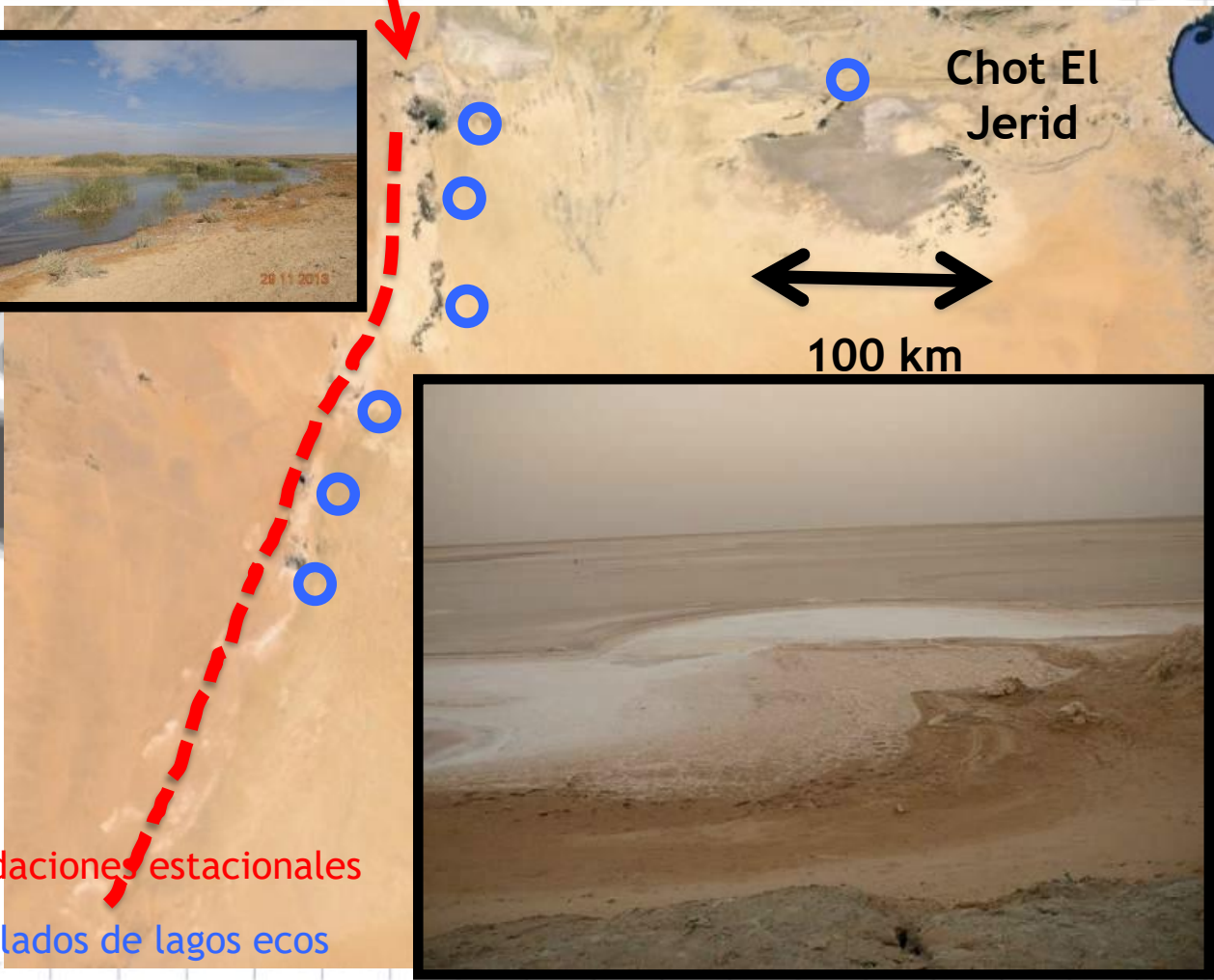
Cuenca Ouargla

Sahara

chots



- bajas topográficas
- Wakis: barrancos con inundaciones estacionales
- chotts, sabkhas: lechos salados de lagos ecos



Chot El Jerid

100 km

chotts, sabkhas, wadis, salares

1. what is dust ?

There are several types of sources, but the mayor dust sources are associate with dry lakes/river beds

sediments, fluvial & alluvial deposits

2. chemistry and mineralogy

clays, feldspars, oxides, evaporites

3. Size and morphology

1 and 20 μm
agglomerates

Chotts, Sabkhas



Dry lakes beds



wadis



**strong link between water and dust
natural sources**

types of dust sources:

desert dust

paraglacial dust

paraglacial dust

paraglacial regions:

- > 50°N
- > 40°S

Paraglacial means unstable conditions caused by a significant relaxation time in processes and geomorphic patterns following glacial climates.

When a large mass of ice melts:

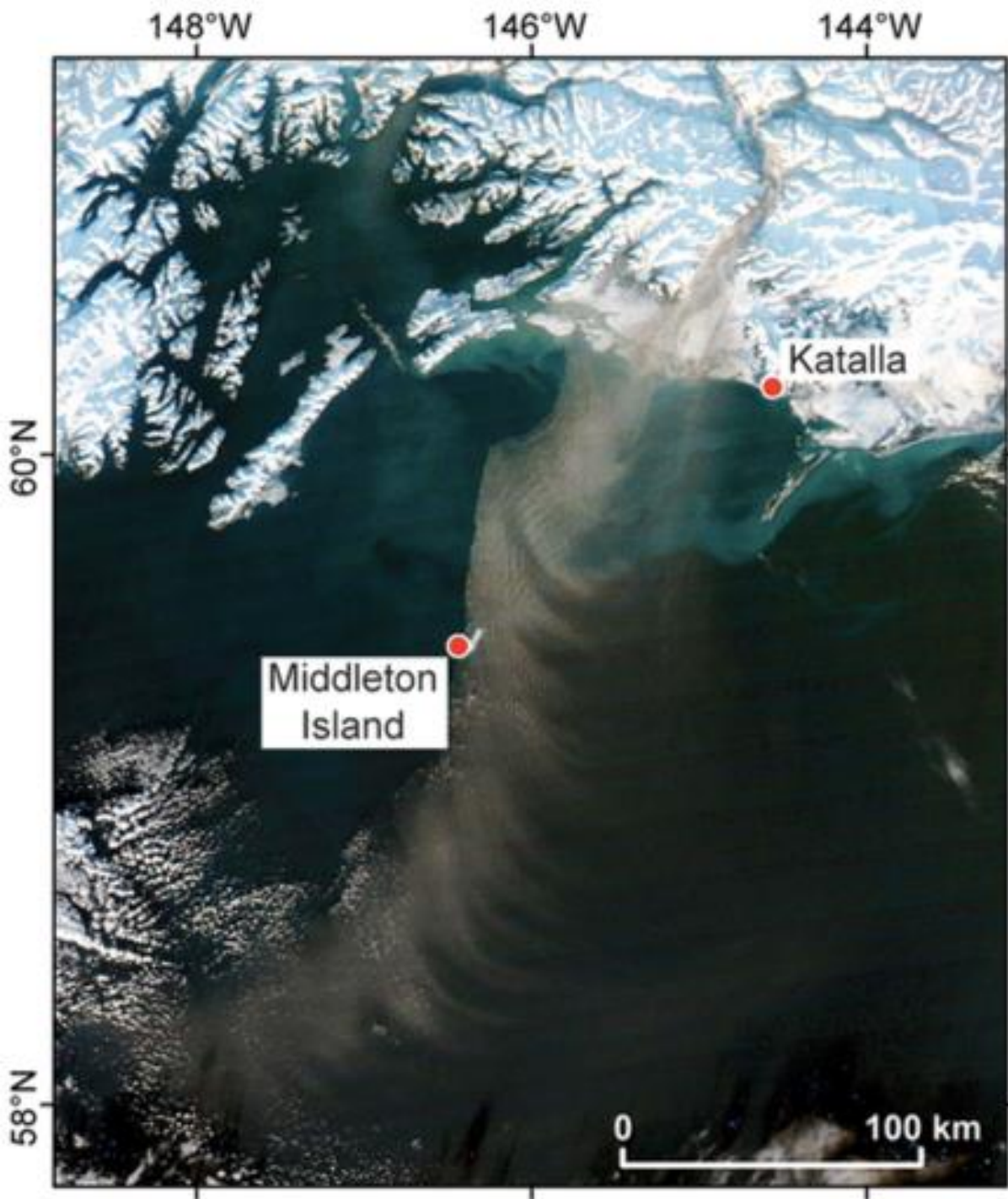
- newly exposed landscape free of vegetation
- water stream discharge, increasing erosion
- sediment deposition

➔ **dust source**





Hubbard Glacier, Alaska



MODIS Aqua
Gulf of Alaska
4-Dec-2015

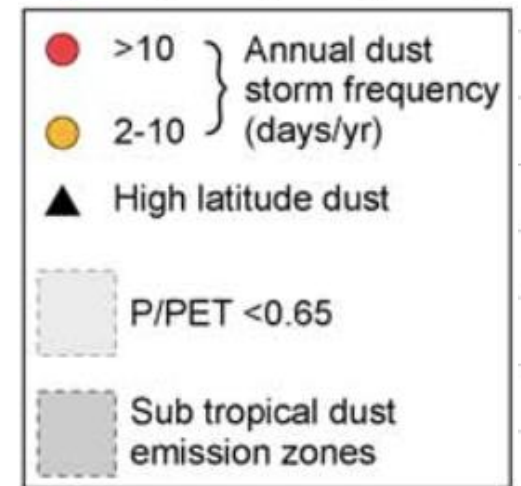
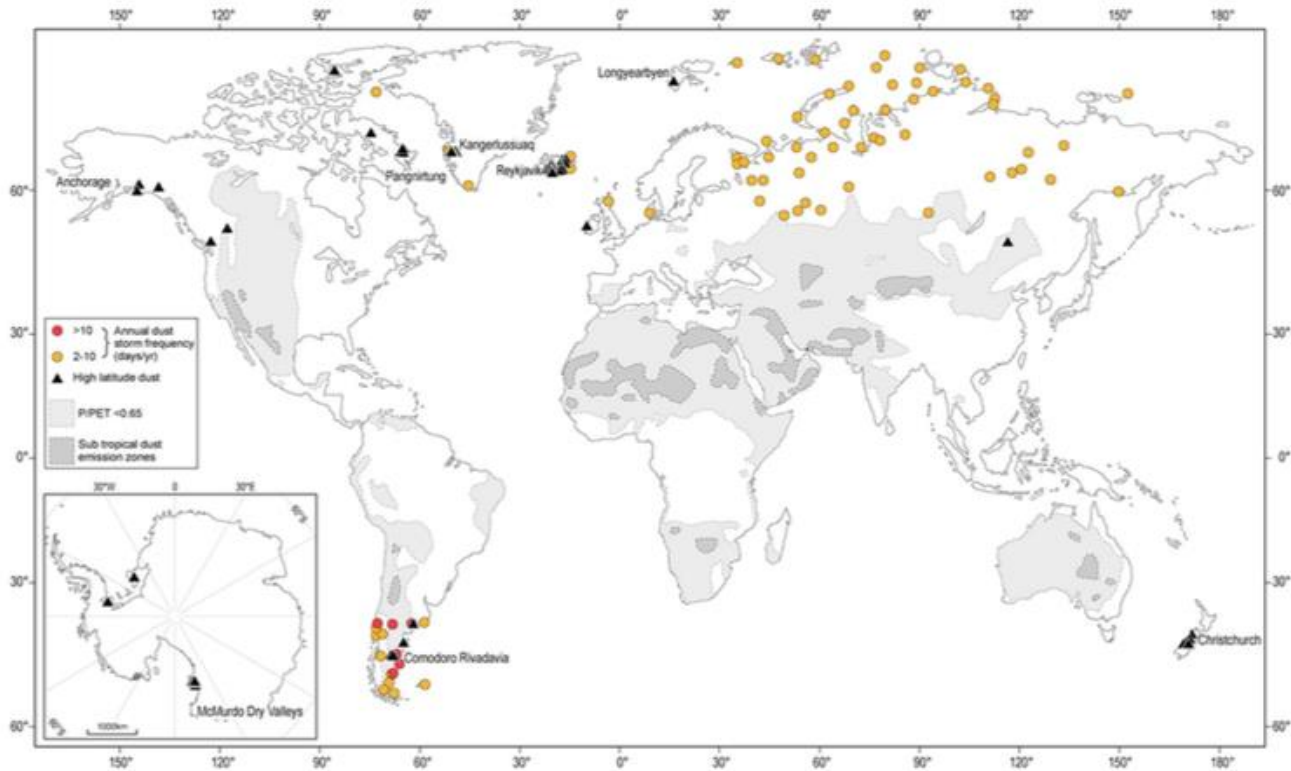
paraglacial dust

paraglacial regions:

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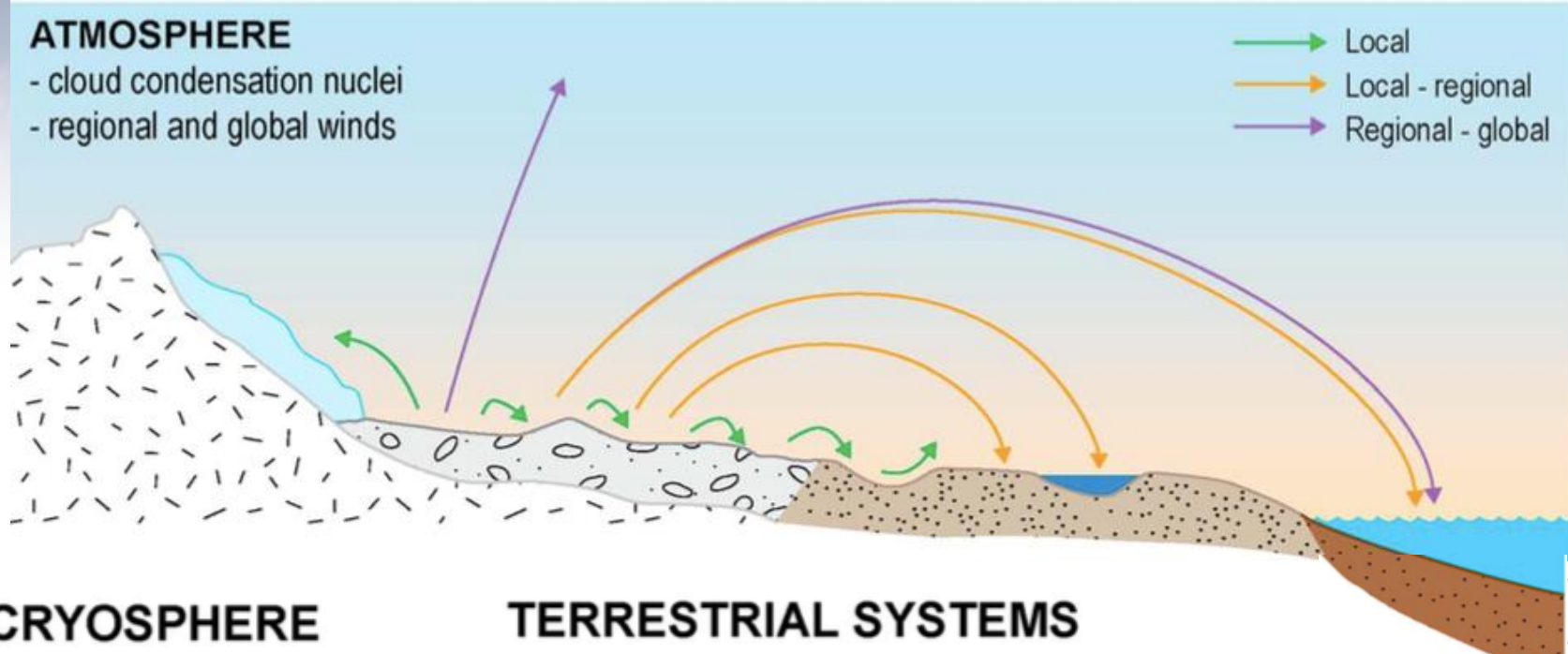
5% of global dust budget

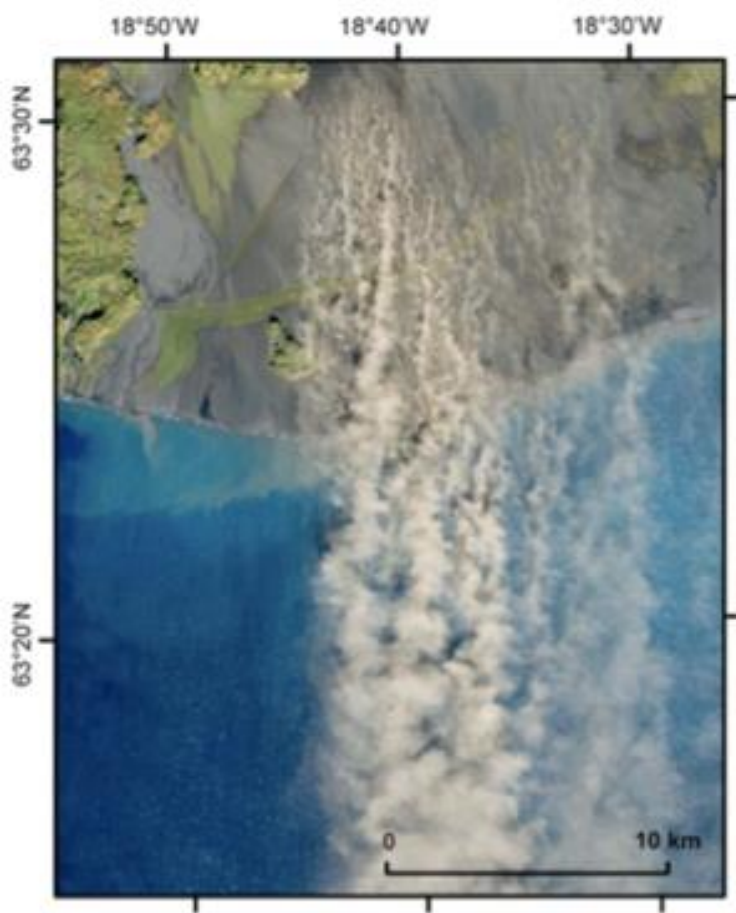


Reviews of Geophysics

High-latitude dust in the Earth system

Joanna E. Bullard¹, Matthew Baddock¹, Tom Bradwell², John Crusius³, Eleanor Darlington¹, Diego Gaiero⁴, Santiago Gassó⁵, Gudrun Gisladottir⁶, Richard Hodgkins¹, Robert McCulloch², Cheryl McKenna-Neuman⁷, Tom Mockford¹, Helena Stewart², and Throstur Thorsteinsson⁸



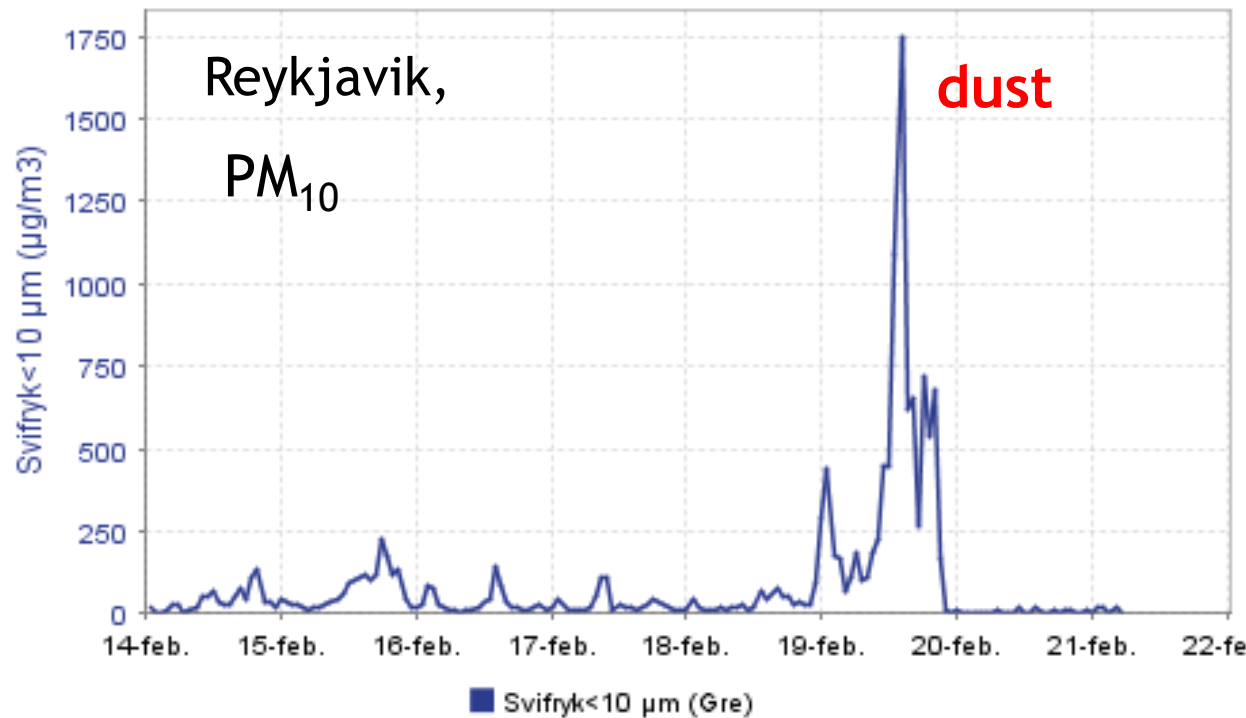


Landsat, 17 Sep 2013,
Mýrdalssandur -
Iceland

Bullard et al., 2016

Dangerous air pollution hit Iceland's capital

Posted by [Chillymanjaro](#) on February 21, 2014 in categories [Follow @TheWatchers_](#)
[Dust and haze](#), [Pollution](#)



types of dust sources:

desert dust

they exists by natural causes

paraglaciatic dust

by man influence:

new climate-change-related

glacier, climate change

Glacier change and glacial lake outburst flood risk in the Bolivian Andes

Simon J. Cook^{1,2}, Ioannis Kougkoulos^{1,2}, Laura A. Edwards^{2,3}, Jason Dortch^{2,3}, and Dirk Hoffmann⁴

The Cryosphere, 10, 2399–2413, 2016

Bolivia:
surface covered by glacier
decreased 43% (1986-2014)

Proglacial lakes
future dust sources

types of dust sources:

desert dust
glacier dust

they exists by natural causes

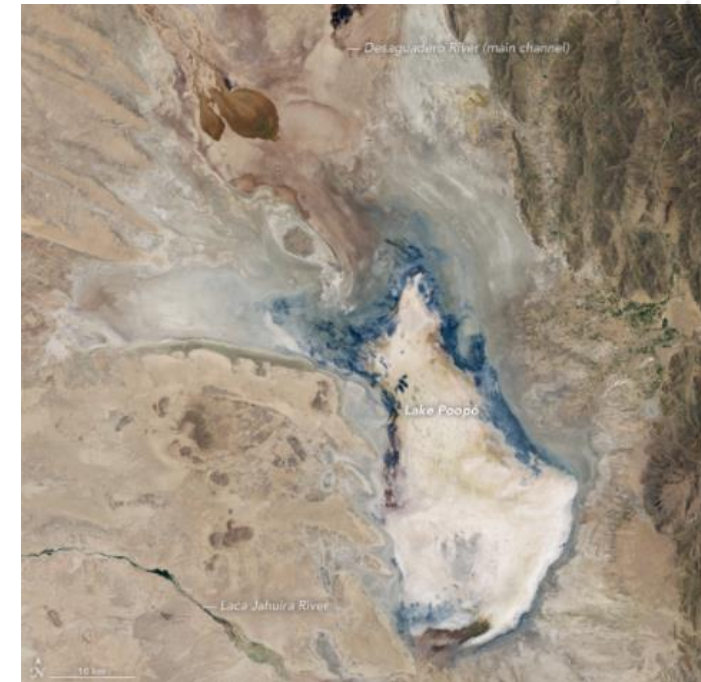
by man influence:

new climate-change-related
new lakes desiccation

Bolivia's Lake Poopó Disappears

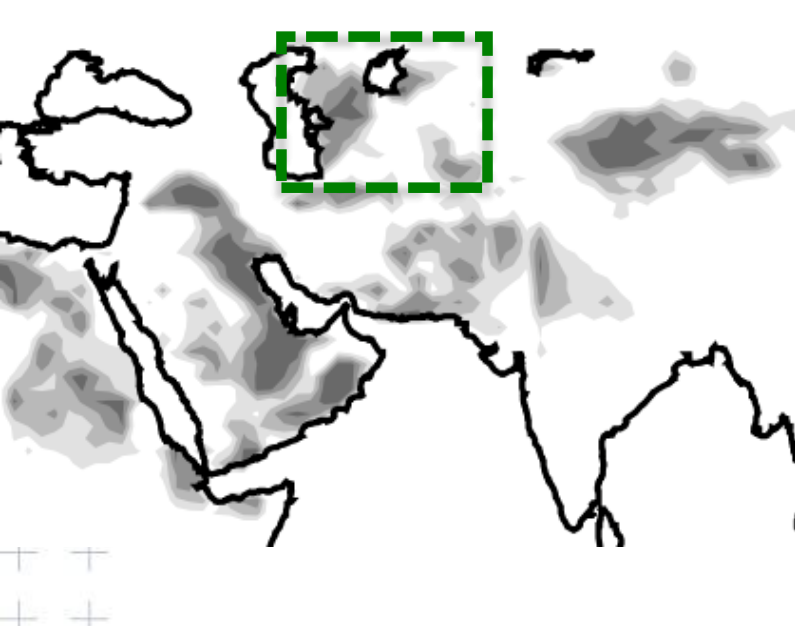


April 2013



Jan 2016

2015-2016 drought ENSO enhanced



During the 1960s, the Syr y Amu rivers were re-chanelled for crop cultivation and the Aral Sea diminished increasing dust soruces

Caspian Kazakhstan Aral Sea
Sea



Uzbekistan

Turkmenistan

1989



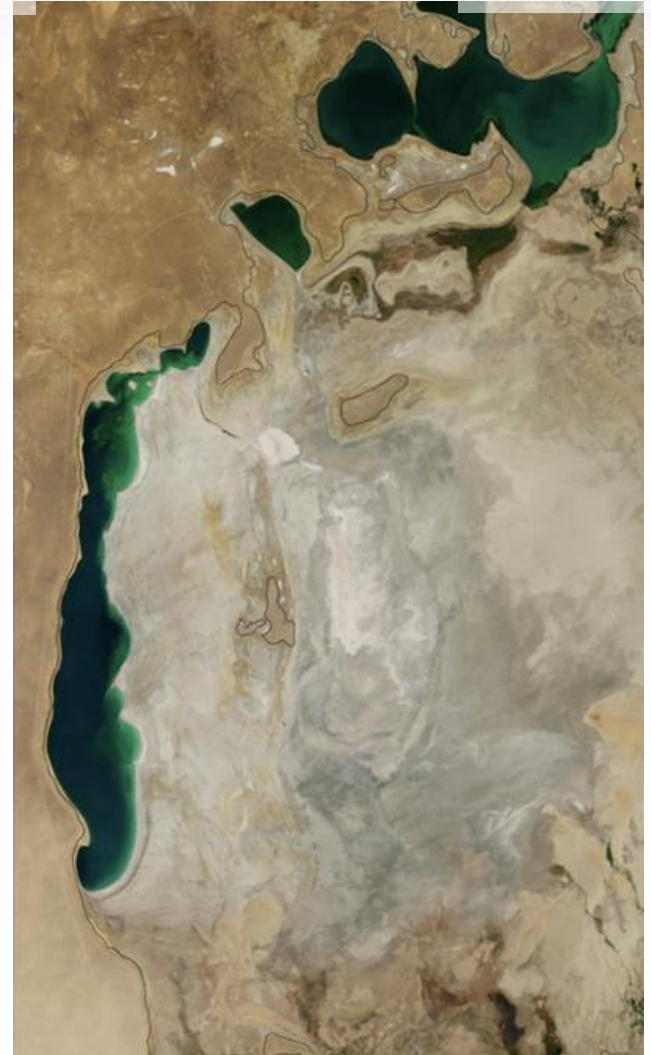
July - September, 1989

2003



August 12, 2003

2014



Aral Sea



Urmia lake

an emerging important
dust source



1972



1984



1987



1989



1998



2000



2002



2006



2009



2011



2012



2014

<https://www.rt.com/viral/353940-urmia-lake-drought-red/>

types of dust sources:

desert dust
glacier dust

they exists by natural causes

by man influence:

new climate-change-related
new lakes desiccation

agriculture dust

Major Dust Activity Frequency

Aerosol Index > 1

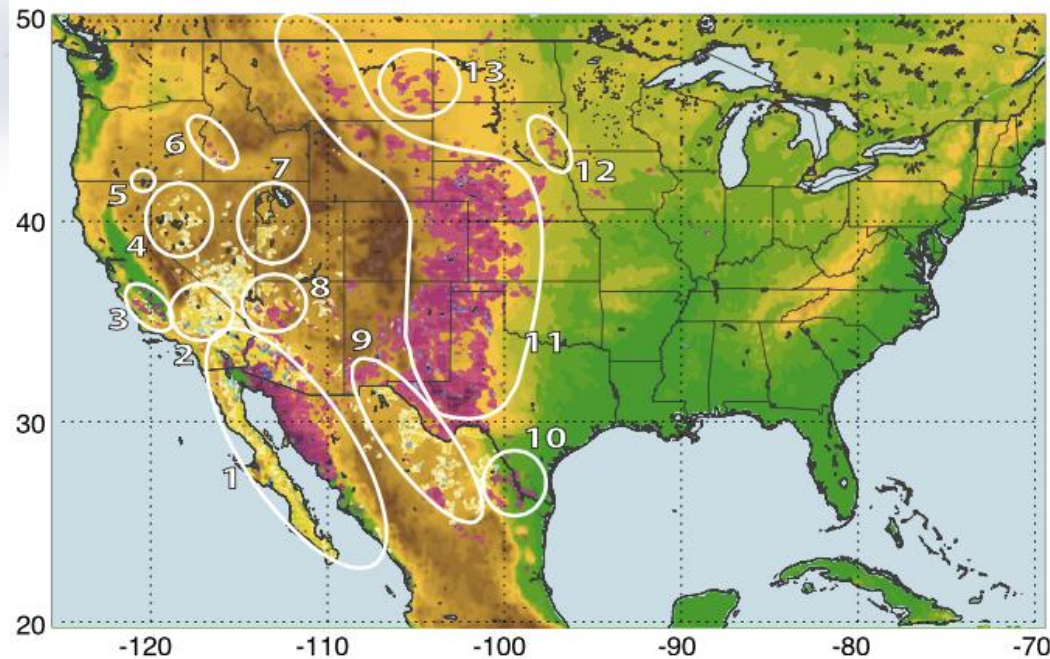
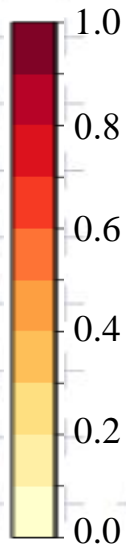
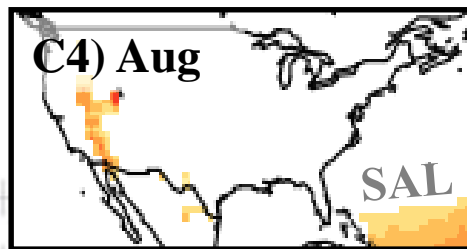
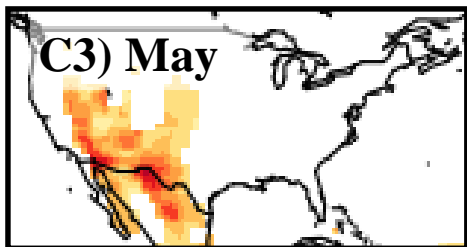
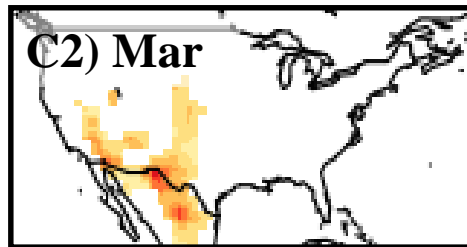
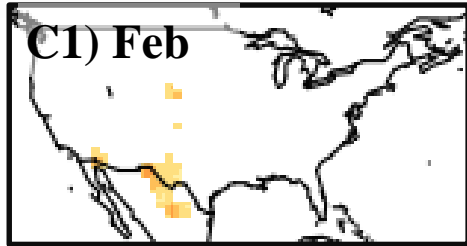
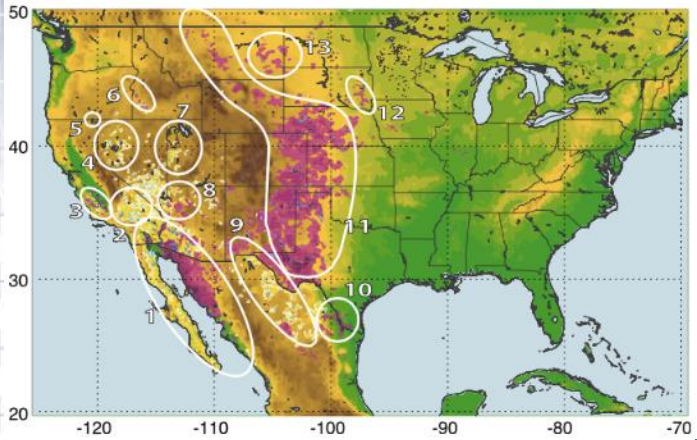


Figure 11. Distribution of the percentage number of days per season (March, April, and May) M-DB2 DOD > 0.2 over North America with color code as in Figure 6. The white circled sources are numbered as follows: 1, Sonoran Desert; 2, Mojave Desert; 3, San Joaquin Valley; 3, Black Rock-Smoke Creek deserts; 4, Goose Lake; 6, Snake River; 7, Great Salt Lake Desert; 8, Colorado River; 9, Chihuahuan Desert; 10, Rio Grande; 11, High Plains; 12, Big Sioux River; and 13, lower Yellowstone Valley.

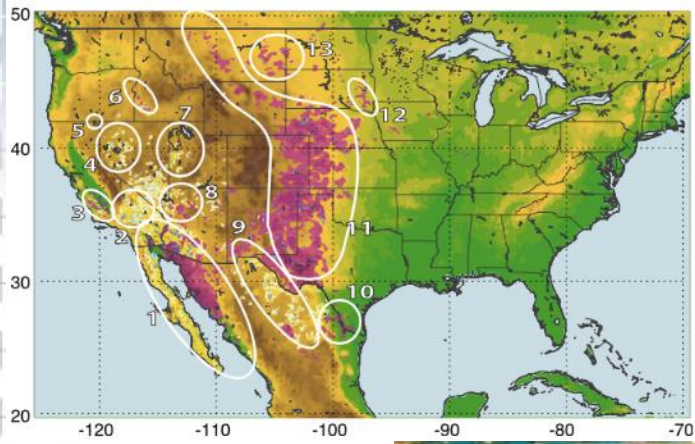
Joseph M. Prospero,¹ Paul Ginoux,² Omar Torres,³ Sharon E. Nicholson,⁴ and Thomas E. Gill⁵

ENVIRONMENTAL CHARACTERIZATION OF GLOBAL SOURCES OF ATMOSPHERIC SOIL DUST IDENTIFIED WITH THE NIMBUS 7 TOTAL OZONE MAPPING SPECTROMETER (TOMS) ABSORBING AEROSOL PRODUCT

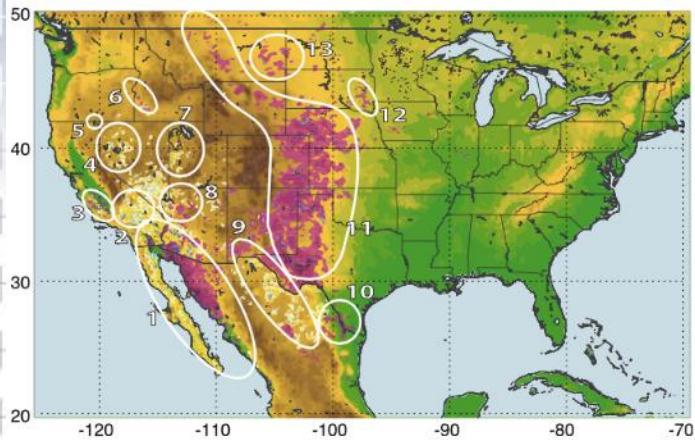
Great Plains



Great Plains



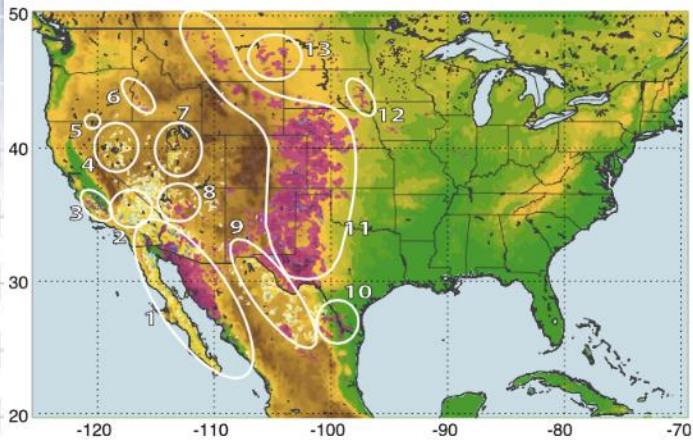
Great Plains



oklahoma



Great Plains



Managing *wind erosion* on the Plains

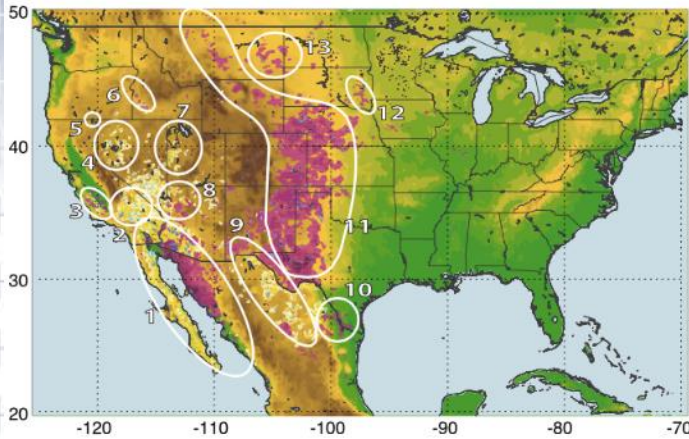
Clay Robinson

Crops & Soils Magazine - Article

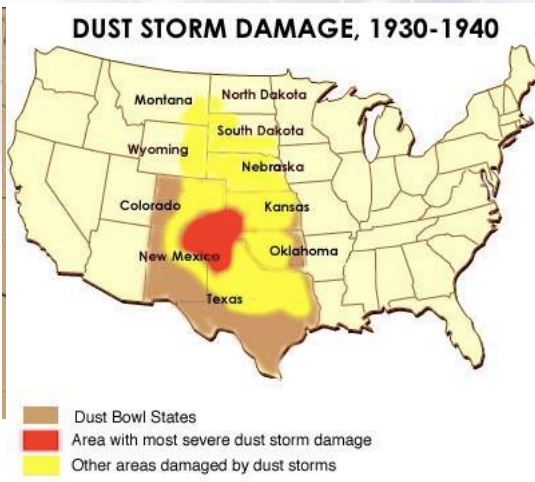
<https://dl.sciencesocieties.org/publications/cns/articles/48/1/12>

All that was left after the dust settled

Great Plains



Dust Bowl: 1930s



Dust Bowl: 1930s

affected 400,000 km² along Texas and Oklahoma and adjacent regions of New Mexico, Colorado and Kansas.

dust - "black blizzards" or "black rollers" - traveled cross country, reaching the East Coast, including New York City and Washington, D.C



https://en.wikipedia.org/wiki/Dust_Bowl#cite_note-5

<http://geol105naturalhazards.voices.wooster.edu/eating-sleeping-breathing-dust-health-hazards-of-the-dust-bowl/>

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by man influence:

new climate-change-related
new lakes desiccation
agriculture dust

*Regional to
synoptic scale*

industrial dust

*Local to regional
scale*

mines



fertilizers plants *phosphate rocks*



cement factories



ceramic manufactures



types of dust sources:

desert dust
glacier dust

new climate-change-related
new lakes desiccation
agriculture dust

industrial dust
construction dust

construction & demolition dust



guidelines for preventing dust emissions



1. Introduction

- 1.1. How to use this guidance

2. Air Quality Impact Evaluation

- 2.1. Site evaluation
- 2.2. Site impact
- 2.3. Site evaluation guidelines
- 2.4. Mitigation measures for low risk sites
- 2.5. Mitigation measures for medium risk sites
- 2.6. Mitigation measures for high risk sites

3. Method Statement

- 3.1. For all sites
- 3.2. Site waste management plans
- 3.3. Additional information for high risk sites
- 3.4. Specific site issues (asbestos contaminated land)

4. Dust and Emission Control Measures

- 4.1. Pre site preparation
- 4.2. Haulage routes
- 4.3. Site entrances and exits
- 4.4. Mobile crushing plant
- 4.5. Concrete batching
- 4.6. Excavation and earthworks
- 4.7. Stockpiles and storage mounds
- 4.8. Cutting, grinding and sawing
- 4.9. Chutes and skips
- 4.10. Scabbling
- 4.11. Waste disposal
- 4.12. Dealing with spillages
- 4.13. Demolition activities
- 4.14. Hazardous and contaminated materials
- 4.15. Specific site activities

5. Site Monitoring

- 5.1. Site monitoring protocols
- 5.2. Site action levels

Introduction

What are the benefits of effective dust control?

How does the community view dust from construction sites?

How does the industry view dust from construction sites?

Why is dust a problem?

Constraints on dust control

Dust control measures

PRE-CONSTRUCTION MEASURES

SITE MEASURES

STORAGE PILES/GENERAL MATERIAL STORAGE

HAULED MATERIALS

PAVED ROAD TRACKOUT

types of dust sources:

desert dust
glacier dust

new climate-change-related
new lakes desiccation
agriculture dust

industrial dust
construction dust
road dust

material accumulated on road and suspended vehicles:

- construction/demolition dust
- industrial dust
- settled desert dust
- settled air pollutants
- pavement

-**brakes:** barite (BaSO_4), hematite (Fe_2O_3), tenorite (CuO), zircon (ZrSiO_4), calcite (CaCO_3), periclase (MgO), vermiculite, and sulphide species such as stibnite (Sb_2S_3), pyrite (FeS_2), chalcopyrite (CuFeS_2), covellite (CuS), sphalerite (ZnS), hauerite (MnS_2), and molybdenite (MoS_2).

-**tyres:** rubber and metals (steel, Zn,...)



road dust

http://www.ehu.eus/sem/macla_pdf/macla16/Macla16_154.pdf

types of dust sources:

desert dust
glacier dust

they exists by natural causes

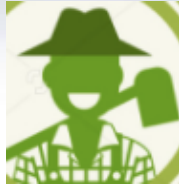
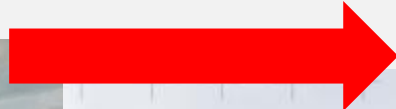
by man influence:

new climate-change-related
new lakes desiccation
agriculture dust

*Regional to
synoptic scale*

industrial dust
construction dust
road dust

*Local to regional
scale*



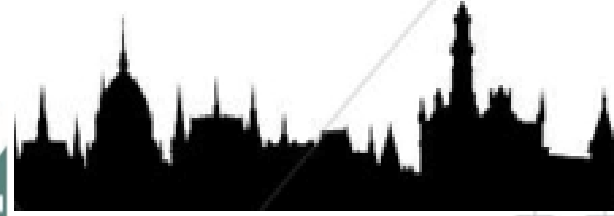
agriculture



construction



industry



road dust

desert dust



people live in cities and breath a cocktail
different of dust
+ pollutants

dust = desert + agriculture + construction + industrial + road-dust + ...

dust, aerosols and pollutants

in-situ observations

PM_{10} and $PM_{2.5}$ levels

PM_{10} and $PM_{2.5}$ composition

complementary observations

remote sensing observations

let's build our observation network !!!



aerosols, a cocktail of chemicals:

- dust
- sulphate
- nitrate
- organic matter
- black carbon (soot)
- metals (Ni, As, Cd, V, Co...)
- sea salt

size: 1 nm (10^{-9} m) to 20 μ m (10^{-6} m)

human hair: 70 μ m

people live in cities and breath a cocktail dust + pollutants

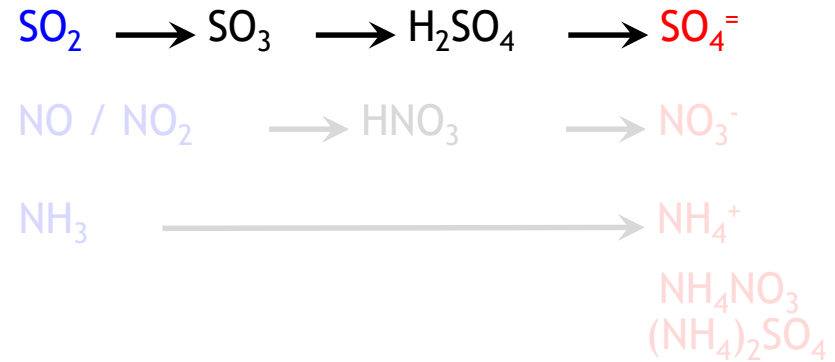


aerosols, a cocktail of chemicals:

- dust
- sulphate
- nitrate
- organic mater
- black carbon (soot)
- metals (Ni, As, Cd, V, Co...)
- sea salt

gas precursor

aerosol

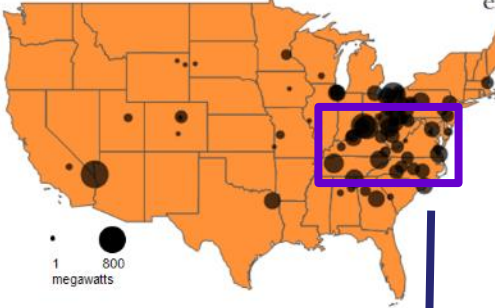


SO₂: oil refineries, coal power plants, ships, industry

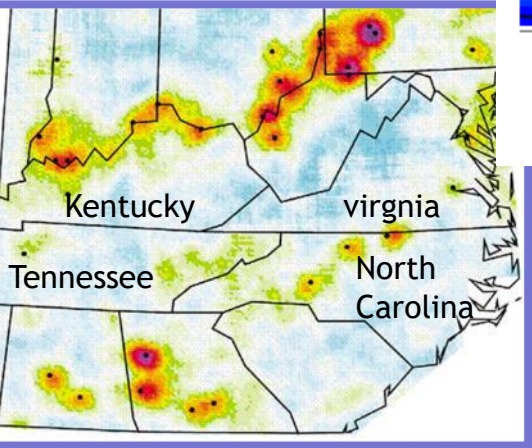
sulfato

122 Tg/y

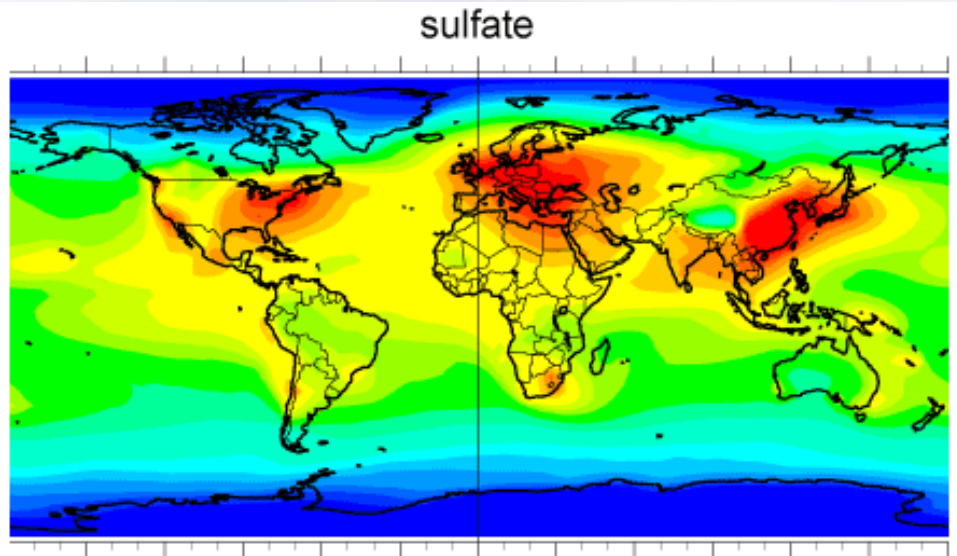
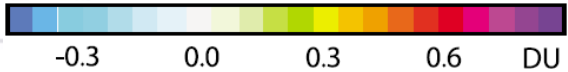
coal power plants



promedio 2005-2007



OMI SO₂ DU



coal power plants

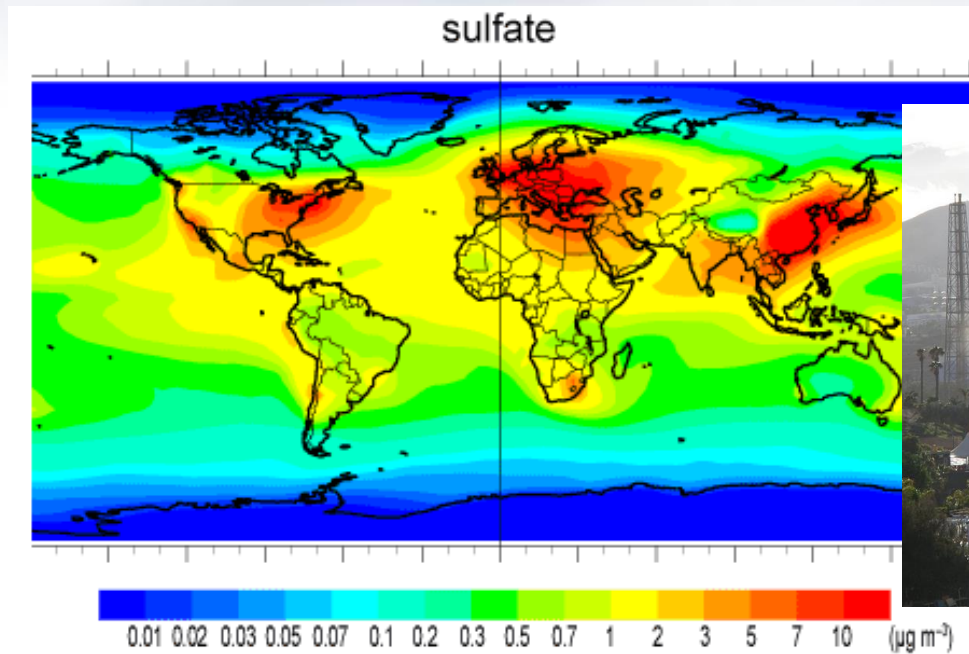


sulfato

Oil refinery



Oil refinery



ships

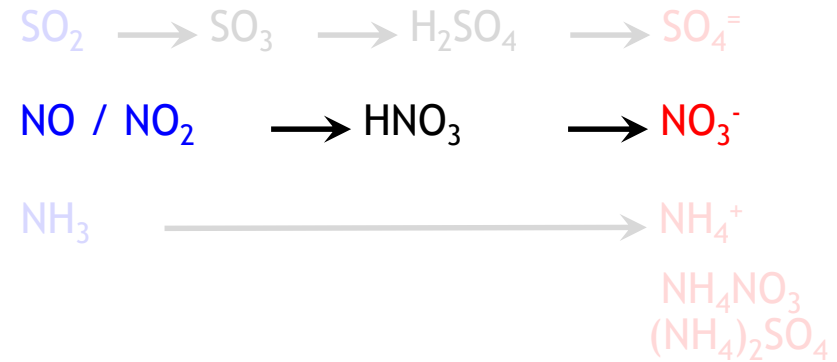


aerosols, a cocktail of chemicals:

- dust
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gas precursor

aerosol

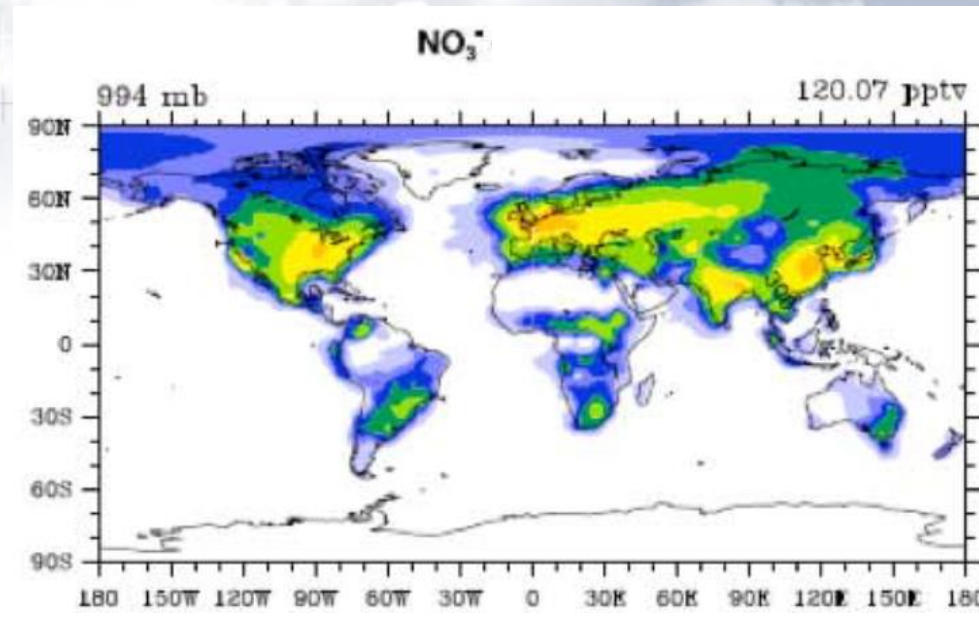


NO_x : vehicle exhaust, power plants, industry

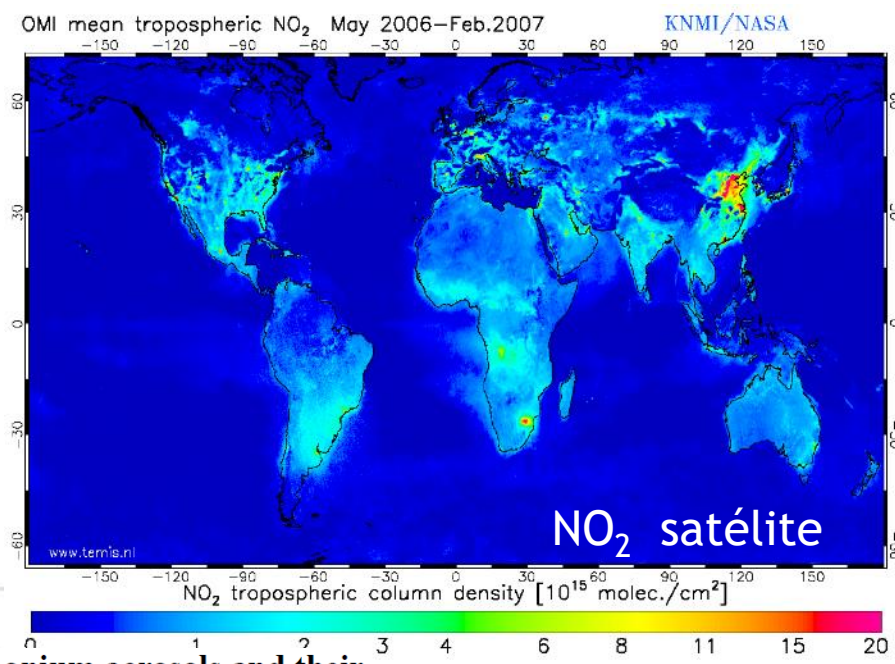


nitrate

18 Tg/y



NH₄NO₃

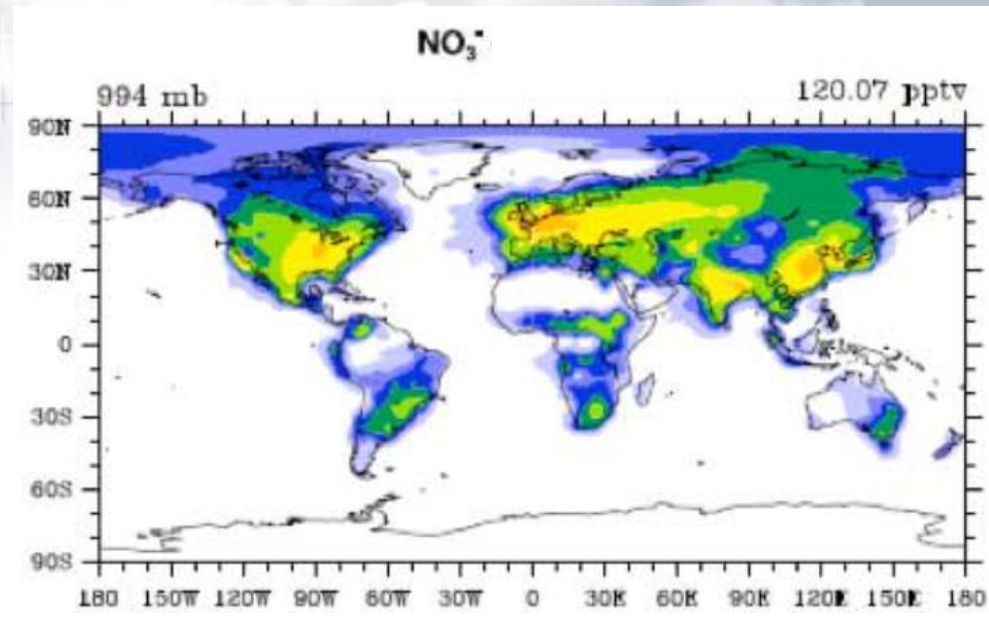


Atmos. Chem. Phys., 12, 9479–9504, 2012
www.atmos-chem-phys.net/12/9479/2012/

L. Xu and J. E. Penner

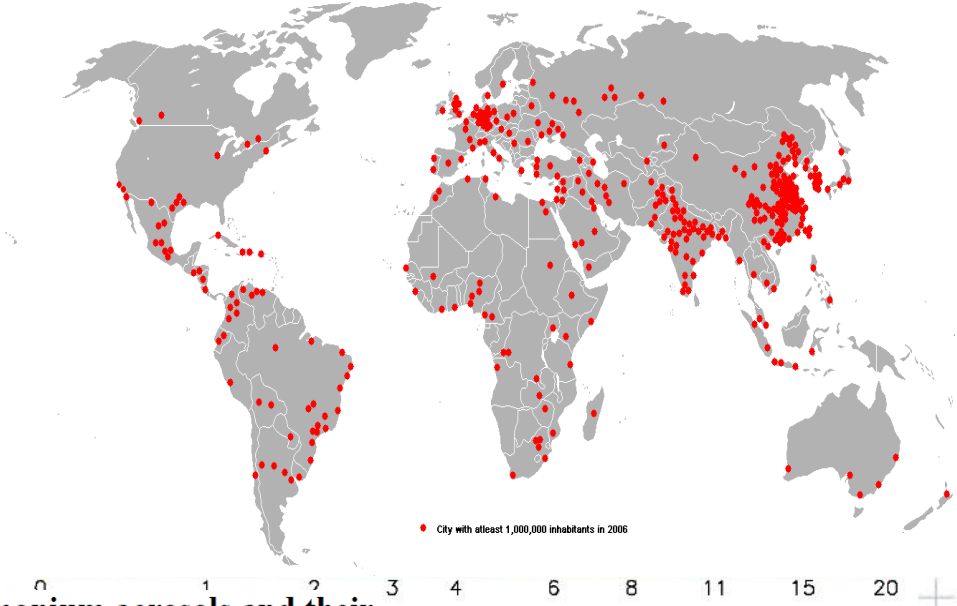
Global simulations of nitrate and ammonium aerosols and their radiative effects

nitrate



NH₄NO₃

OMI mean tropospheric NO₂ May 2006–Feb.2007 KNMI/NASA



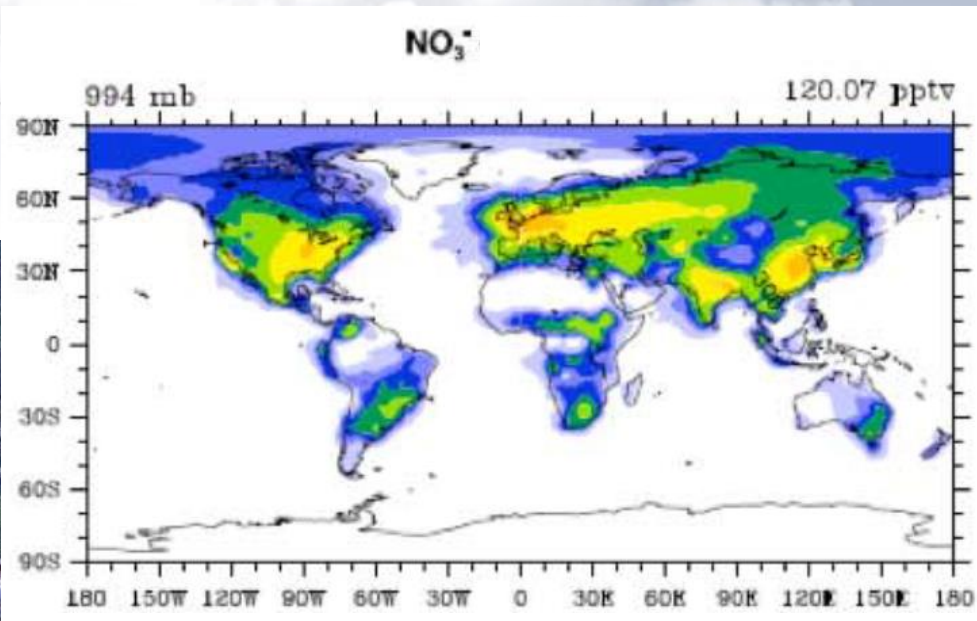
cities > 1 million inhabitants

Atmos. Chem. Phys., 12, 9479–9504, 2012
www.atmos-chem-phys.net/12/9479/2012/

L. Xu and J. E. Penner

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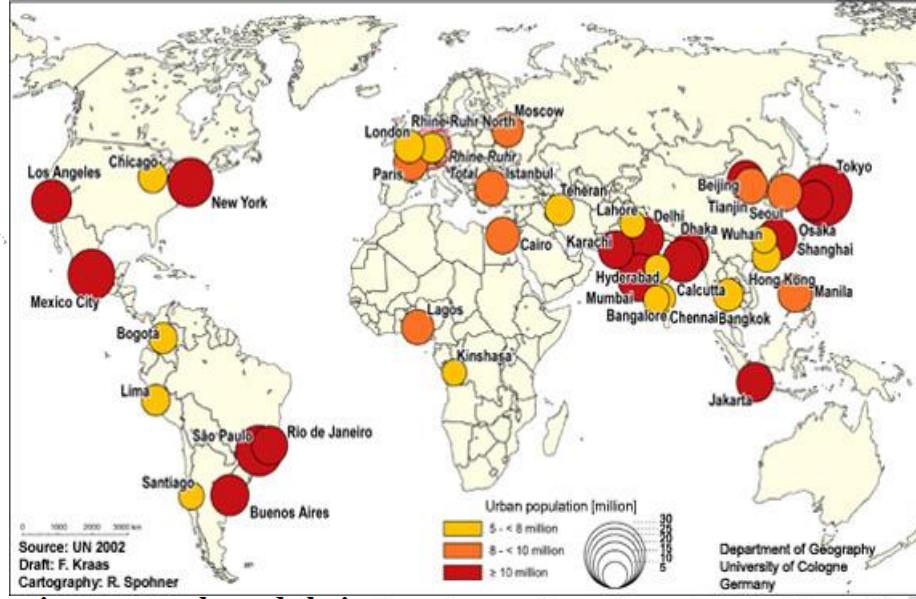


NH₄NO₃



OMI mean tropospheric NO₂ May 2006–Feb.2007 KNMI/NASA

–150 –120 –90 –60 –30 0 30 60 90 120 150



Mega-cities, > 5 Millones habitantes

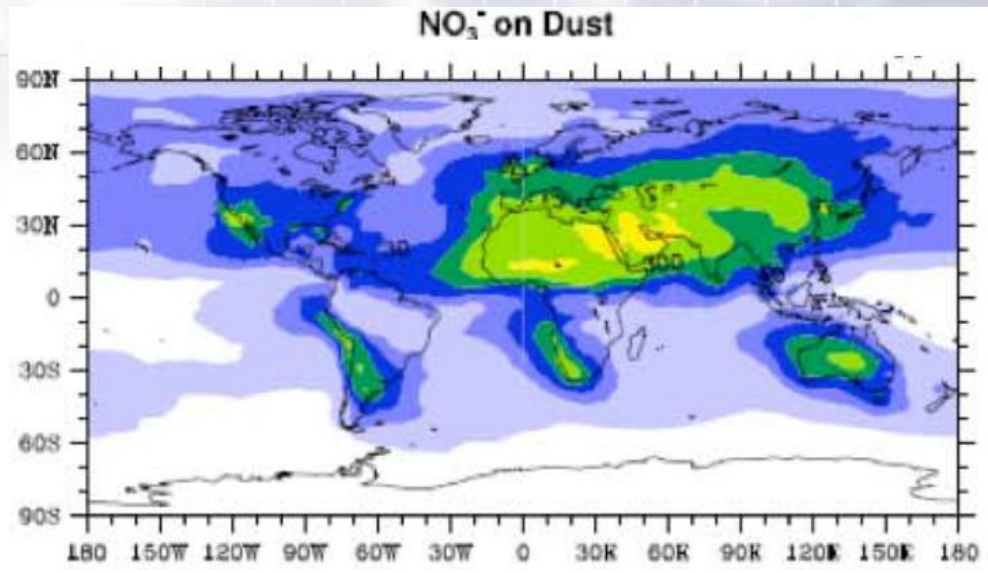


Atmos. Chem. Phys., 12, 9479–9504, 2012
www.atmos-chem-phys.net/12/9479/2012/

L. Xu and J. E. Penner

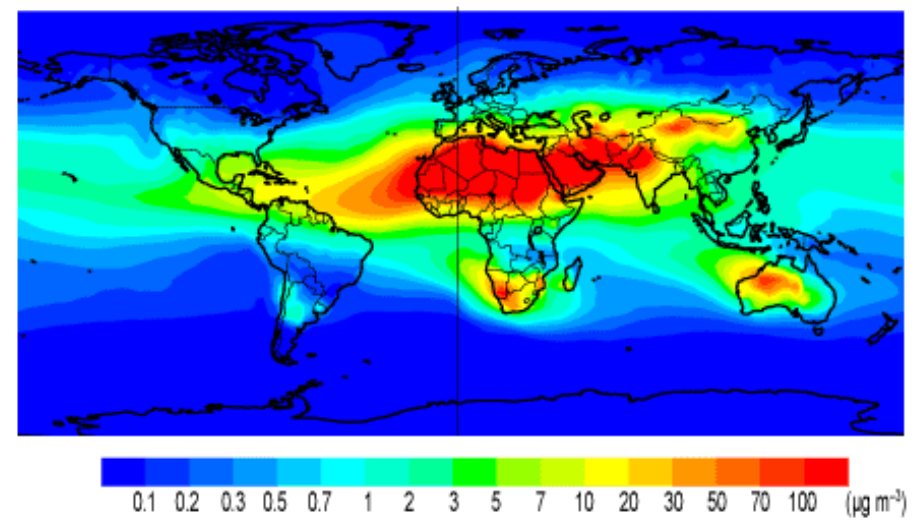
Global simulations of nitrate and ammonium aerosols and their radiative effects

nitrate



NO_3^- - dust
 $\text{Ca}(\text{NO}_3)_2$

soil dust



Atmos. Chem. Phys., 12, 9479–9504, 2012
www.atmos-chem-phys.net/12/9479/2012/

L. Xu and J. E. Penner

Global simulations of nitrate and ammonium
radiative effects

<http://www.knmi.nl/omi/research/product/index.php>

aerosols, a cocktail of chemicals:

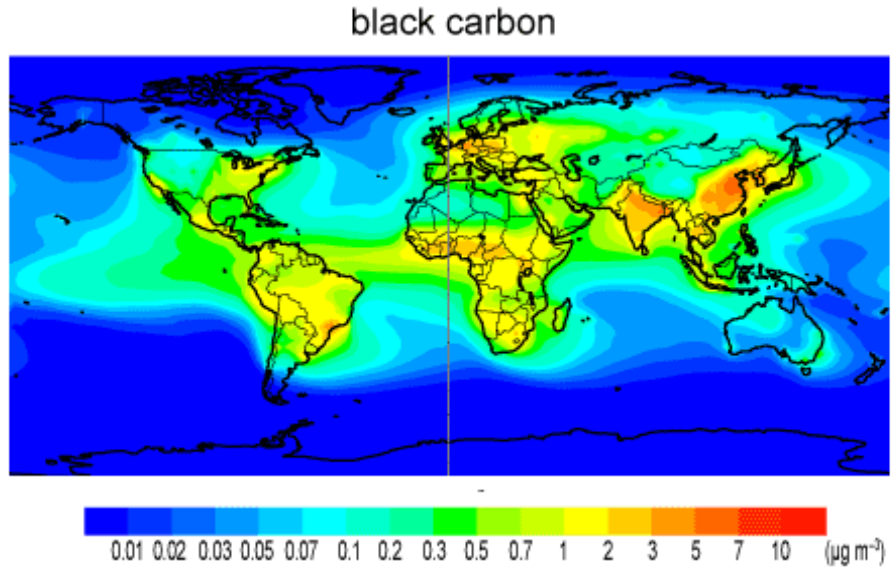
dust
sulphate
nitrate
organic mater
black carbon (soot)
metals (Ni, As, Cd, V, Co...)
sea salt

Black carbon: vehicle exhaust (diesel) , combustion sources

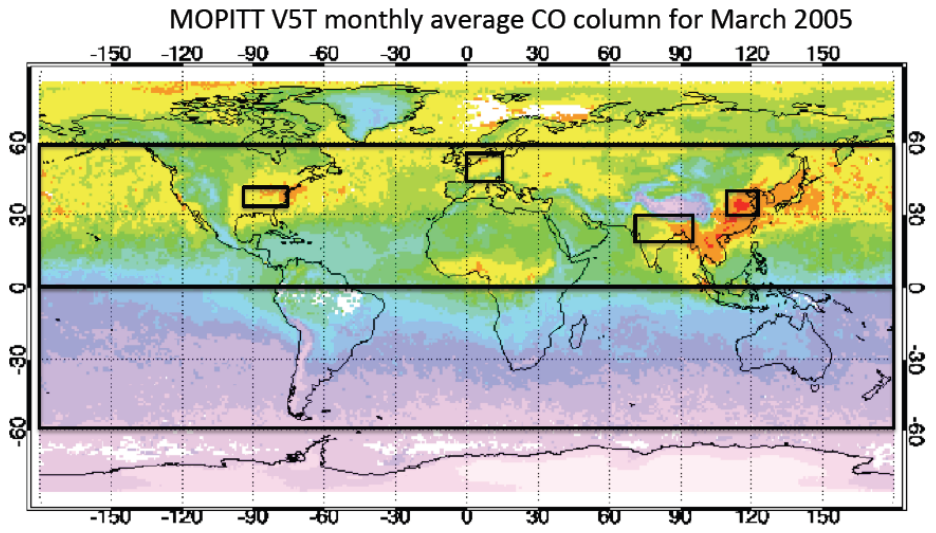
black carbon

10.5 Tg/y

diesel, 4x4, camiones



automóviles



India

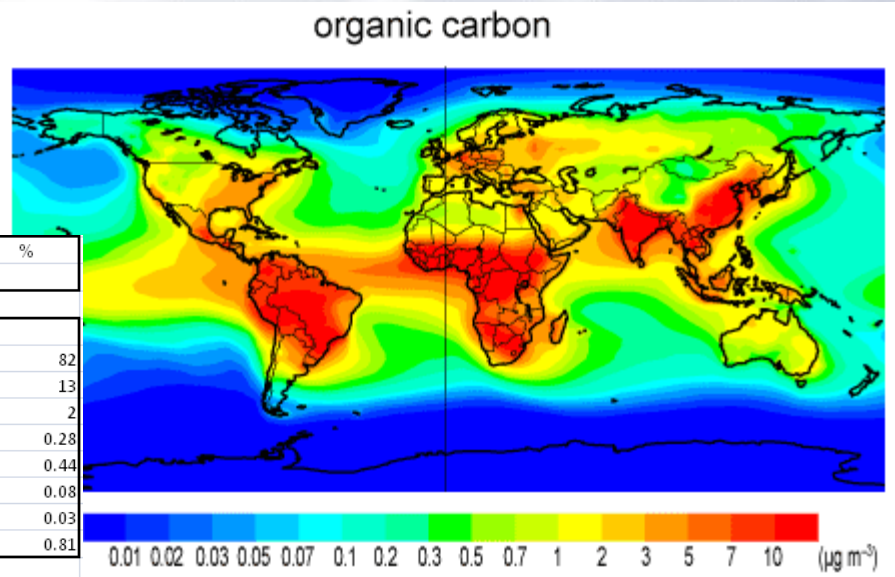


aerosols, a cocktail of chemicals:

dust
sulphate
nitrate
organic matter
black carbon (soot)
metals (Ni, As, Cd, V, Co...)
sea salt

organic matter: combustion sources, vehicle exhaust

organic carbon

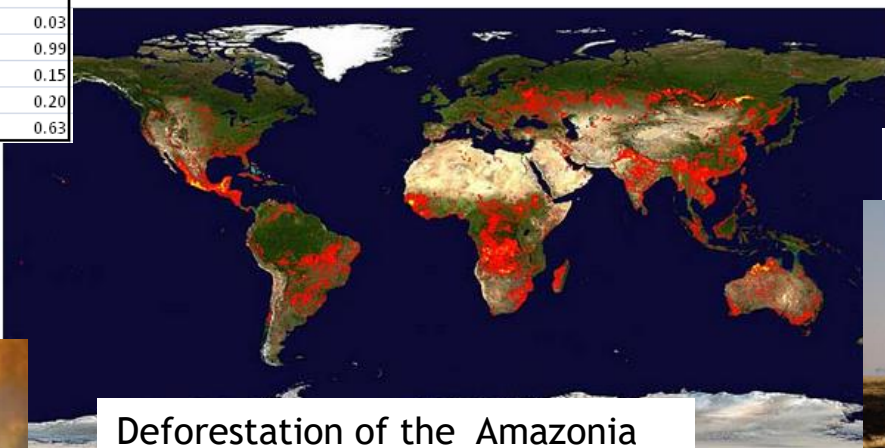


		Tg/y	%
	TOTAL:	12380	
PRIMARY			
Prim Nat	sea salt:	10130	82
Prim Nat	desert dust:	1600	13
Prim Nat	fine volcanic ashes:	200	2
Prim Nat	biogenic:	35	0.28
Prim Nat + Ant	POA (biomass burning, biofuels):	54	0.44
Prim Ant	black carbon:	10.5	0.08
Prim Ant	POA (combustion fossil fuels):	4	0.03
Prim Ant	Industrial dust:	100	0.81
SECONDARY			
Sec Ant	SOA (industrial + fossil fuels):	3.5	0.03
Sec Ant	Sulphate (fossil fuels):	122	0.99
Sec Ant+Nat	Nitrate:	18	0.15
Sec Nat	SOA (biogenic):	25	0.20
Sec Nat	Sulphate (volcanic and biogenic):	78	0.63

Vehicle exhaust



←Satelite detecion of fires



Sabana Africana



Paraguay: burn forest for cultivation of soja and sugar cane

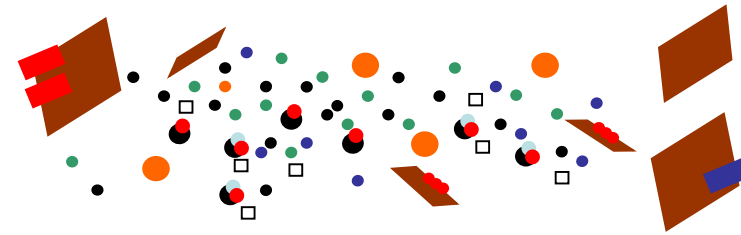


Deforestation of the Amazonia



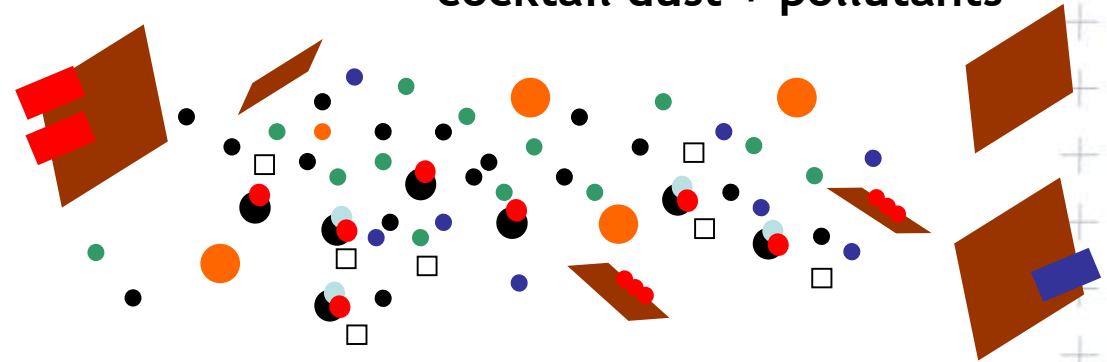


people live in cities and breath a cocktail dust + pollutants





people live in cities and breath a cocktail dust + pollutants



In air quality, aerosols:

PM₁₀: mass concentration ($\mu\text{g}/\text{m}^3$) of all aerosols smaller than $10\ \mu\text{m}$
inhalable particles

PM_{2.5}: mass concentration ($\mu\text{g}/\text{m}^3$) of all aerosols smaller than $2.5\ \mu\text{m}$
alveolar particles



people live in cities and breath a cocktail dust + pollutants



PM_{10} : Σ dust + sea salt + (sulphate + nitrate + organic matter + black carbon + metals)

$PM_{2.5}$: Σ dust + sea salt + (sulphate + nitrate + organic matter + black carbon + metals)

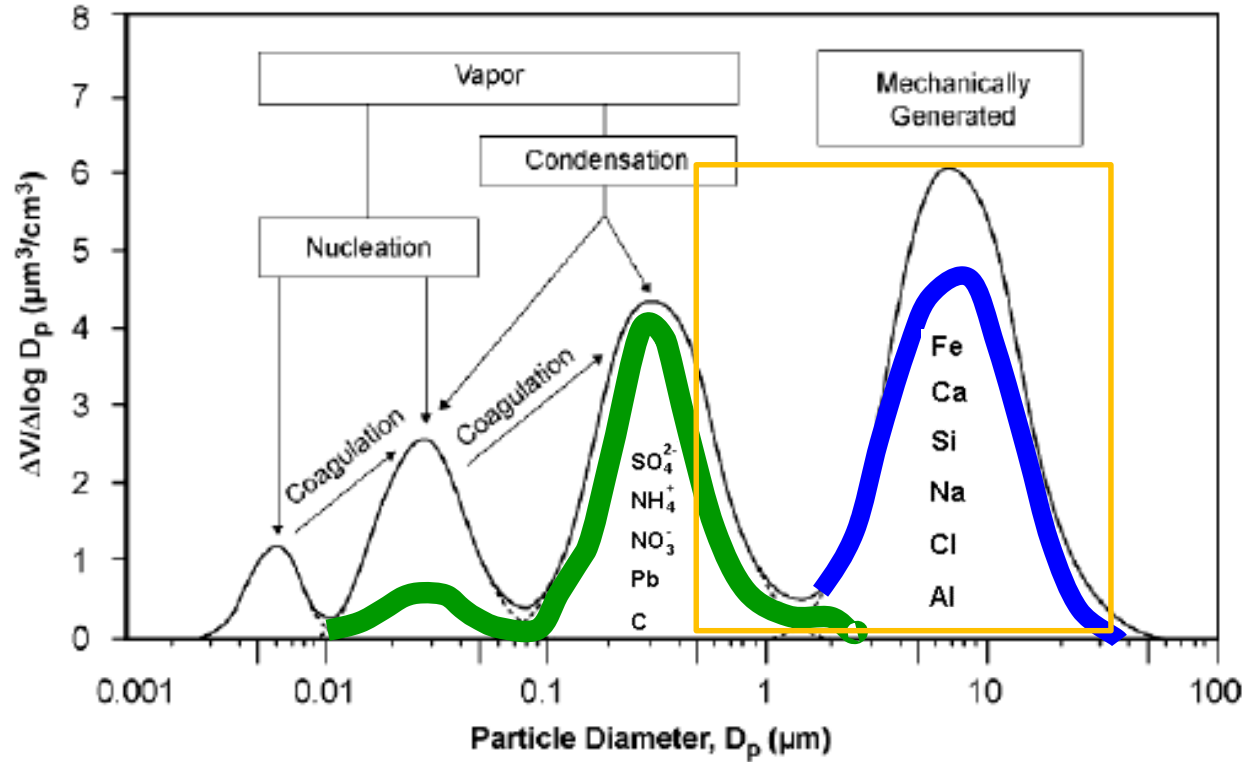
PM_{10} : Σ dust + sea salt + vehicle exhaust + oil refining + power plants + ships + ...

$PM_{2.5}$: Σ dust + sea salt + vehicle exhaust + oil refining + power plants + ships + ...

PM₁₀ (diameter <10 microm)

PM_{2.5}

PM_{2.5-10}



ultrafine
<0.1 μm

accumulation
0.1 - 1 μm

Coarse
1 - 10 μm

Mineral dust :

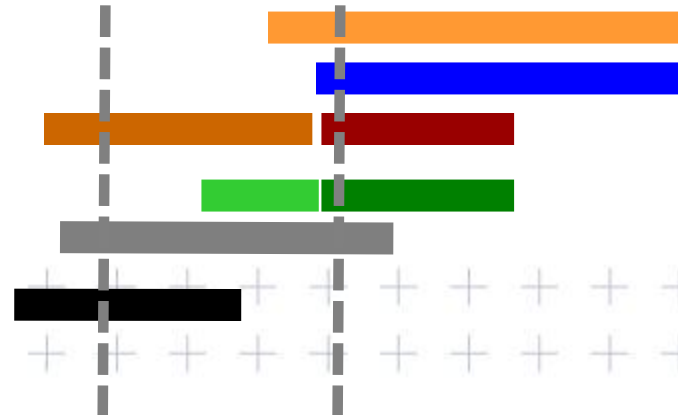
Marine salt:

Sulfate:

Nitrate:

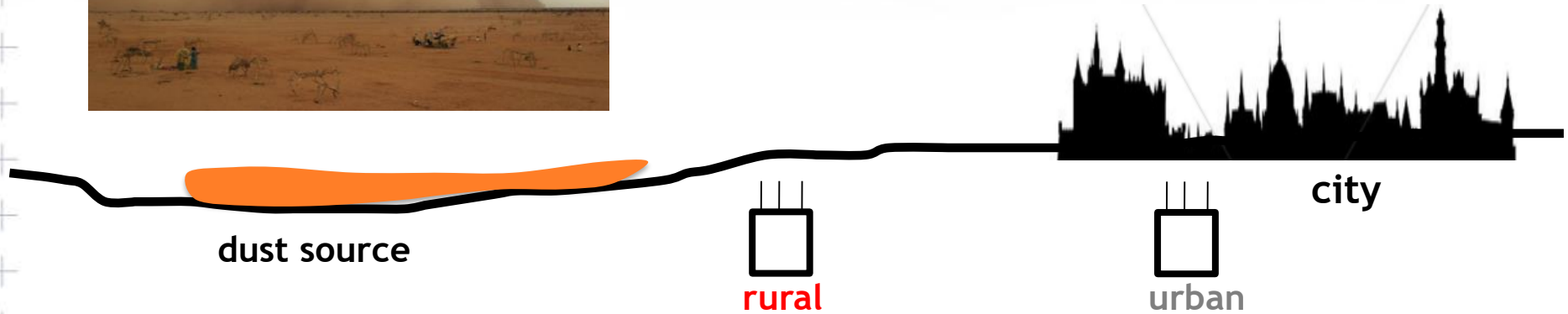
Organic aerosol:

black carbon:



dust, aerosols and pollutants

in-situ observations



how to measure dust aerosol

there are no standardized automatic method for measuring dust

alternatively we measure bulk aerosol PM...

PM_{10} : Σ dust + sea salt + (sulphate + nitrate + organic matter + black carbon +metals)

$PM_{2.5}$: Σ dust + sea salt + (sulphate + nitrate + organic matter + black carbon +metals)

dust, aerosols and pollutants

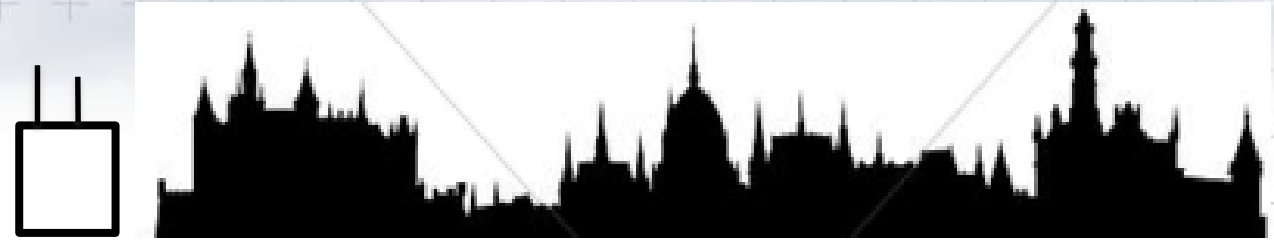
in-situ observations

PM₁₀ and PM_{2.5} levels

PM₁₀ and PM_{2.5} composition

complementary observations

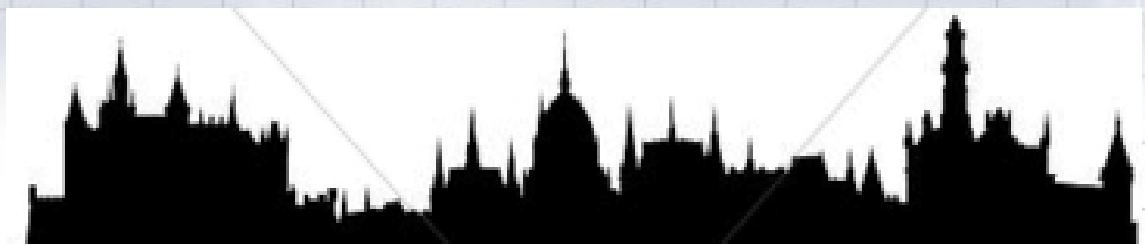
observation network



dust air quality

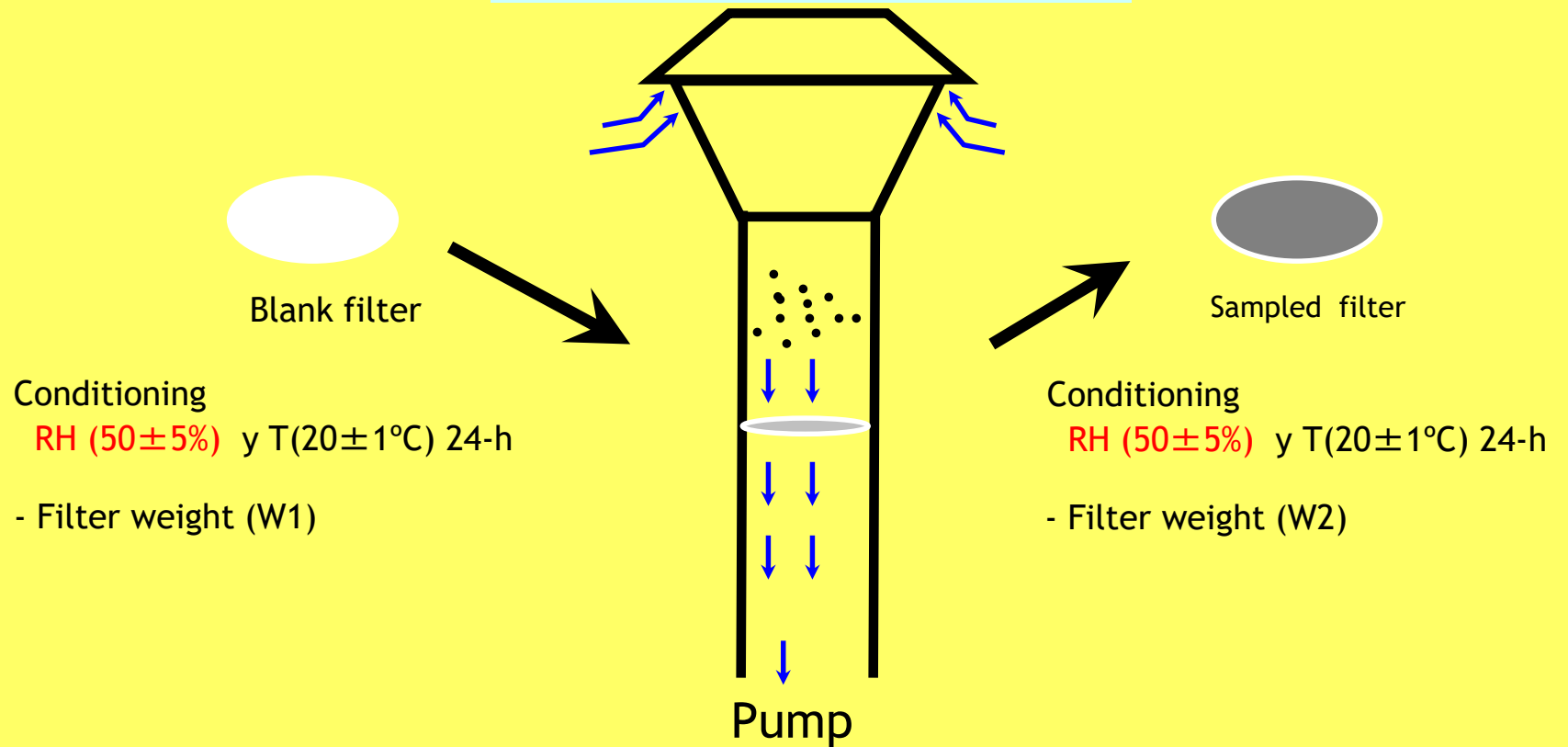
1. PM_{10} and $PM_{2.5}$ levels

-method-01 manual gravimetry method



-method-01: reference - manual gravimetry

$$PM = \frac{(W2 - W1)}{\text{Volume}} \mu\text{g}/\text{m}^3$$



It is recommended to use standardised protocols
national standard method
or already existing international standard methods

- PM₁₀ and PM_{2.5} sampler**
- sampling procedure**
- weighing procedure**

example:

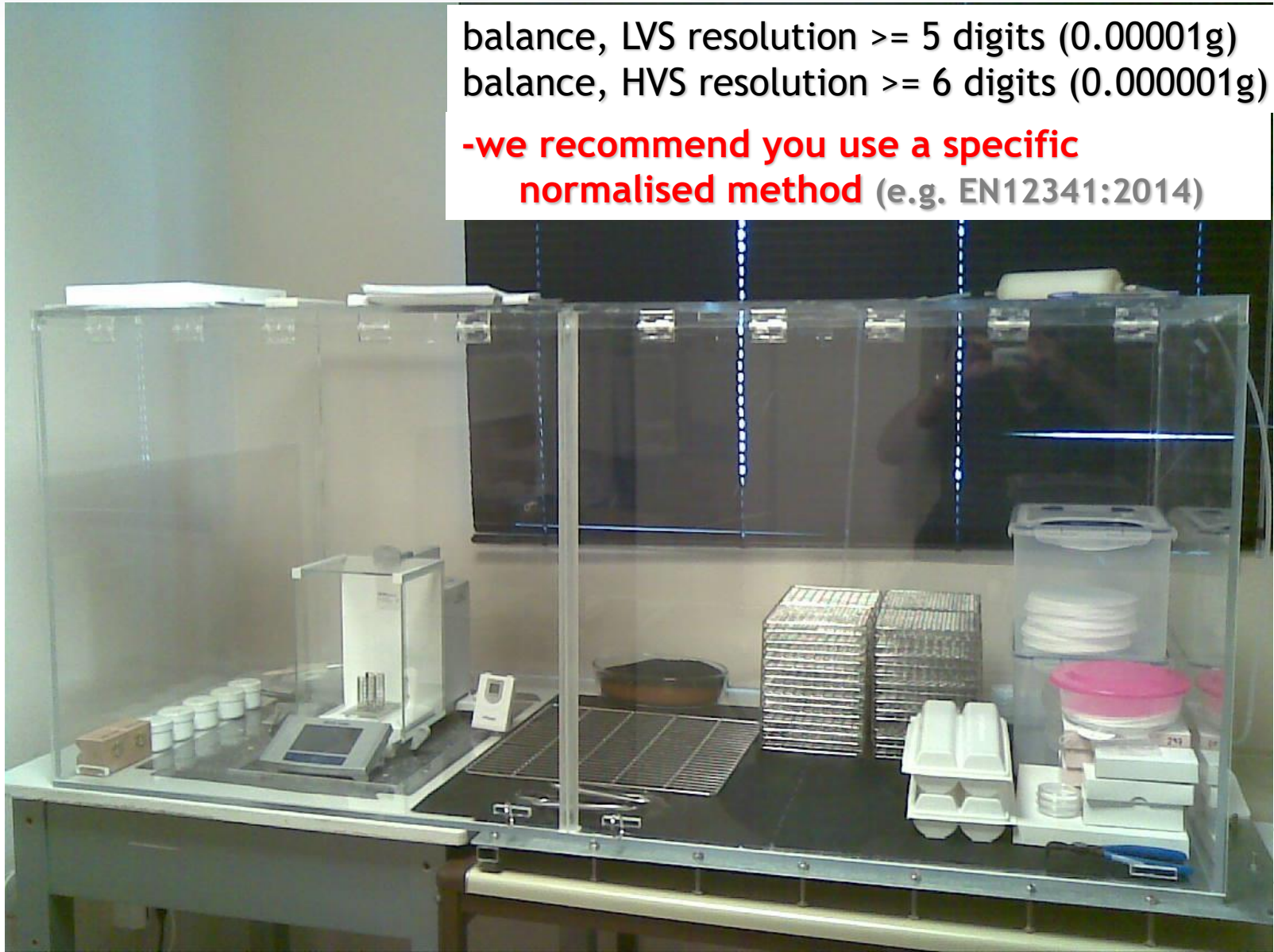
EN 12341:2014

Ambient air. Standard gravimetric measurement method for the determination of the PM₁₀ or PM_{2,5} mass concentration of suspended particulate matter

Room for weighting the filters: RH =50% (30 %) and 20°C

balance, LVS resolution ≥ 5 digits (0.00001g)
balance, HVS resolution ≥ 6 digits (0.000001g)

-we recommend you use a specific normalised method (e.g. EN12341:2014)



PM_{10}
Blank filter

PM_{10}
sample urban air

PM_{10}
sample in dust days



-we recommend you use a specific
normalised method (e.g. EN12341:2014)

Filters: Quartz, Teflon, Cellulose

Low Volume Sampler

LVS: **2.3 m³/h**



High Volume Sampler

HVS: **68 m³/h**



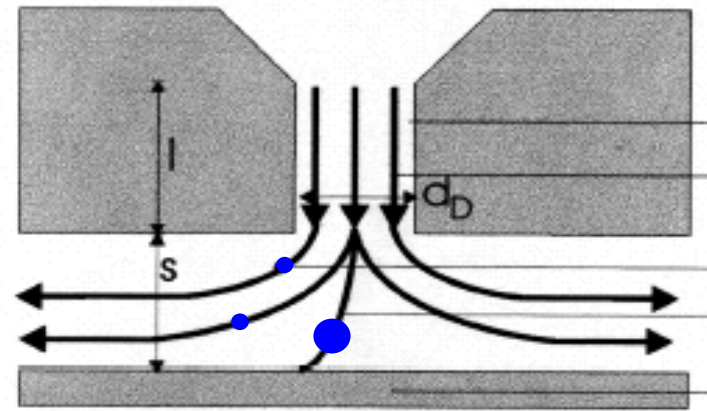
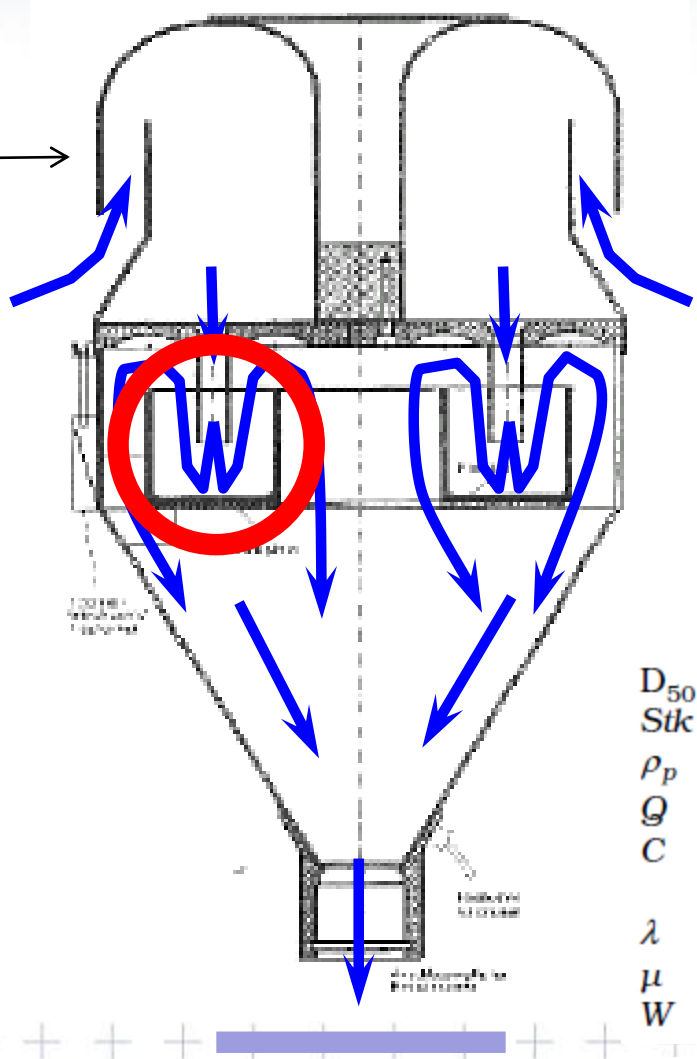
HVS: **30 m³/h**



-we recommend you use a specific normalised method (e.g. EN12341:2014). Ask to the distributor if the sampler is designed to any standards

Inlets, airflows....

PM₁₀, PM_{2.5}



$$D_{50} = \sqrt{\frac{9\pi Stk \mu W^3}{4\rho_p C Q}}$$

- D_{50} = particle cut-point diameter centimeter
- Stk = Stokes number = 0.23
- ρ_p = particle density (g/cm³)
- Q = volumetric flow rate (cm³/s)
- C = Cunningham slip correction
 $= 1 + 2.492 \lambda/D_{50} + 0.84 \lambda/D_{50} \exp(-0.435 D_{50}/\lambda)$
- λ = gas mean free path
- μ = gas viscosity (dyne•s/cm²)
- W = nozzle diameter (cm)

The Stokes number is a dimensionless parameter that characterizes impaction.

Filter



dust air quality

1. PM_{10} and $PM_{2.5}$ levels

-method-01: reference - manual gravimetry

Manual gravimetry

advantage: reference method

disadvantage: poor time resolution, 24-h average
manual work
takes 3 days to know PM_{10} concentration

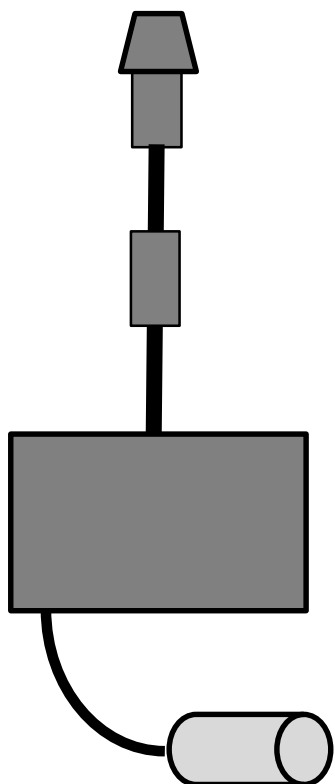


dust air quality

1. PM_{10} and $PM_{2.5}$ levels

- method-01: reference - manual gravimetry
- method-02: automatic beta, teom, OPS

-method-02: automatic



1. Impactor PM_{10} / $PM_{2.5}$

2. RH reductor / heater

3. Sensor

Beta radiation attenuation

4. Pump / Flow meter

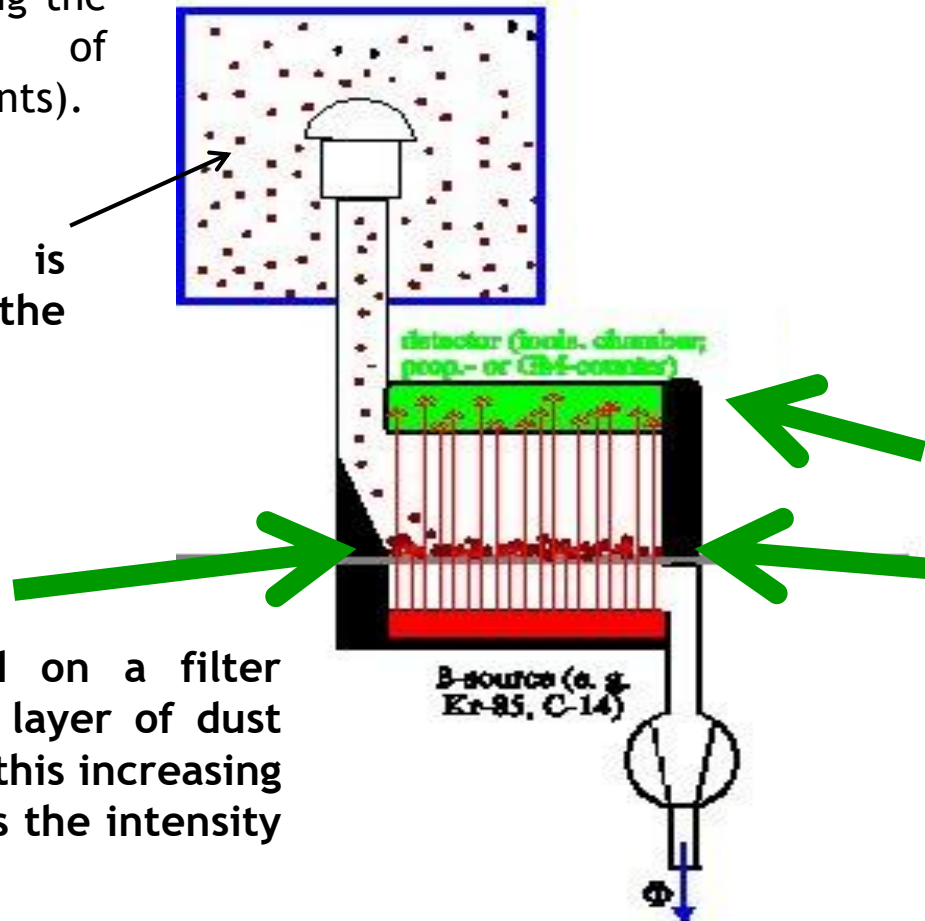
Continuous measurements of PM (PM_{10} , $PM_{2.5}$, PM_1 or TSP)

PM with Beta attenuation

Krypton-85 or Carbon-14 is used as source of beta radiation (emitted by electrons during the nuclear decay of radioactive elements).

*Beta Attenuation:
β-Ray Absorption in Matter*

Ambient air is drawn through the sample system



Beta rays detector
 Beta rays source (Kr-85)

Dust is deposited on a filter continuously. The layer of dust is building up and this increasing dust mass weakens the intensity of the beta beam.

Pump and flowmeter

PM with Beta attenuation

Krypton-85 or Carbon-14 is used as source of beta radiation (emitted by electrons during the nuclear decay of radioactive elements).

Beta Attenuation:
 β -Ray Absorption in Matter

$$I = I_0 e^{-\mu_\beta \cdot x}$$

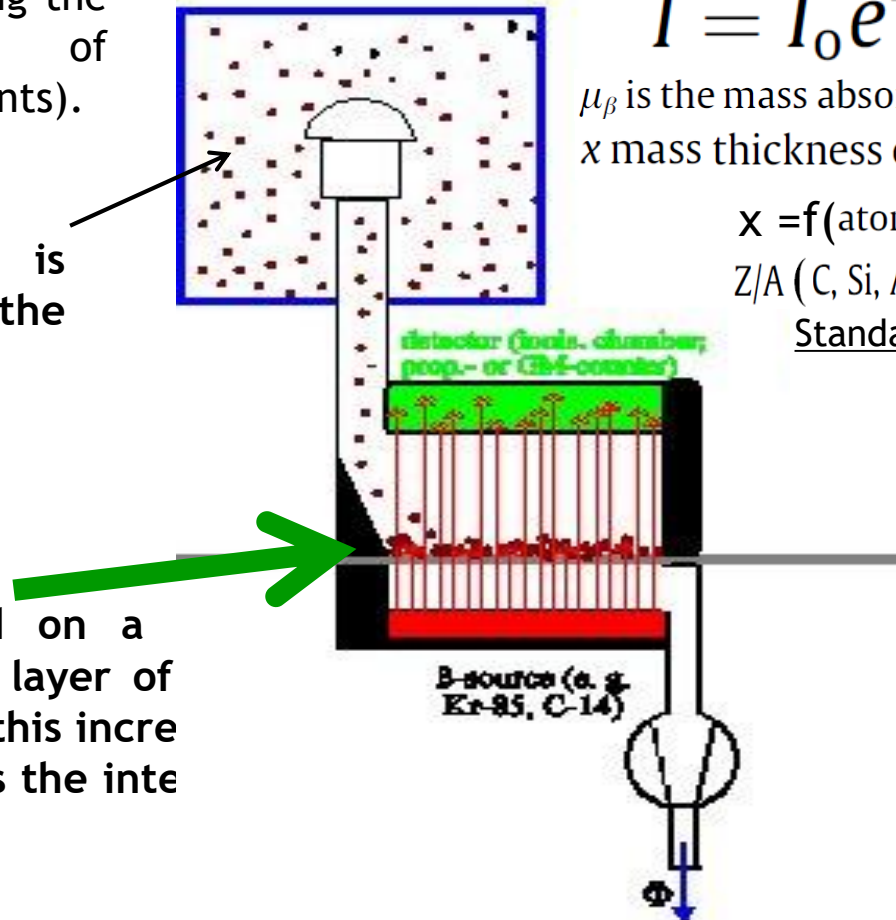
μ_β is the mass absorption coefficient for beta radiation
 x mass thickness of the sample

$x = f(\text{atomic number to atomic mass ratio } (Z/A))$
 Z/A (C, Si, Al, Ca, Fe, Mg, K, Cl, Na, N, O and S) 0.47-0.50
Standard foil calibration

typical elements of aerosols; fixed Z/A ratio: error of about 10%

Ambient air is drawn through the sample system

Dust is deposited on a continuously. The layer of is building up and this incre dust mass weakens the inte of the beta beam.



Pump and flowmeter

PM with Beta attenuation (2)

$$m = F_{cal} \ln \left(\frac{I_0}{I} \right)$$

- **m**: increasing particle mass [μg]
- **F_{cal}**: calibration factor
- **I₀** beta ray intensity at empty filter
- **I** beta ray intensity at loaded filter

The intensities I_0 and I are measured with the detector system. F_{cal} has to be measured directly during the calibration procedure. This is accomplished by replacing the filter with the element having a known mass (mass calibration kit)

The mass concentration is calculated from:

$$PM_{10} \ \& \ PM_{2.5} \approx c = \frac{m}{Ft}$$

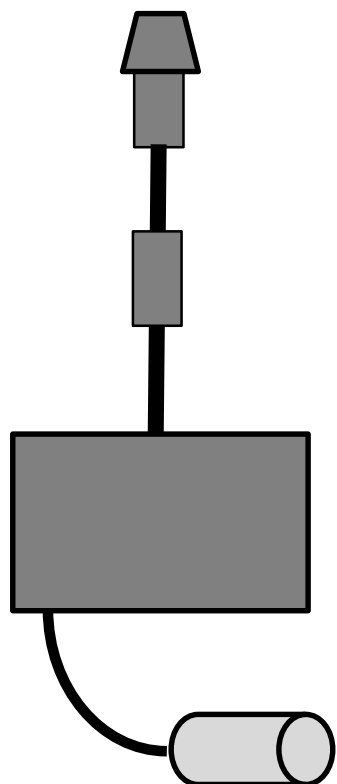
Where:

c: concentration [$\mu\text{g}/\text{m}^3$]

F: measured air flow [m^3/h]

t: time [h]

-method-02: automatic



1. Impactor PM₁₀ / PM_{2.5}

2. RH reductor / heater

3. Sensor

Beta radiation attenuation
TEOM

4. Pump / Flow meter

Continuous measurements of PM (PM₁₀, PM_{2.5}, PM₁ or TSP)

Mass concentration

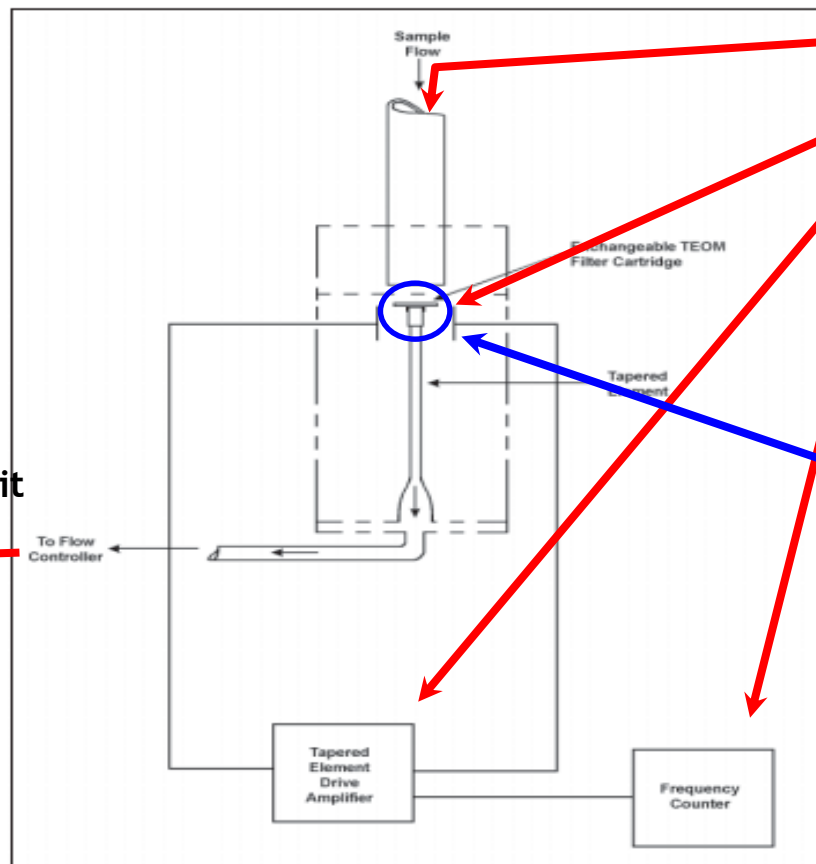
Automatic continuous measurements

TEOM : Tapped Element Oscillating Microbalance

1. TEOM mod. 1400a

mass=function (frequency)

sensor



Sampling flow rate (16.67 l/m)

Sample accumulated in the filter

Micro-oscilation of constante amplitue
GENERATOR

Frequency sensor

An increase in the amount of sample
(dust) accumulated in the filter →
decrease in the oscillation frequency

Mass concentration

Automatic continuous measurements

TEOM : Tapped Element Oscillating Microbalance

1. TEOM mod. 1400a

sensor

mass=function (frequency)

more dust → lower oscillation frequency

In a spring-mass system the frequency follows the equation:

$$f = (K / M)^{0.5}$$

where:

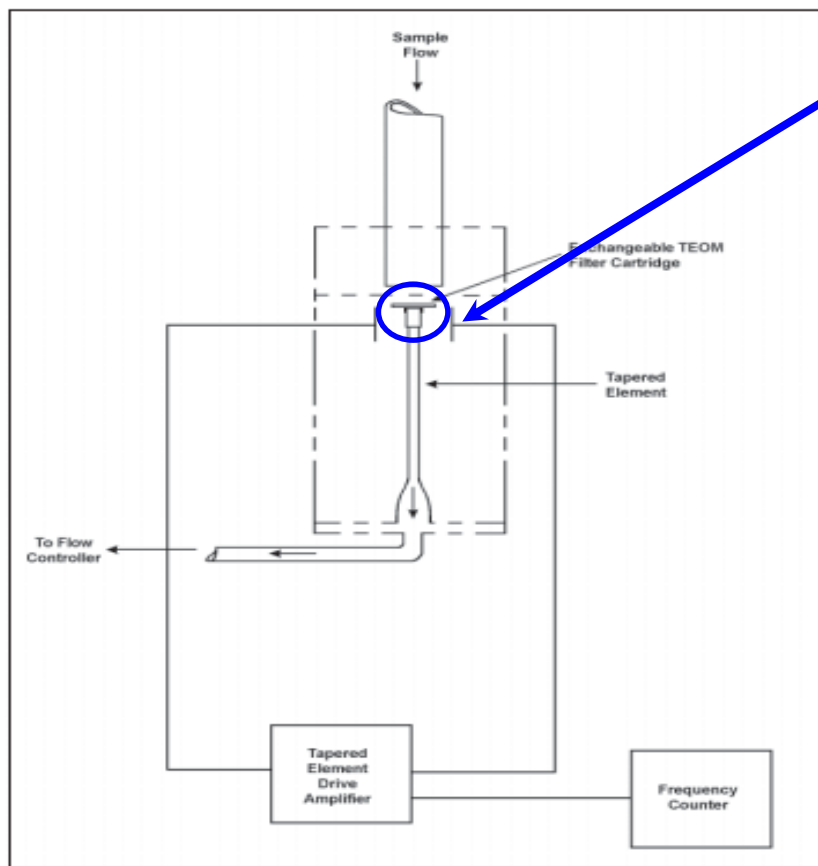
- f = frequency (radians/sec)
- K = spring rate
- M = mass

K and M are in consistent units. The relationship between mass and change in frequency can be expressed as:

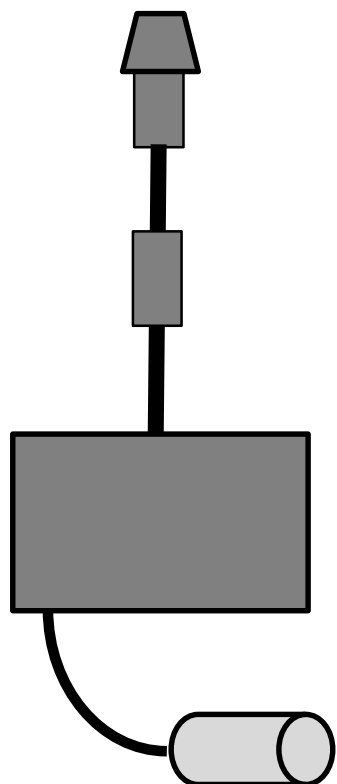
$$dm = K_0 \left(\frac{1}{f_1^2} - \frac{1}{f_0^2} \right) \tag{2}$$

where:

- dm = change in mass
- K₀ = spring constant (including mass conversions)
- f₀ = initial frequency (Hz)
- f₁ = final frequency (Hz)



-method-02: automatic



1. Impactor PM_{10} / $PM_{2.5}$

2. RH reductor / heater

3. Sensor

Beta radiation attenuation
TEOM

Optical Particle Sizers

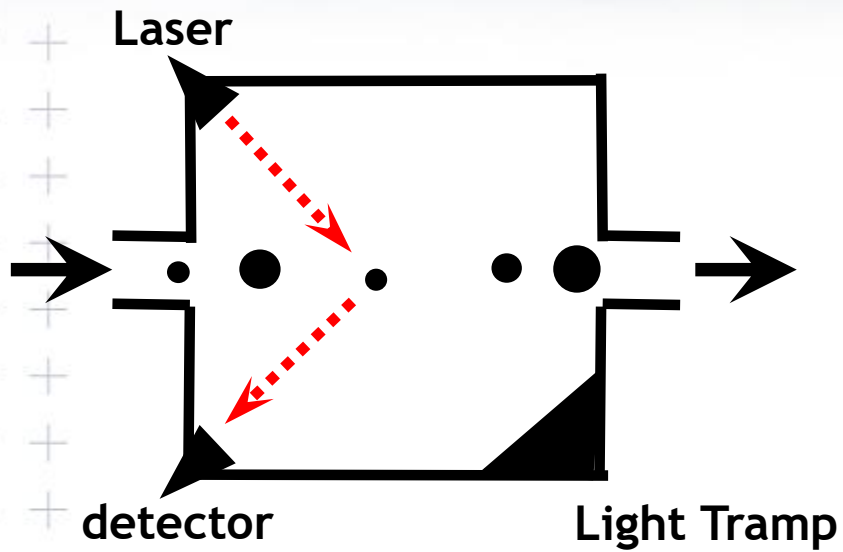
4. Pump / Flow meter

Continuous measurements of PM (PM_{10} , $PM_{2.5}$, PM_1 or TSP)

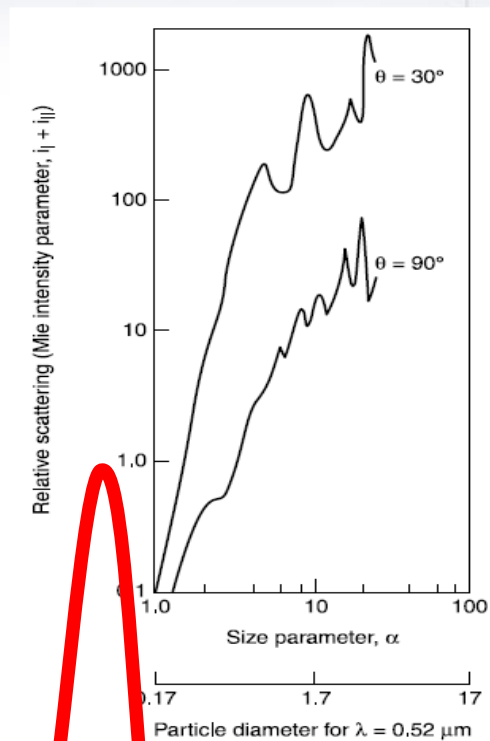
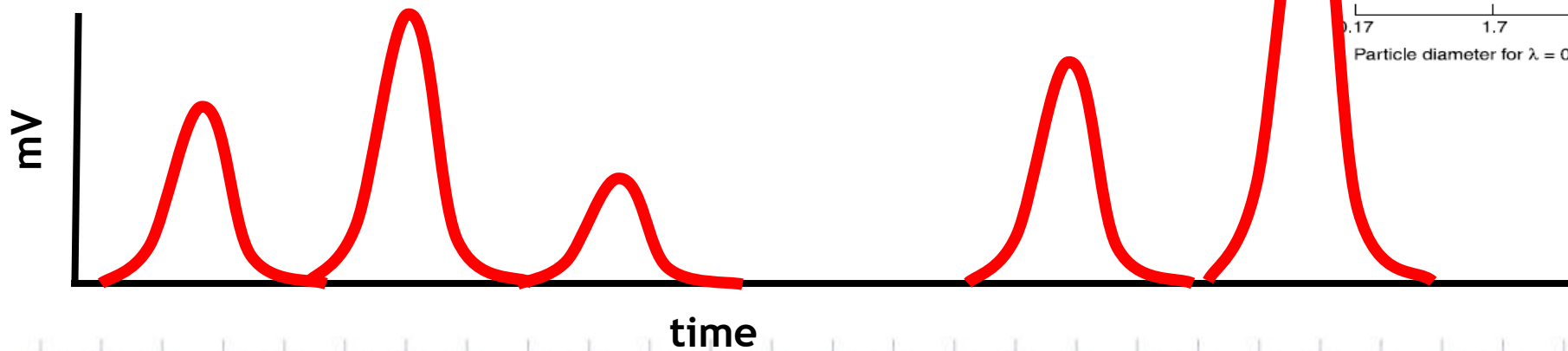
Optical Particle Sizer

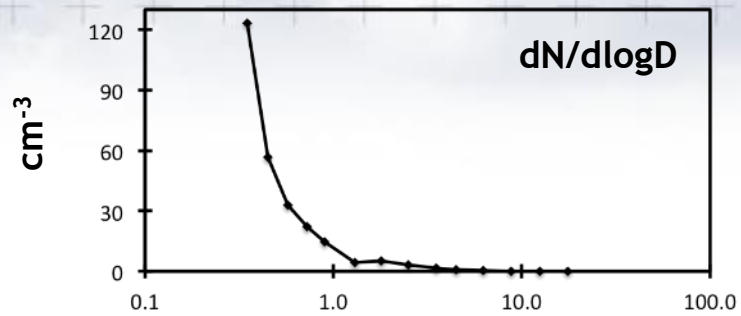
number size distribution 0.3 - 20 μm



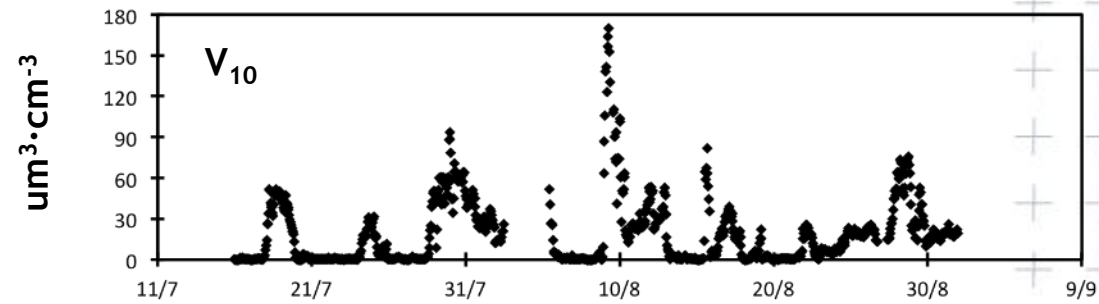
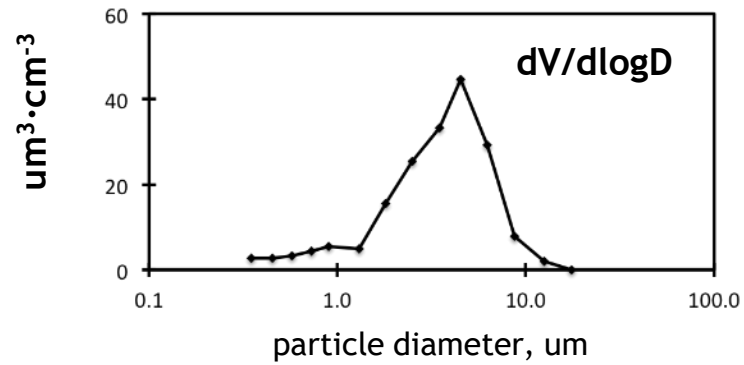


Intensity of scattering
 $I(dp, \theta, \lambda, m)$





$$\downarrow \quad \frac{3}{4} \cdot \pi \cdot r^3$$



$$PM_{10} = V_{10} \cdot \text{density}$$

Density: 1.6 to 2.65 g/cm^3

-method-02: automatic

The most extended method and the most robust for dusty regions

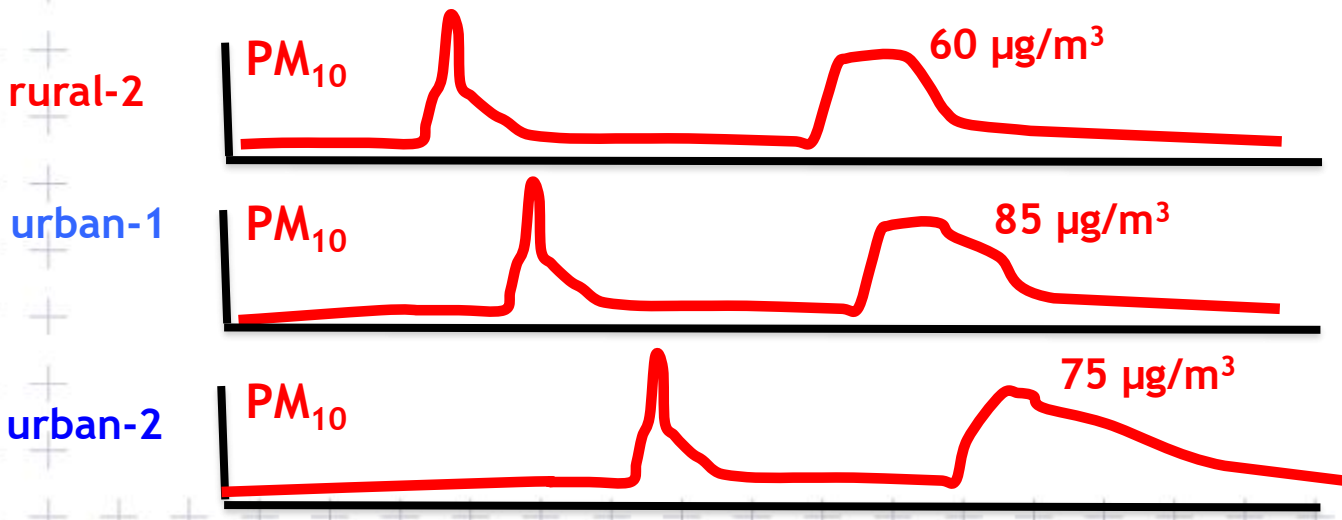
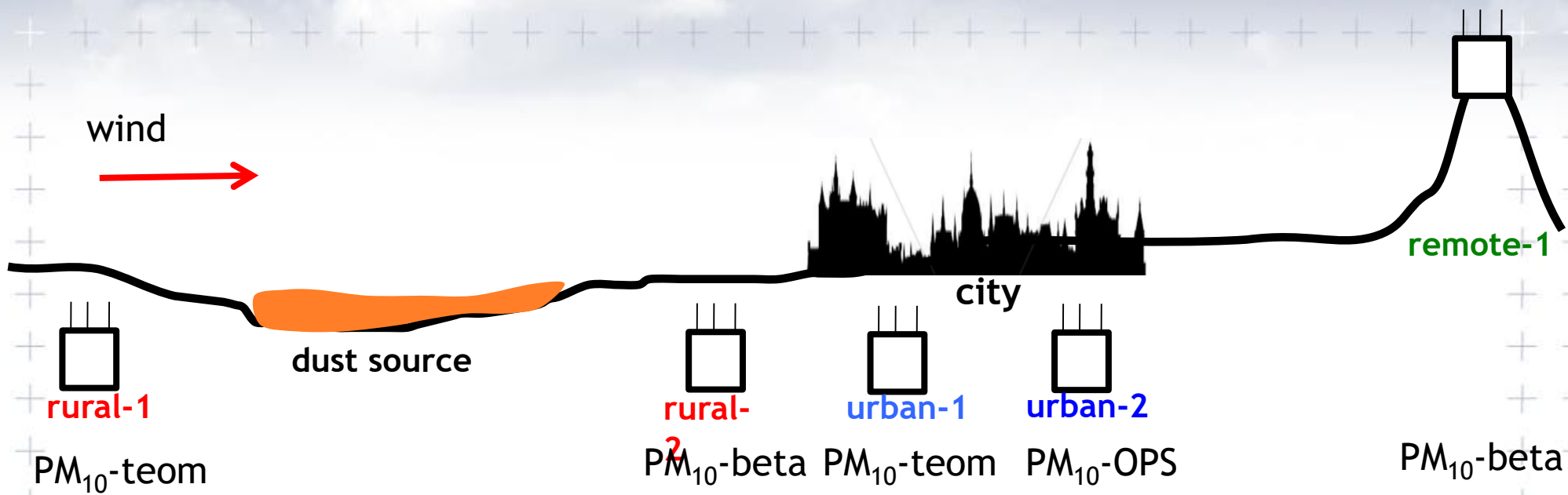
beta



Tapered Oscillating Microbalance
TEOM
Manual change of the filter

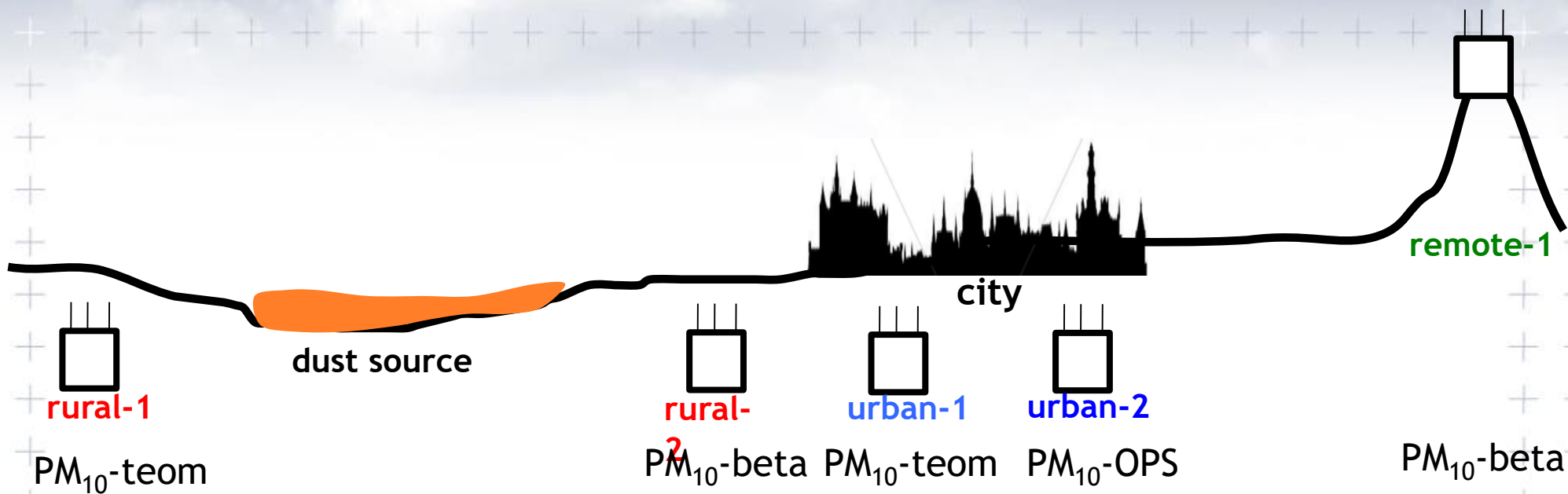


Optical Particle Counters
cleaning of optics
laser maintenance



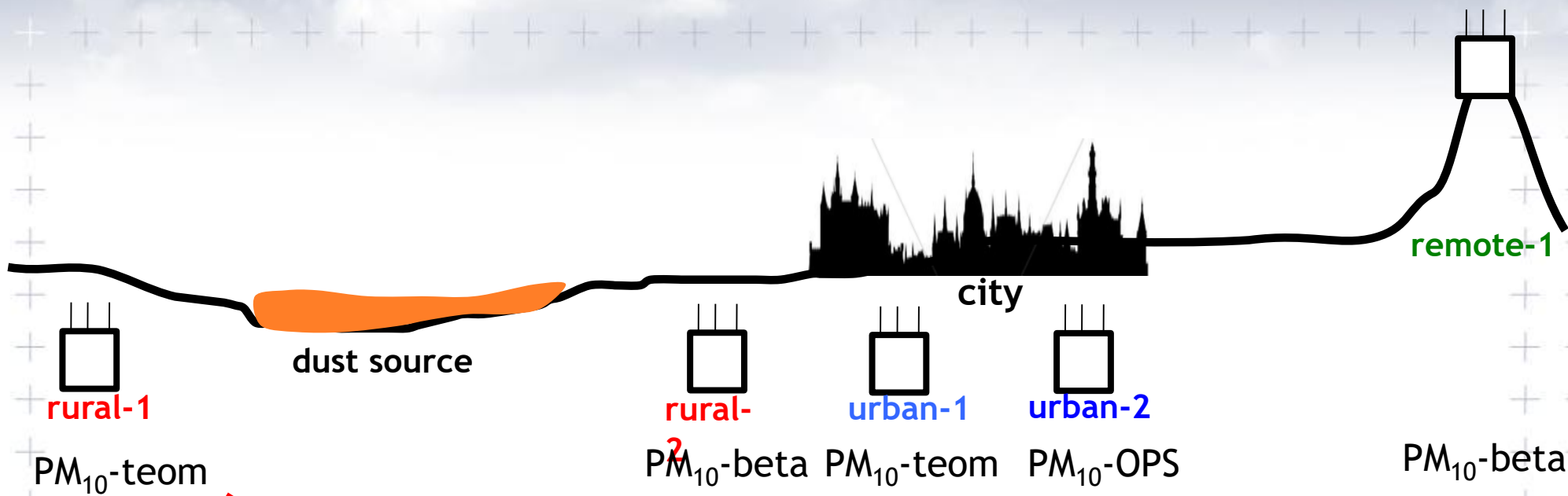
are PM₁₀ data collected with different methods comparable?

we need a standard for the network

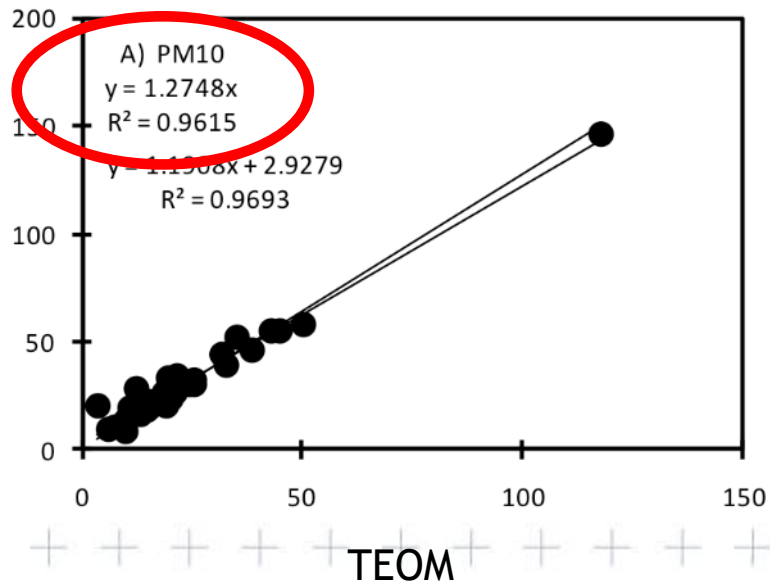


the standard in the network:

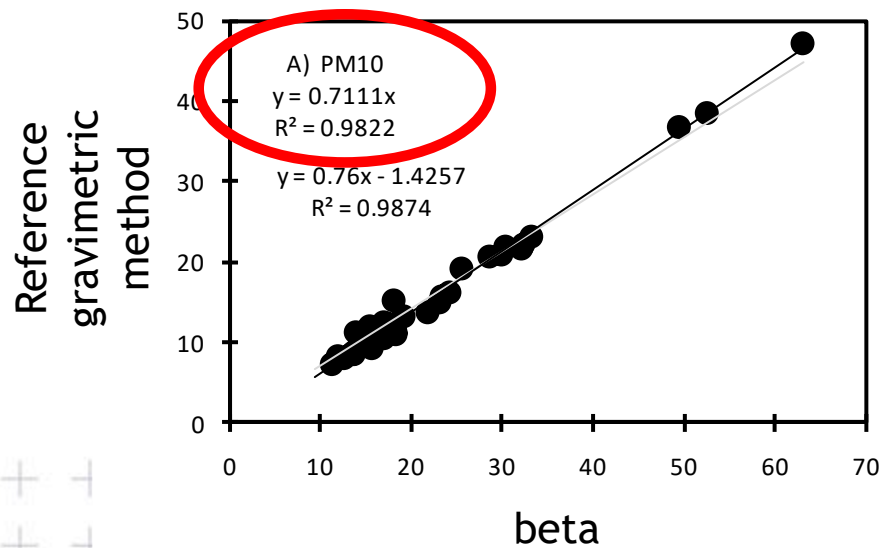
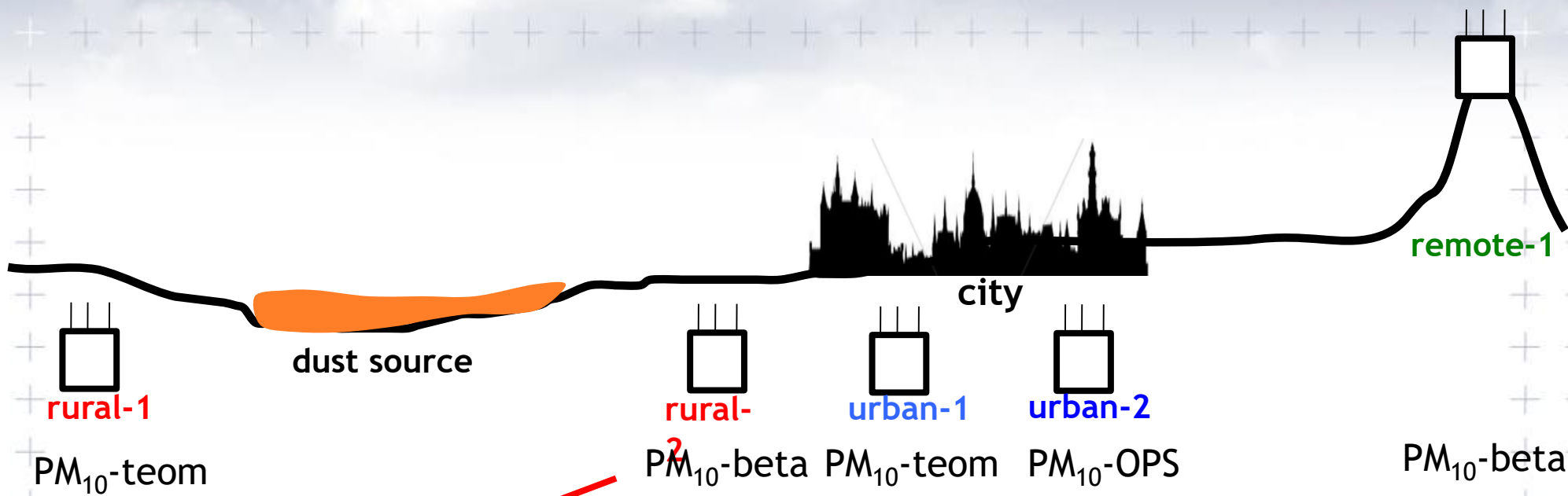
intercomparison of each automatic instrument with the manual reference method



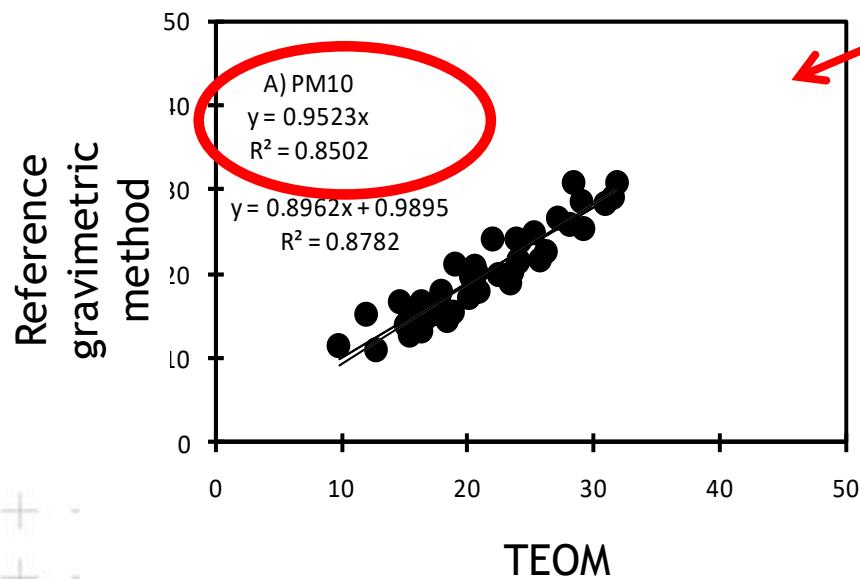
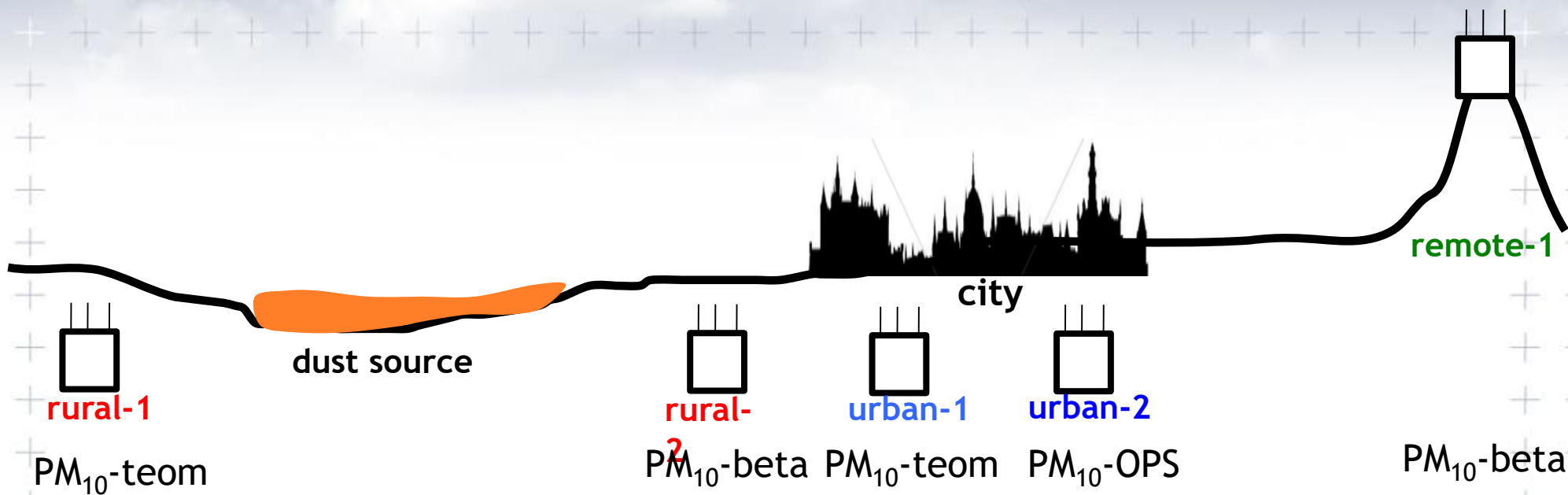
Reference gravimetric method



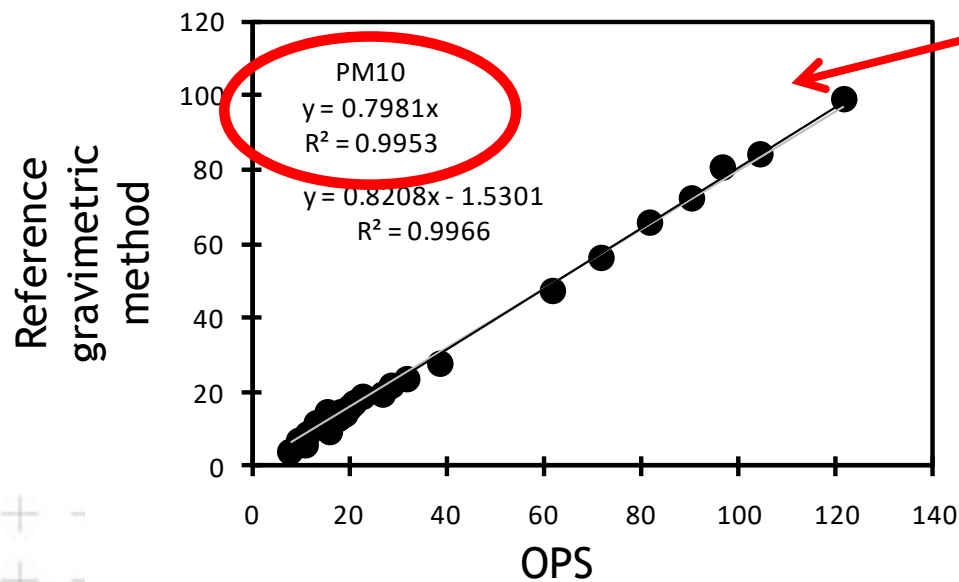
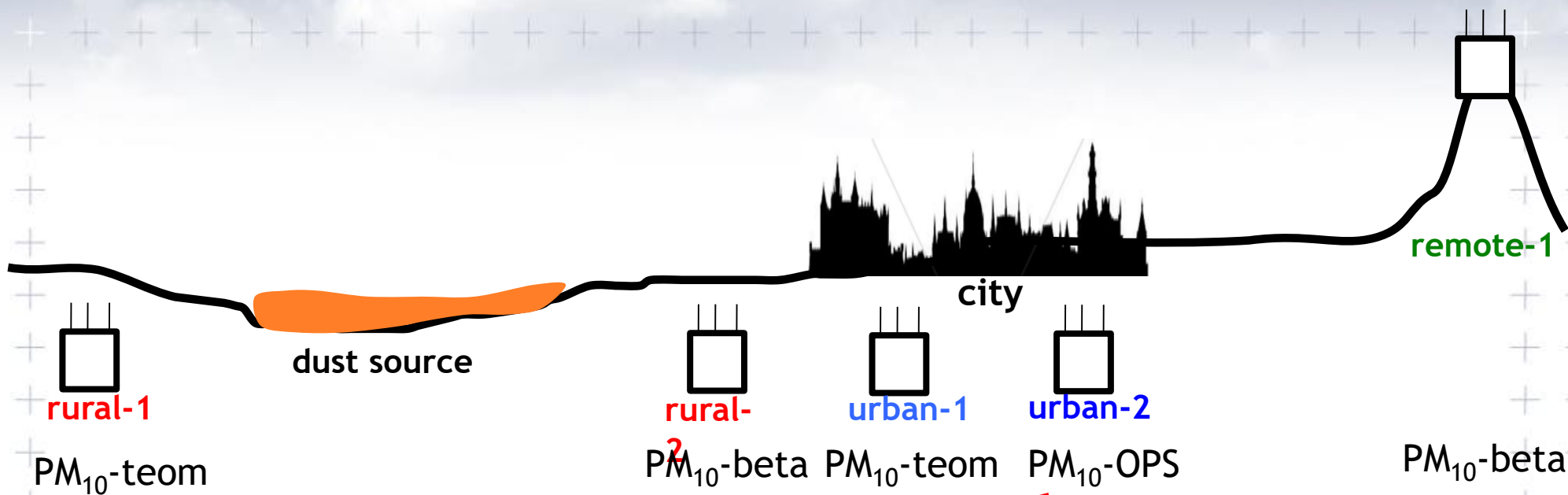
PM_{10} (grav equiv) = 1.27 PM_{10} (TEOM)
 Valid for rural-1 TEOM



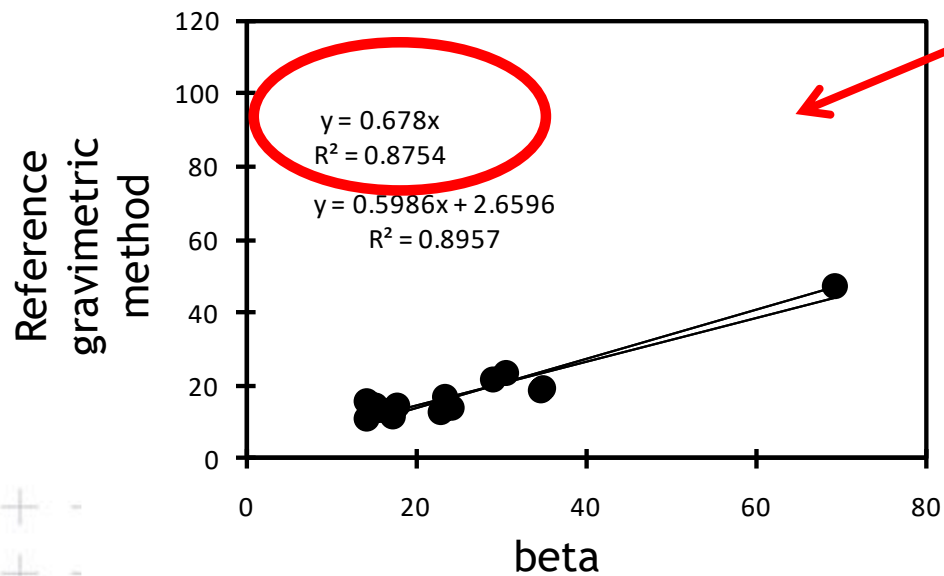
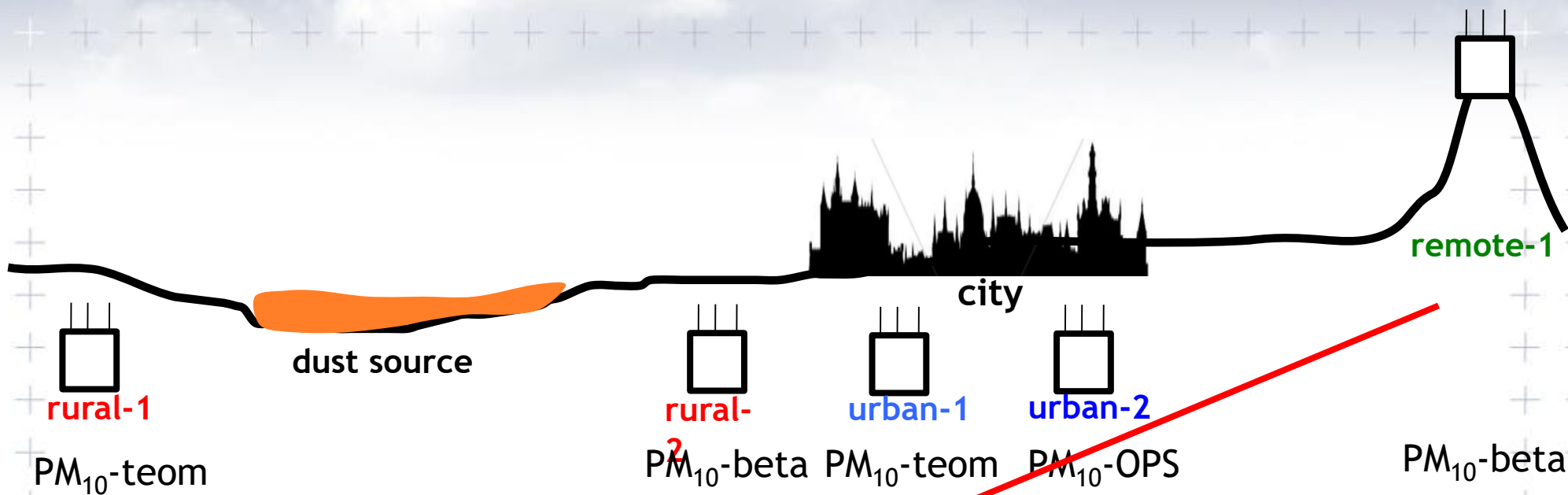
PM_{10} (grav equiv) = 0.71 PM_{10} (BETA)
 Valid for rural-2 BETA



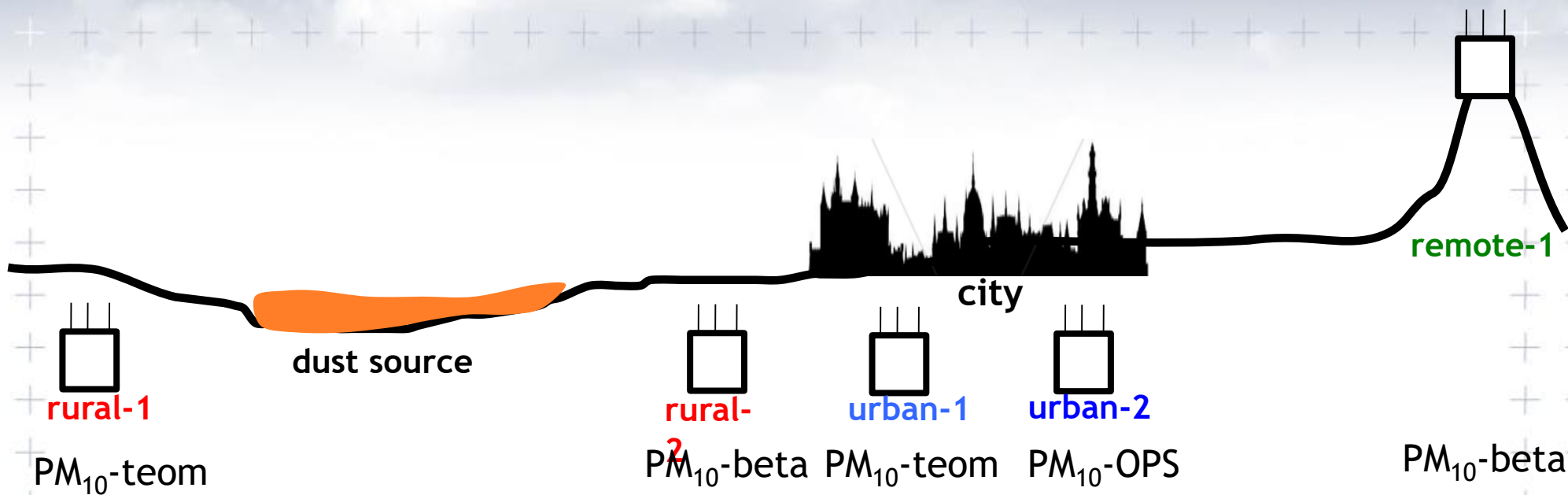
**PM₁₀ (grav equiv) = 0.95 PM₁₀ (TEOM)
Valid for urban-1 TEOM**



**PM₁₀ (grav equiv) = 0.79 PM₁₀ (OPS)
Valid for urban-2 OPS**



PM₁₀ (grav equiv) = 0.67 PM₁₀ (BETA)
Valid for remote-1 BETA



Standardized data

raw data

rural-1

$$PM_{10} \text{ (grav equiv)} = 1.27 PM_{10} \text{ (TEOM)}$$

rural-2

$$PM_{10} \text{ (grav equiv)} = 0.71 PM_{10} \text{ (BETA)}$$

urban-1

$$PM_{10} \text{ (grav equiv)} = 0.95 PM_{10} \text{ (TEOM)}$$

urban-2

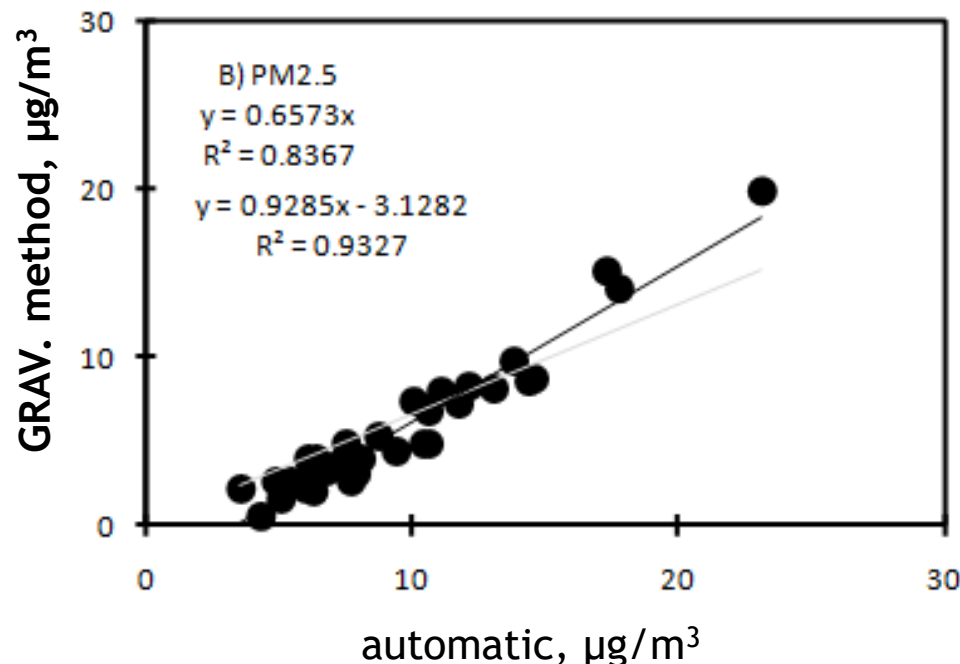
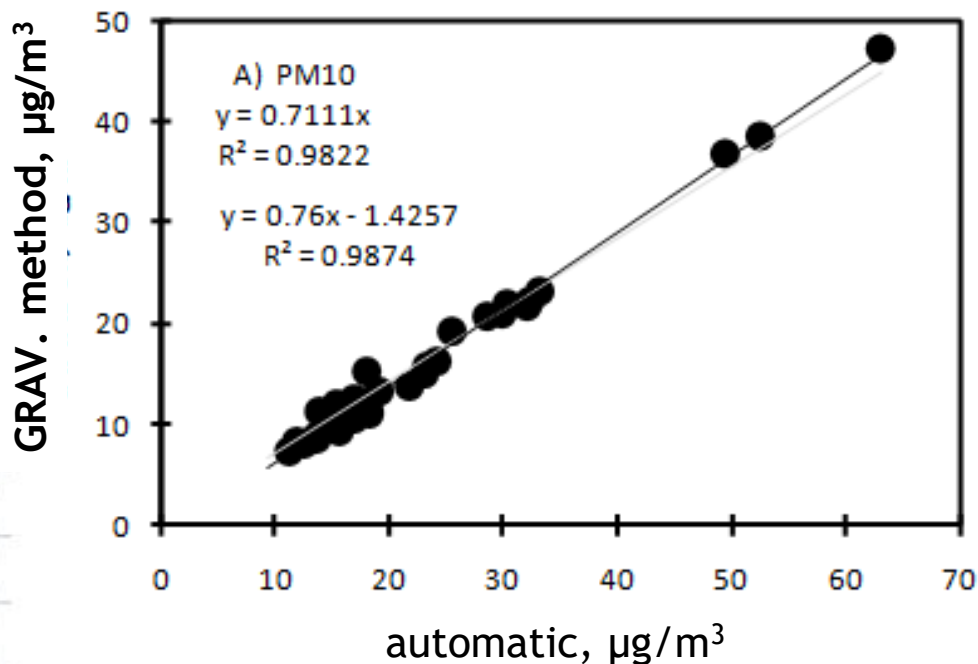
$$PM_{10} \text{ (grav equiv)} = 0.79 PM_{10} \text{ (OPS)}$$

remote-1

$$PM_{10} \text{ (grav equiv)} = 0.67 PM_{10} \text{ (BETA)}$$

Validation of the automatic measurements

Intercomparisons for calibrations



Data evaluation:

automatic data are valid if they fit A or B:

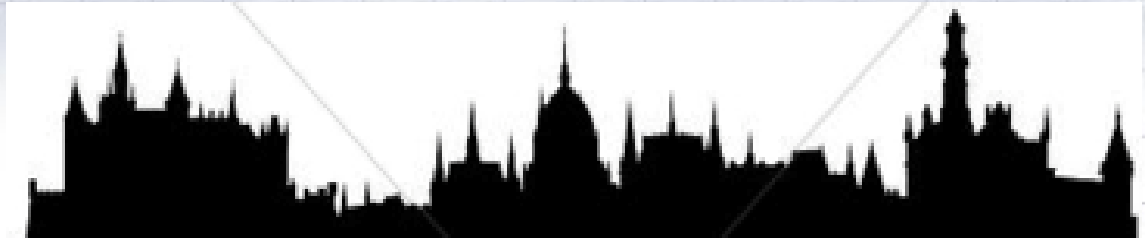
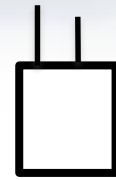
A) $Y = a \cdot X; r^2 \geq 0.8$

B) $Y = a \cdot X + b; r^2 \geq 0.8; \text{abs}(b) < 5$

Y= gravimetric method,
 X= Automatic analyzer

$PM_{10} \text{ (grav)} = 0.71 \cdot PM_{10} \text{ (automatic)}$

$PM_{2.5} \text{ (grav)} = 0.65 \cdot PM_{2.5} \text{ (automatic)}$



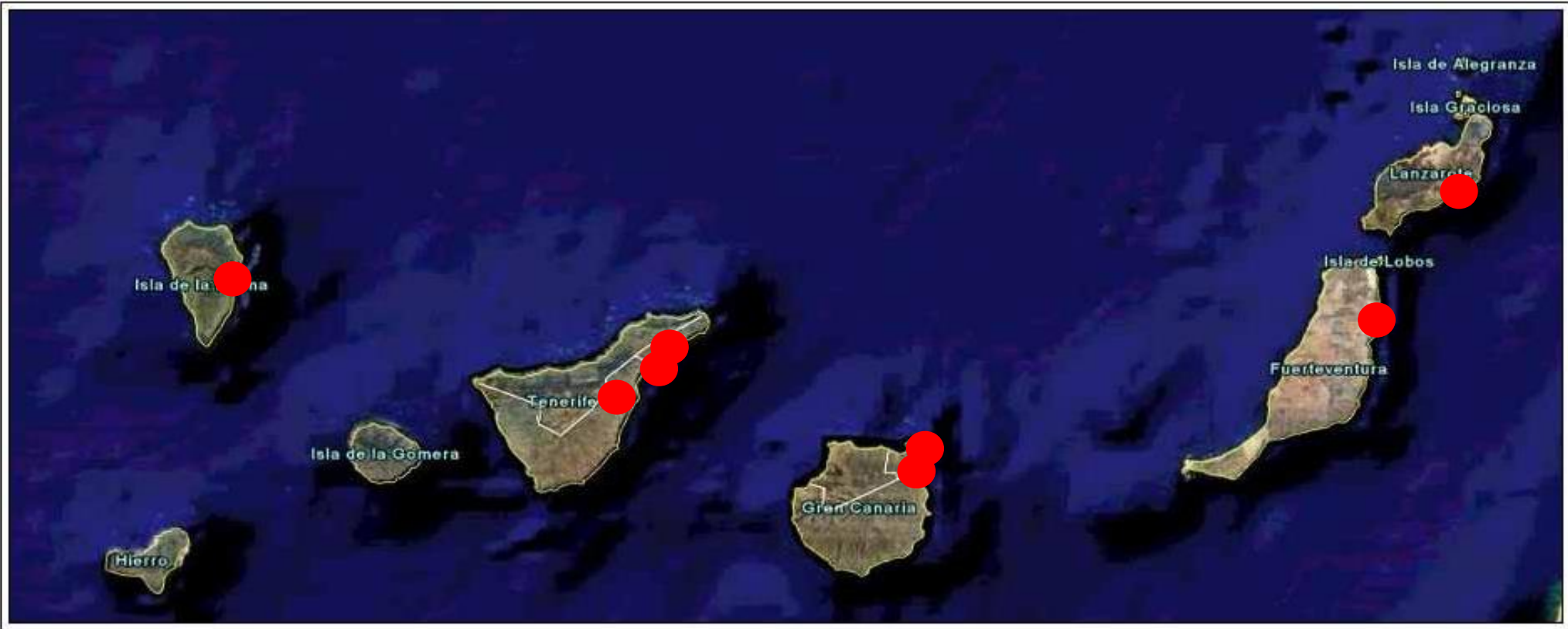
dust air quality

1. PM_{10} and $PM_{2.5}$ levels

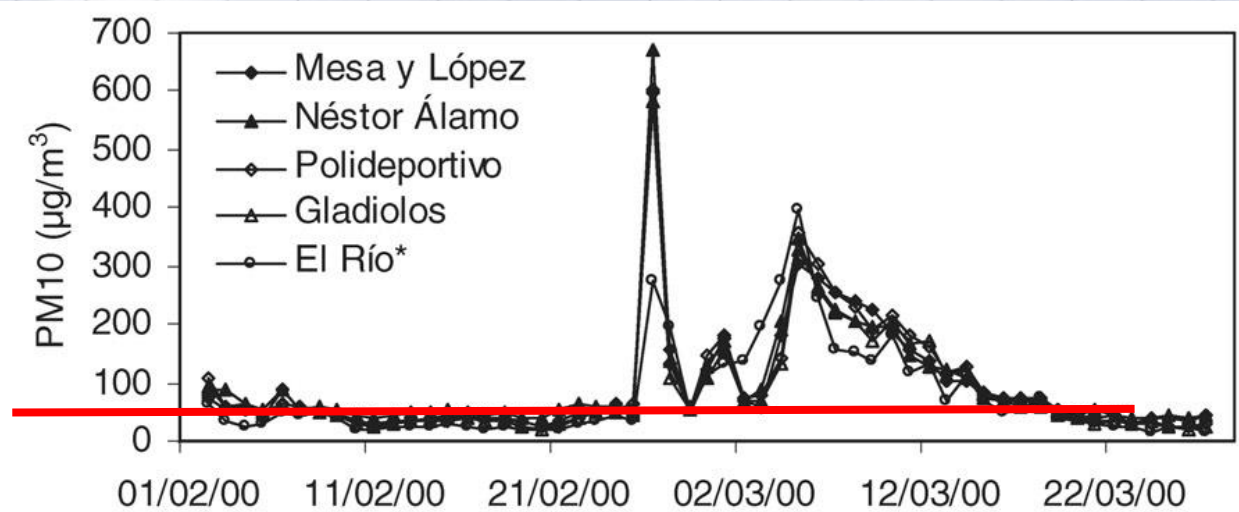
- method-01: reference - manual gravimetry
- method-02: automatic

We recommend to convert PM_{10} and $PM_{2.5}$ data obtained with automatic instruments to gravimetric equivalent data.
For this a standard obtained with intercomparisons is necessary

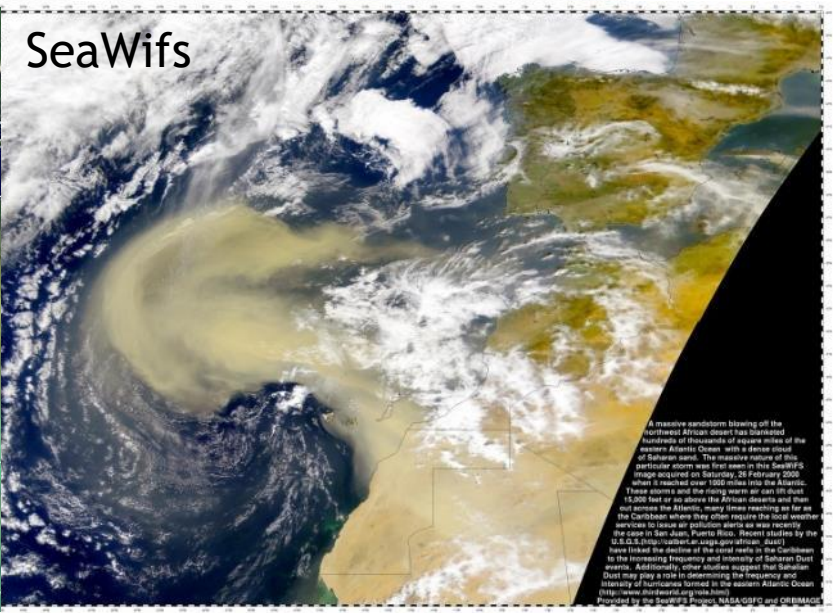
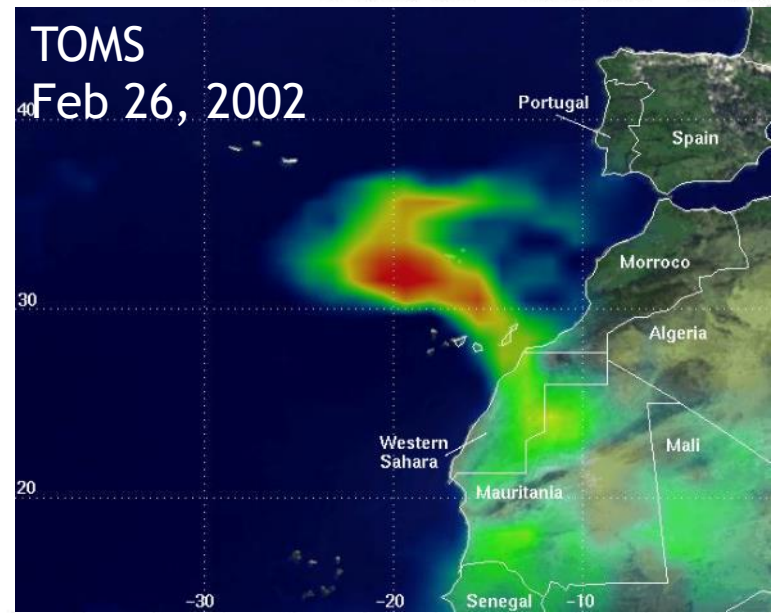
Standardization of PM_{10} y $PM_{2.5}$ in a regional network



Air quality stations at Tenerife Island



The WHO recommend PM_{10} (24-h) do not exceed $50 \mu\text{g}/\text{m}^3$



Viana et al., Atmospheric Environment, 2002

Standardization of PM_{10} y $PM_{2.5}$ in a regional network





samplers of PM₁₀ and PM_{2.5}

room of conditioning and weighting filters

1 month in summer (30 days) sampling
1 month in winter (30 days) sampling
at each station



samplers of PM₁₀ and PM_{2.5}

ARAFO

GLADIOLOS

CIUDAD DEPORTIVA



MERCADO CENTRAL

TOME CANO



REHOYAS

TELDE



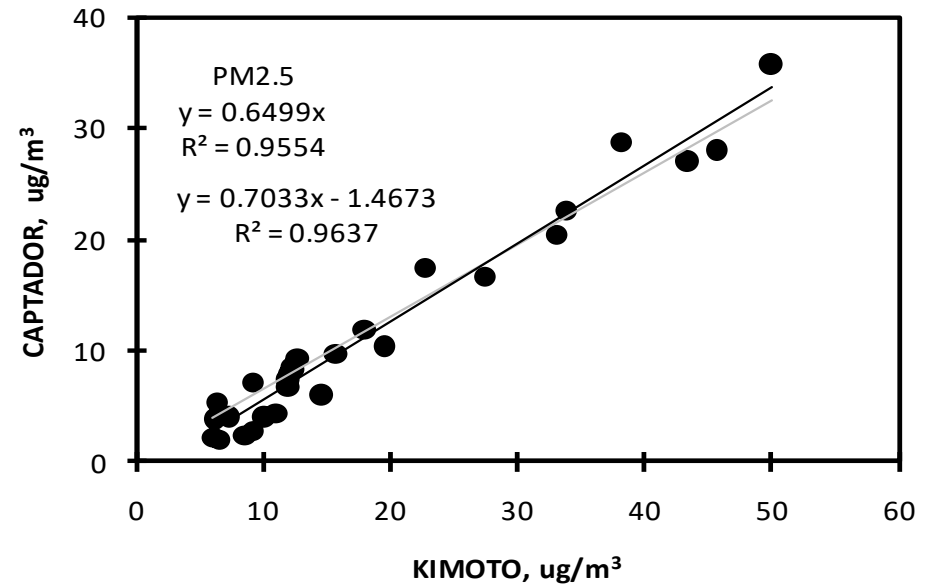
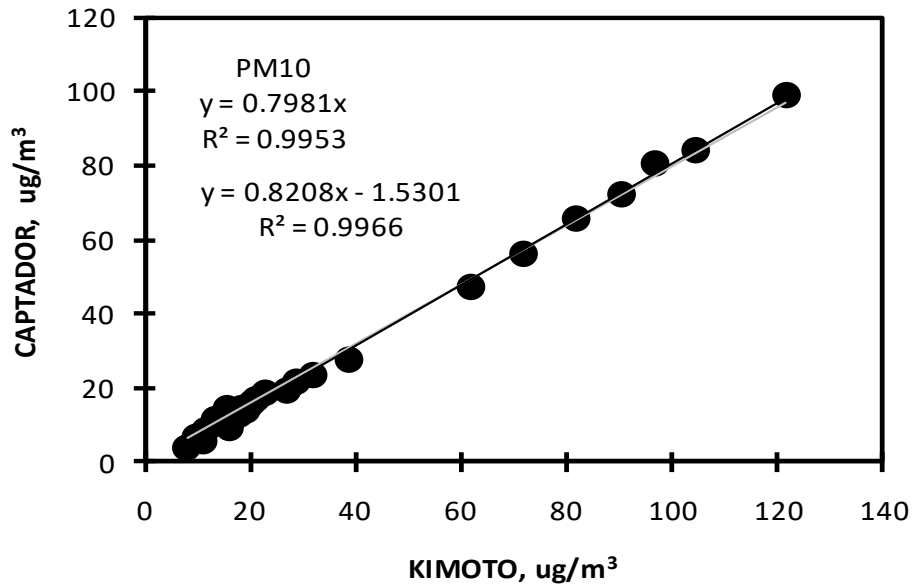
QUALITY CONTROL

SAMPLER

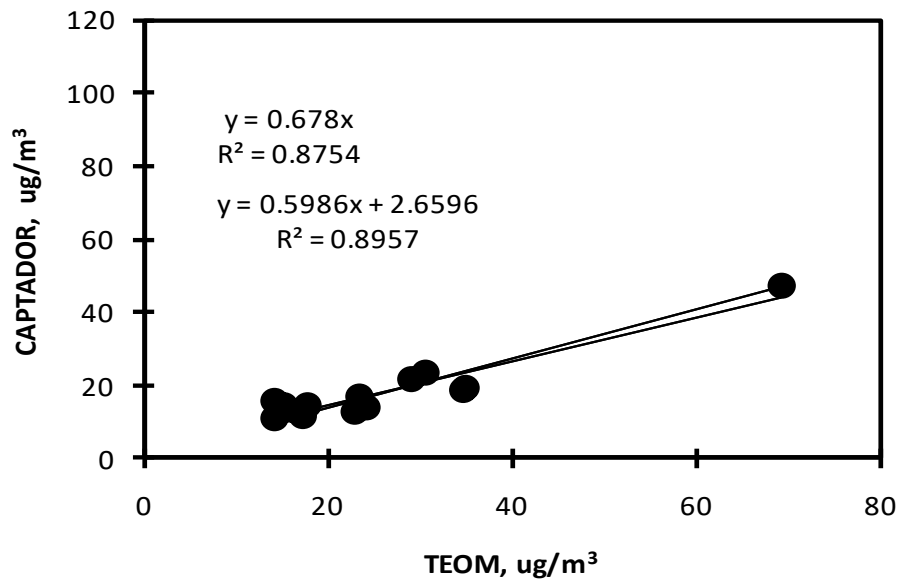
PM monitor



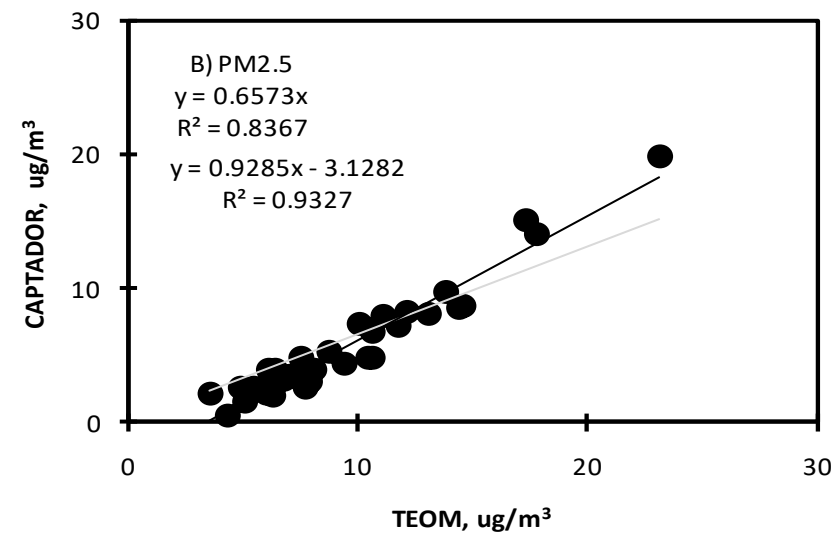
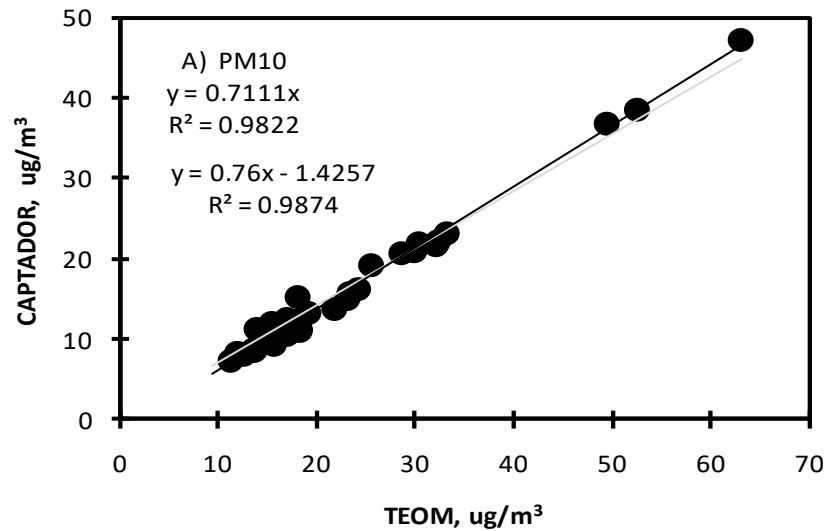
airflow accuracy
calibration of the sensor
leaks
cleaning



UNIDAD MOVIL



GLADIOLOS



TOME CANO

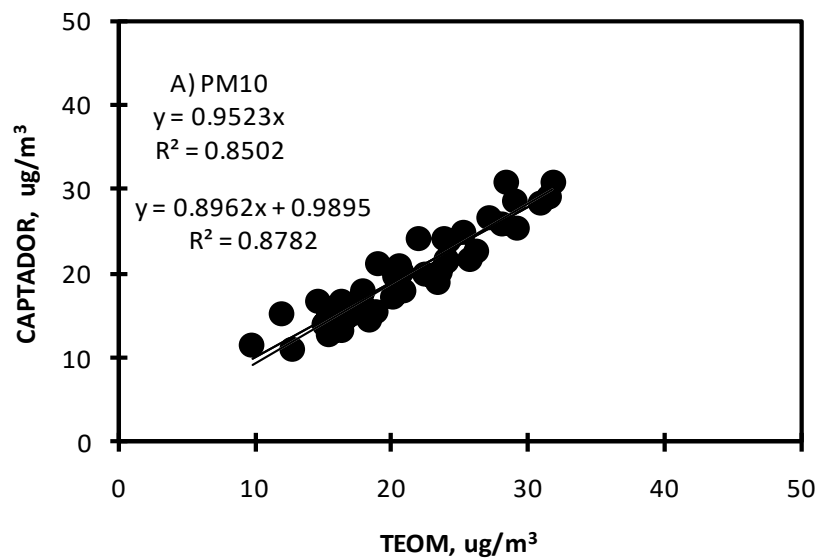


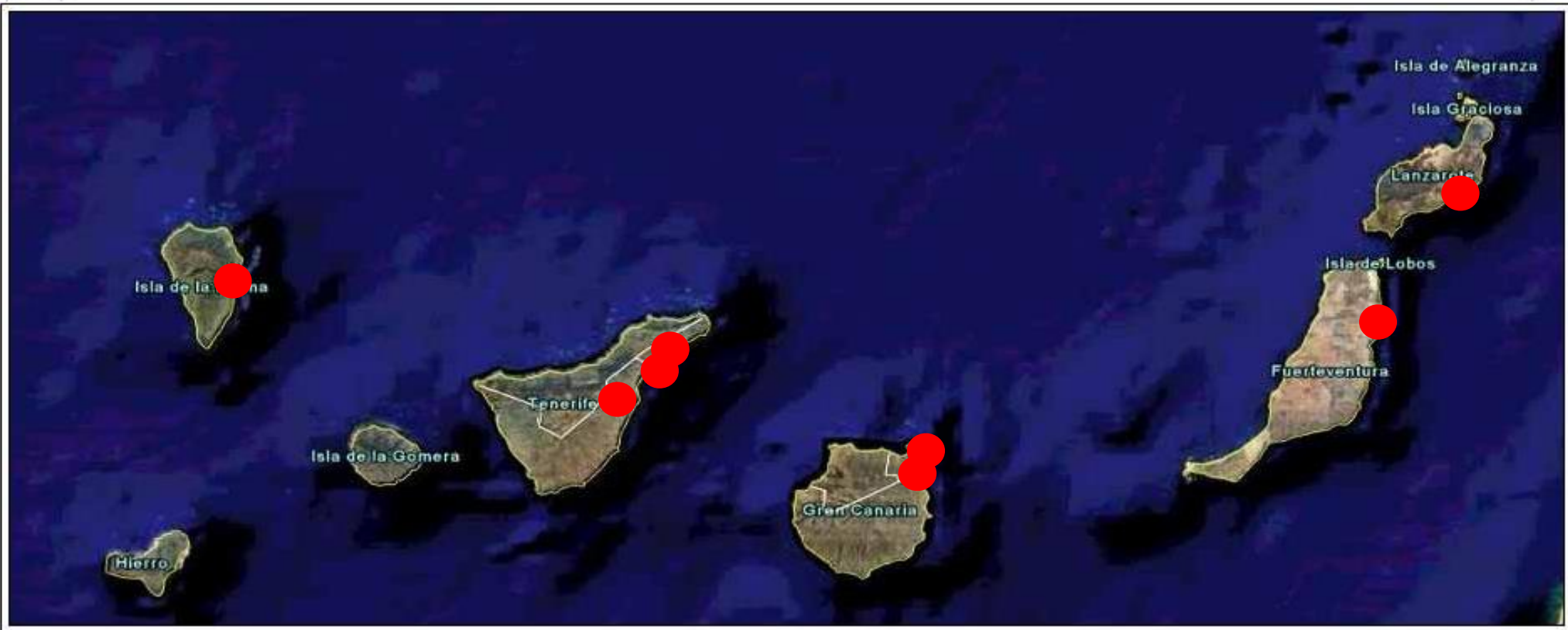
Tabla 1. Recopilación de ecuaciones obtenidas en intercomparaciones de analizadores de PM10.

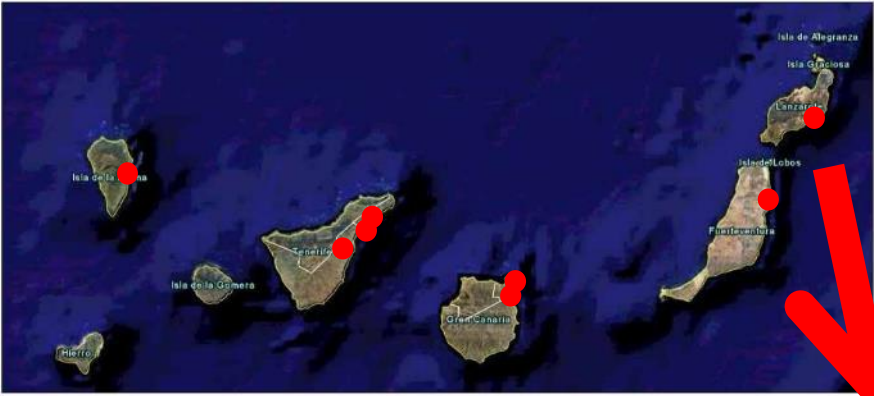
Estación	Fecha	Periodo	T, °C	P hPa	Y=a·x	R ²	¿VALIDA?	Y=a·x+b	R ²	¿VALIDA?	N
LA HIDALGA	21/02/2009 – 24/03/2009	INVIERNO	20.2	972	y=0.798x	0.995	SI	y=0.820x + (-1.530)	0.997	SI	28
LOS GLADIOLOS	27/04/2009 – 09/06/2009	PRIMAVERA	24.4	993	y=0.711x	0.982	SI	y=0.760x +(-1.425)	0.987	SI	34
TOME CANO	04/08/2009-17/09/2009	VERANO	28.7	995	y=0.952x	0.850	SI	y=0.896x +(0.989)	0.878	SI	44
MERCADO CENTRAL	17/11/2009-23/01/2010	INVIERNO	25.1	1015	y=1.275x	0.961	SI	y=1.191x +(2.928)	0.969	SI	49
MERCADO CENTRAL	09/01/2001-28/12/2001	ANUAL	24.8		y=1.285x	0.872	SI	y=1.142x +(7.151)	0.893	SI	88
PARQUE REHOYAS	05/03/2010-21/04/2010	INVIERNO	22.5	1003.8	y=1.032x	0.875	SI	y=1.062x +(-0.561)	0.876	SI	37
LOS GLADIOLOS	24/05/2010-07/06/2010	PRIMAVERA	25.8	1004.3	y=0.778x	0.931	SI	y=0.896x +(-3.8461)	0.951	SI	39
TOME CANO	14/04/2010-29/05/2010	PRIMAVERA	22.2	1007.6	y=0.773x	0.871	SI	y=0.747x +(0.615)	0.872	SI	47
LA HIDALGA	11/06/2010-29/07/2010	VERANO	23.8	985.1	y=0.702x	0.757	NO problemas mantenimien to	y=0.612x +(2.893)	0.776	NO problemas mantenimien to	39
MERCADO CENTRAL	23/06/2010-01/08/2010	VERANO	26.7	1014.7	y=1.172x	0.901	SI	y=1.240x +(-1.694)	0.911	SI	35
PARQUE REHOYAS	20/09/2010-17/10/2010	VERANO	27.0	1000.7	y= 1.017x	0.839	SI	y=1.125X +(-3.067)	0.849	SI	61
CIUDAD DEP. ARRECIFE	26/08/2010-08/10/2010	VERANO	25.2	1010.9	y=1.085x	0.922	SI	y=1.042X +(0.832)	0.923	SI	34

Tabla 2. Recopilación de ecuaciones obtenidas en intercomparaciones de analizadores de PM2.5. N: número de muestras válidas usadas.

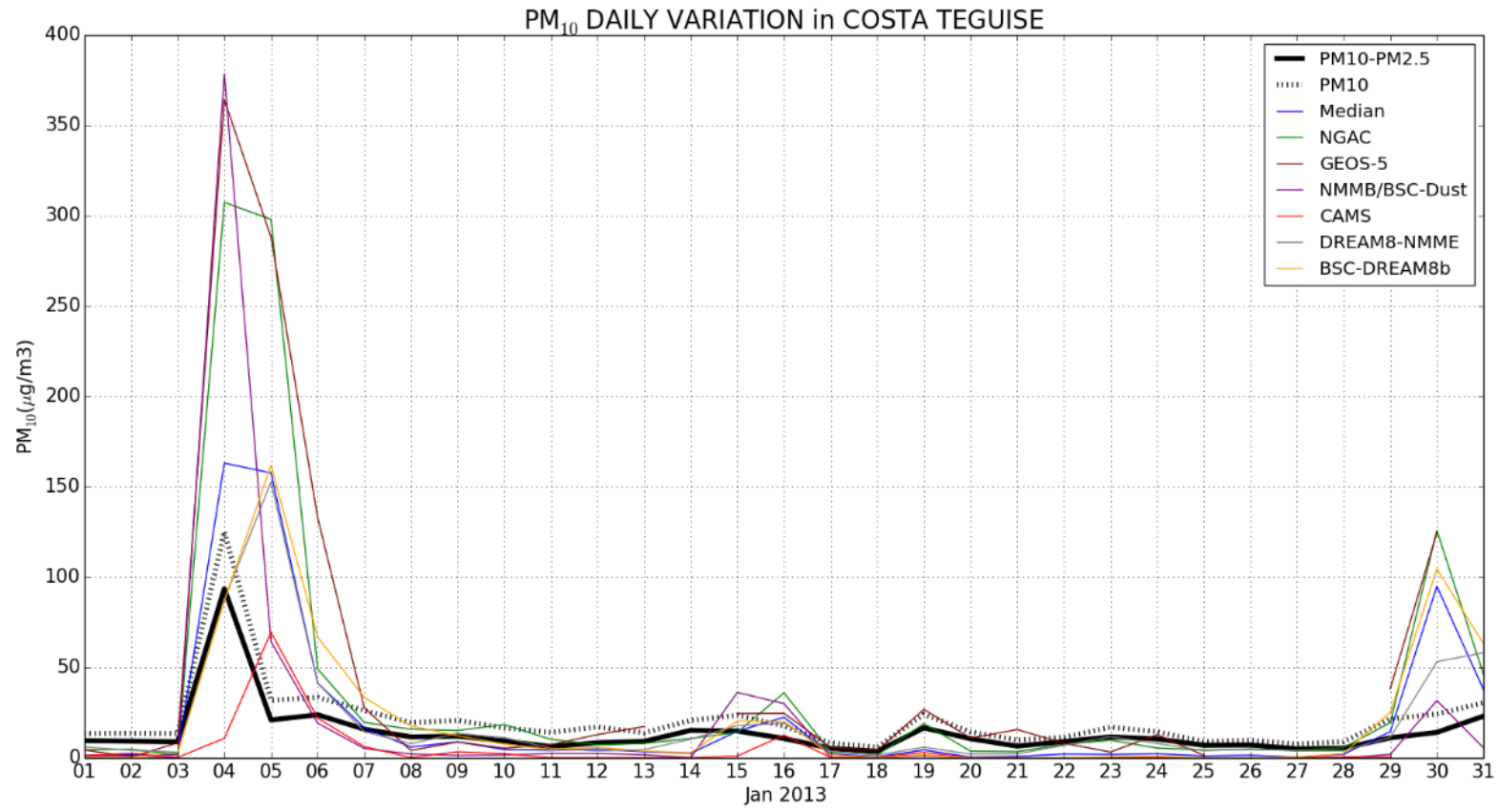
Estación	Fecha	Periodo	T, °C	P hPa	Y=a·x	R ²	¿VALIDO?	Y=a·x+(b)	R ²	¿VALIDO?	N
LA HIDALGA	21/02/2009 – 24/03/2009	INVIERNO	20.2	972	y=0.650x	0.9554	SI	y=0.7033x + (-1.4673)	0.9637	SI	28
LOS GLADIOLOS	27/04/2009 – 09/06/2009	PRIMAVERA	24.4	993	y=0.657x	0.8367	SI	y=0.9285x + (-3.1282)	0.9285	SI	33
MERCADO CENTRAL	17/11/2009-23/01/2010	INVIERNO	25.1	1015	y=0.865x	0.8707	SI	y= 0.7552 + (1.519)	0.8939	SI	45
PARQUE REHOYAS	05/03/2010-21/04/2010	INVIERNO	22.5	1003.8	y=0.768x	0.582	NO, Conc < 10µg/m ³	y=0.908x + (-1.0521)	0.597	NO, Conc < 10µg/m ³	37
LOS GLADIOLOS	24/05/2010-07/06/2010	VERANO	25.8	1004.3	y=0.684x	0.686	NO, Conc < 10µg/m ³	y=0.941x + (-2.462)	0.745	NO, Conc < 10µg/m ³	39
LA HIDALGA	11/06/2010-29/07/2010	VERANO	23.8	985.1	y=0.474x	0.680	NO evalua, Conc < 10µg/m ³	y=0.559x + (-1.254)	0.699	NO evalua, Conc < 10µg/m ³	39
MERCADO CENTRAL	23/06/2010-01/08/2010	VERANO	26.7	1014.7	y= 0.825	0.858	SI	y=0.7494 x + 0.912	0.868	SI	35
PARQUE REHOYAS	20/09/2010-17/10/2010	VERANO	27.0	1000.7	y= 0.797x	0.489	NO evalua, Conc < 10µg/m ³	y=1.192X + (-3.243)	0.553	NO evalua, Conc < 10µg/m ³	61
CIUDAD DEP. ARRECIFE	26/08/2010-08/10/2010	VERANO	25.2	1010.9	y=0.650x	0.627	NO evalua, Conc < 10µg/m ³	y=0.558X + (0.564)	0.635	NO evalua, Conc < 10µg/m ³	34

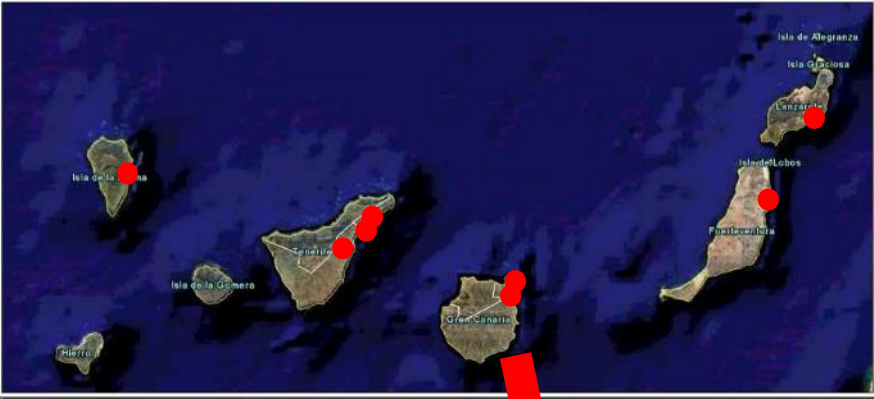
Standardized PM_{10} y $PM_{2.5}$ levels in the network



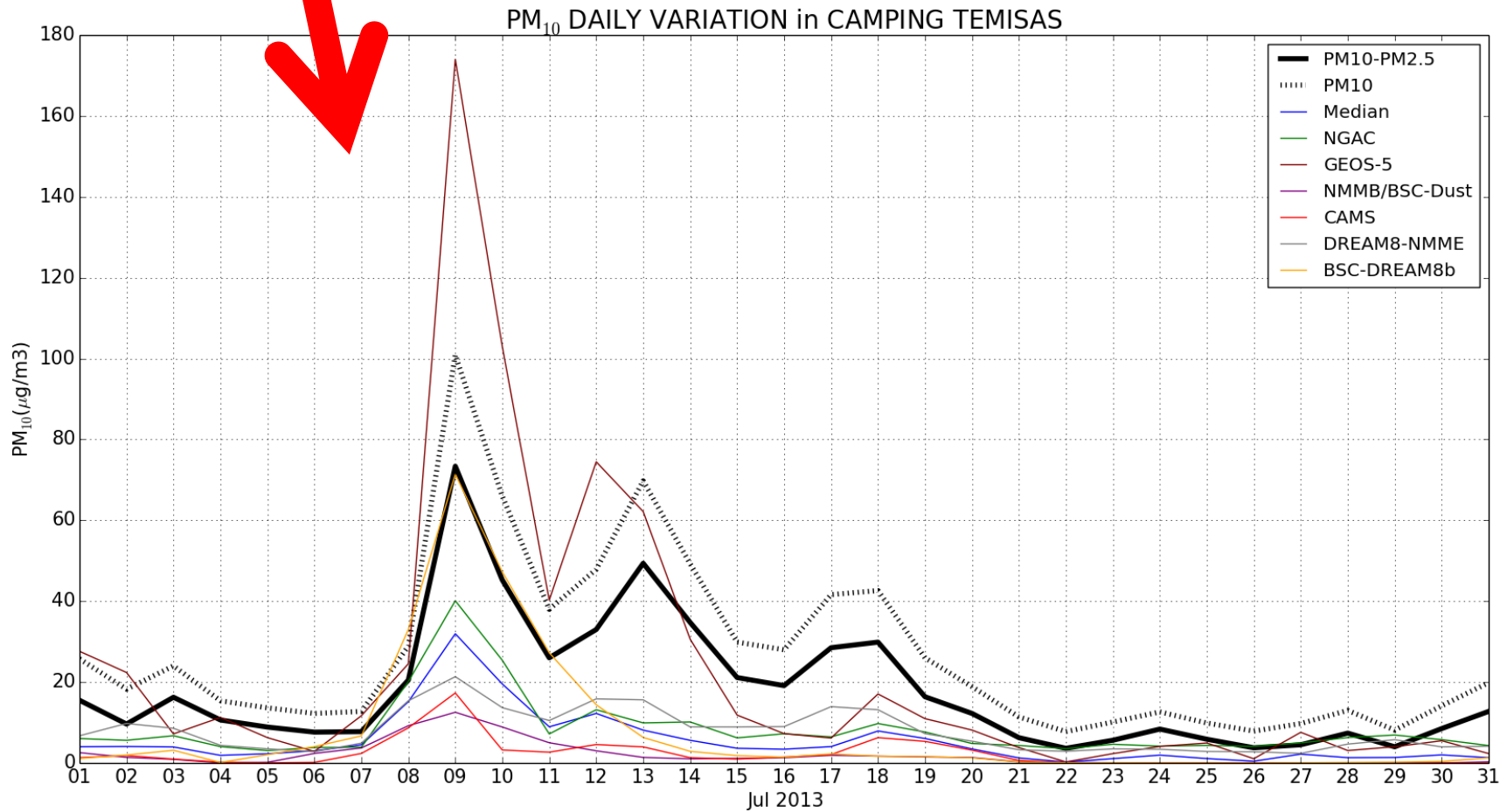


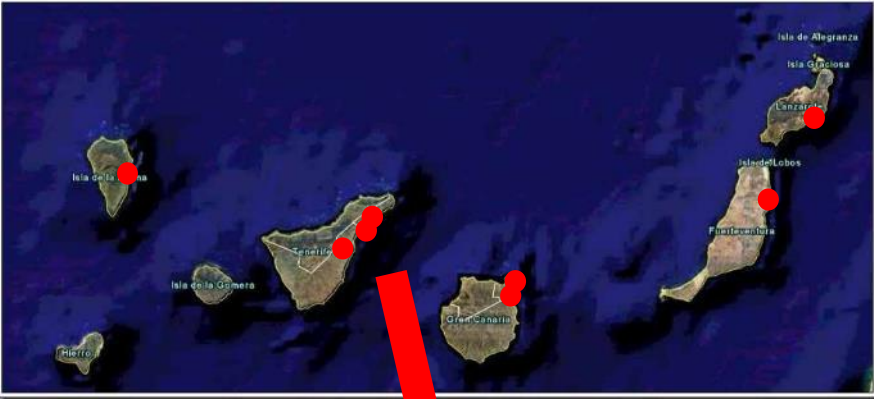
Model validation with standardized PM_{10} y $PM_{2.5}$ data in the network



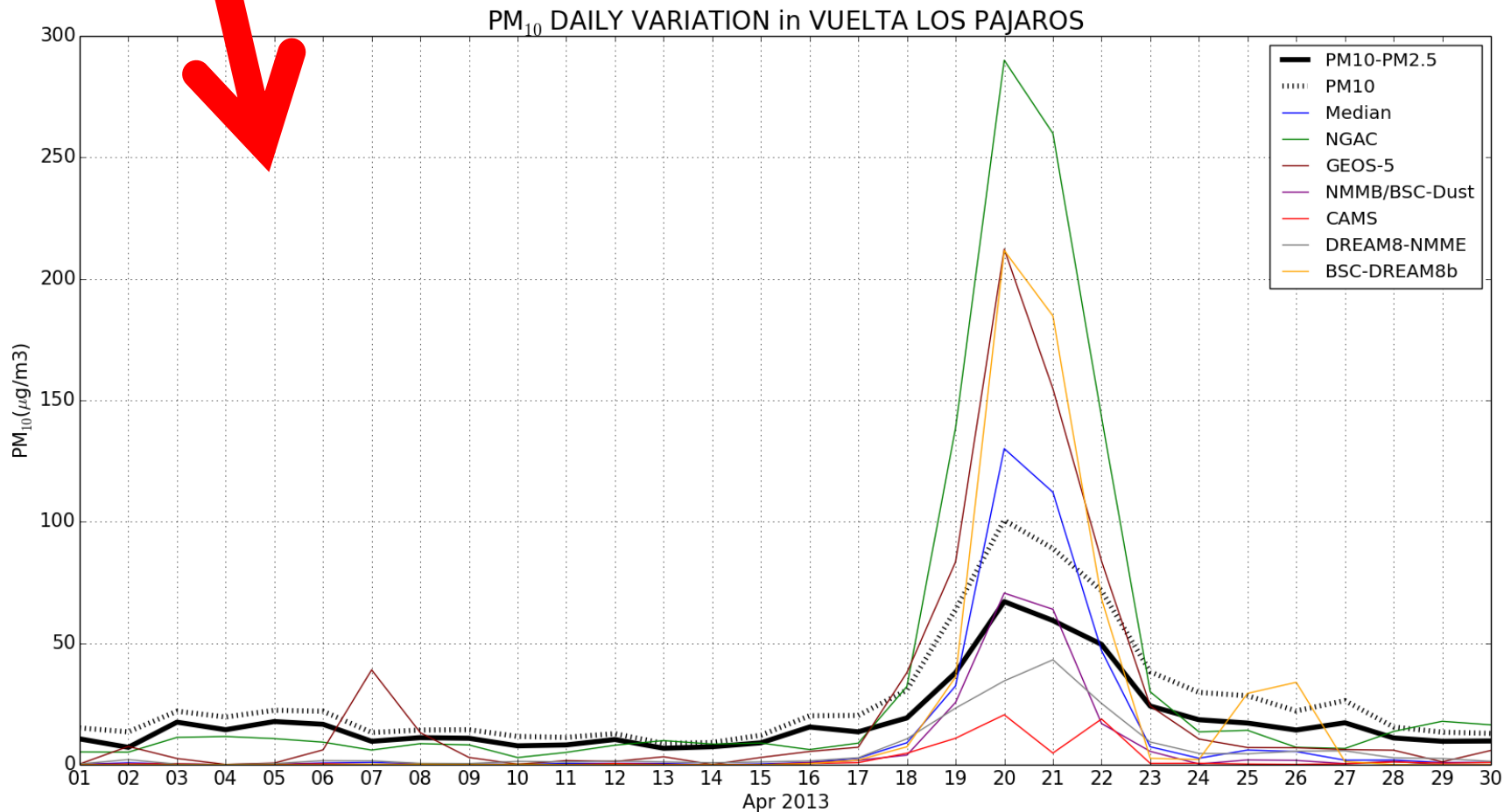


Model validation with standardized PM_{10} y $PM_{2.5}$ data in the network





Model validation with standardized PM_{10} y $PM_{2.5}$ data in the network





dust air quality

1. PM_{10} and $PM_{2.5}$ levels

-method-01: reference - manual gravimetry

-method-02: automatic

Manual gravimetry

automatic

advantage: reference method

high time resolution, 1h

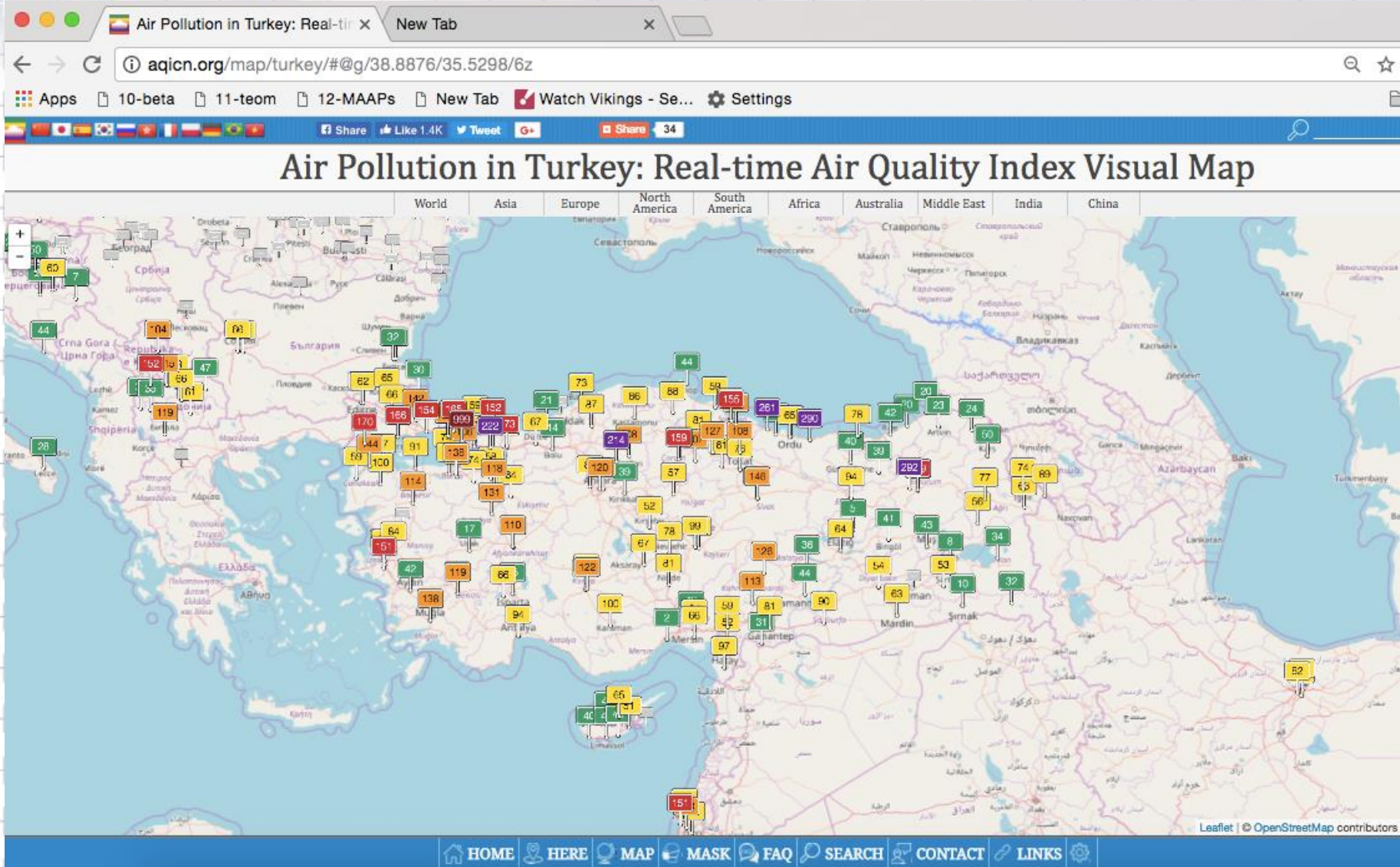
disadvantage: poor time resolution, 24-h average
manual work
takes 3 days to know PM_{10}

Needs validation

we recommend to use the two methods:

-automatic, continuously

-gravimetric: intercomparisons - 1 month summer, 1 month winter



dust, aerosols and pollutants

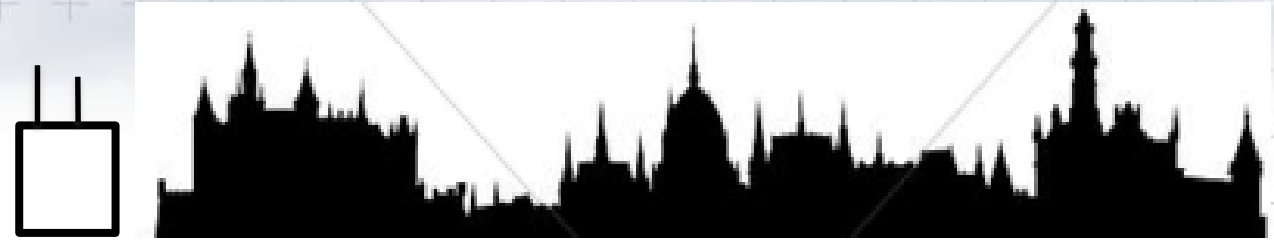
in-situ observations

PM₁₀ and PM_{2.5} levels

PM₁₀ and PM_{2.5} composition

complementary observations

observation network



dust air quality

1. PM_{10} and $PM_{2.5}$ levels

-method-01: reference - manual gravimetry

-method-02: automatic

Manual gravimetry

automatic

advantage: reference method

high time resolution, 1h

CHEMICAL ANALYSIS

disadvantage: poor time resolution, 24-h average
 manual work
 takes 3 days to know PM_{10}

Needs validation

we recommend to use the two methods:

-automatic, continuously

-gravimetric: intercomprisons, 1 month summer, 1 month winter

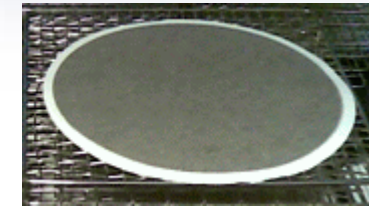
bulk chemical composition

PM samples: $\left\{ \begin{array}{l} \text{fine + coarse (TSP, PM}_{10}\text{)} \\ \text{fine (PM}_{2.5}\text{, PM}_1\text{)} \end{array} \right.$

Saharan dust



Urban particles



PM ($\mu\text{g}/\text{m}^3$) = **dust** + **trace elements** + **ions** (SO_4^- , NO_3^- , NH_4^+ , Na^+ , Cl^-) + OC + EC

Elemental Composition:

Major elements (Al, Si, Ca, K, Na, Mg) + trace elements
 (P, Li, Be, Sc, Ti, V, Cr, Mn, Co, Ni, Cu, Zn, Ga, Ge, As, Se, Rb, Sr, Y, Zr, Nb, Mo, Cd, Sn, Sb, Cs, Ba, La, Ce, Pr, Nd, Sm, Eu, Gd, Tb, Dy, Ho, Er, Tm, Yb, Lu, Hf, Ta, W, Tl, Pb, Bi, Th, U)

Ions: SO_4^- , NO_3^- , NH_4^+ , Na^+ , Cl^-

Ion Chromatography, ICP-AES, ICP-MS,
 selective electrodes and colorimetry

Thermal/optical reflectance (TOR) and/or thermal/optical transmission (TOT)

Inductively coupled plasma Atomic Emission Spectroscopy
 ICP-AES

Destructive techniques

destructive techniques

Inductively coupled plasma Mass spectroscopy
 ICP-MS

Destructive techniques

XRF, PIXE, INAA : none destructive techniques

bulk chemical composition

PM samples: { fine + coarse (TSP, PM₁₀)
fine (PM_{2.5}, PM₁)

dust



Elemental Composition:

Major elements (Al, Si, Ca, K, Na, Mg) + trace elements (P, Li, Be, Sc, Ti, V, Cr, Mn, Co, Ni, Cu, Zn, Ga, Ge, As, Se, Rb, Sr, Y, Zr, Nb, Mo, Cd, Sn, Sb, Cs, Ba, La, Ce, Pr, Nd, Sm, Eu, Gd, Tb, Dy, Ho, Er, Tm, Yb, Lu, Hf, Ta, W, Tl, Pb, Bi, Th, U)

Bulk dust estimations

33% Si

8% Al

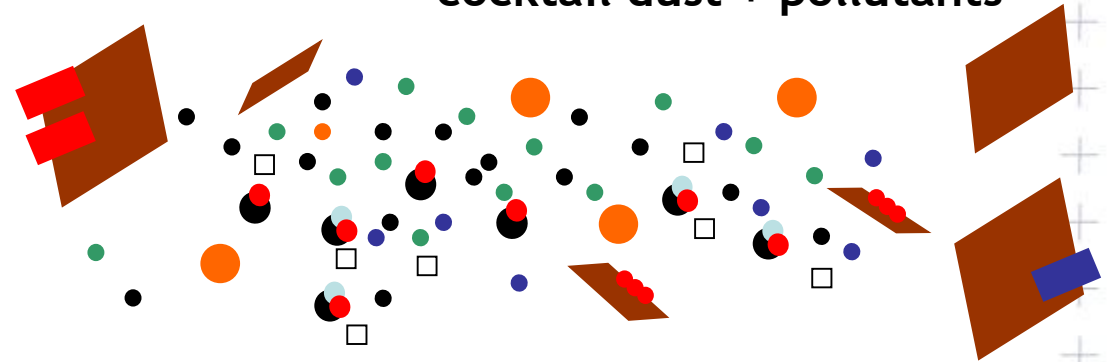
4% Fe

.....

Al₂O₃ + SiO₂ + CaCO₃ + ...



people live in cities and breath a cocktail dust + pollutants

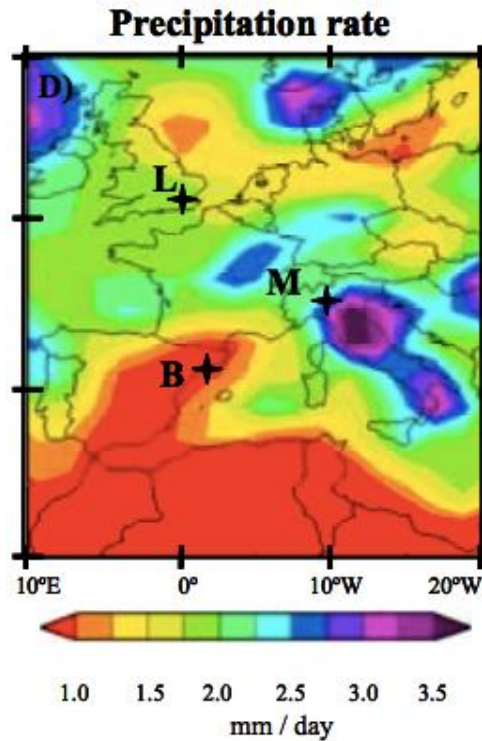
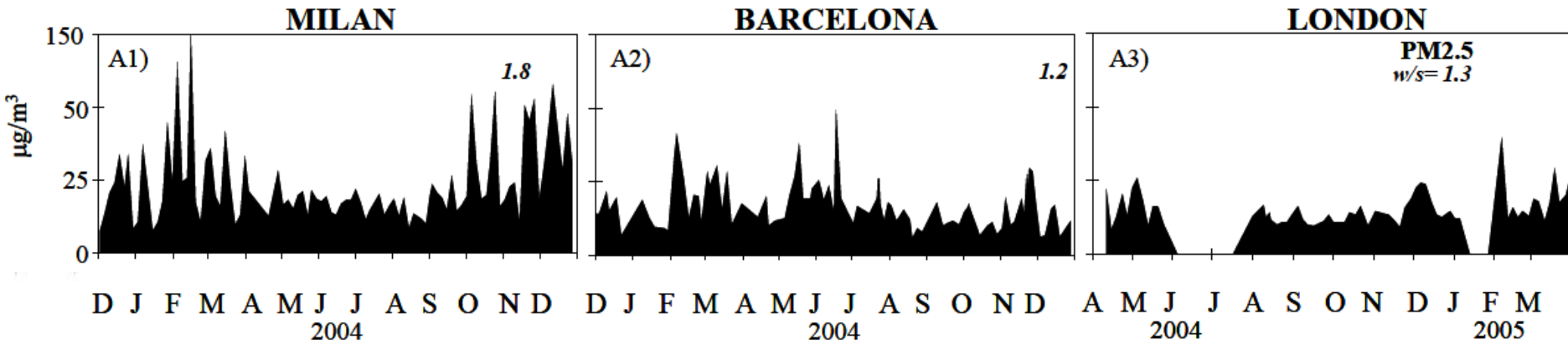


PM_{10} : dust + sea salt + (sulphate + nitrate + organic matter + black carbon + metals) ..

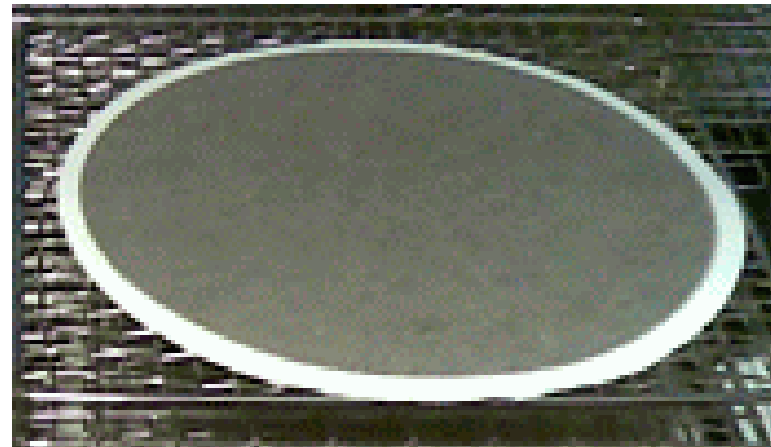
$PM_{2.5}$: dust + sea salt + (sulphate + nitrate + organic matter + black carbon + metals) ..

PM_{10} : dust + sea salt + vehicle exhaust + oil refining + power plants + ships +...

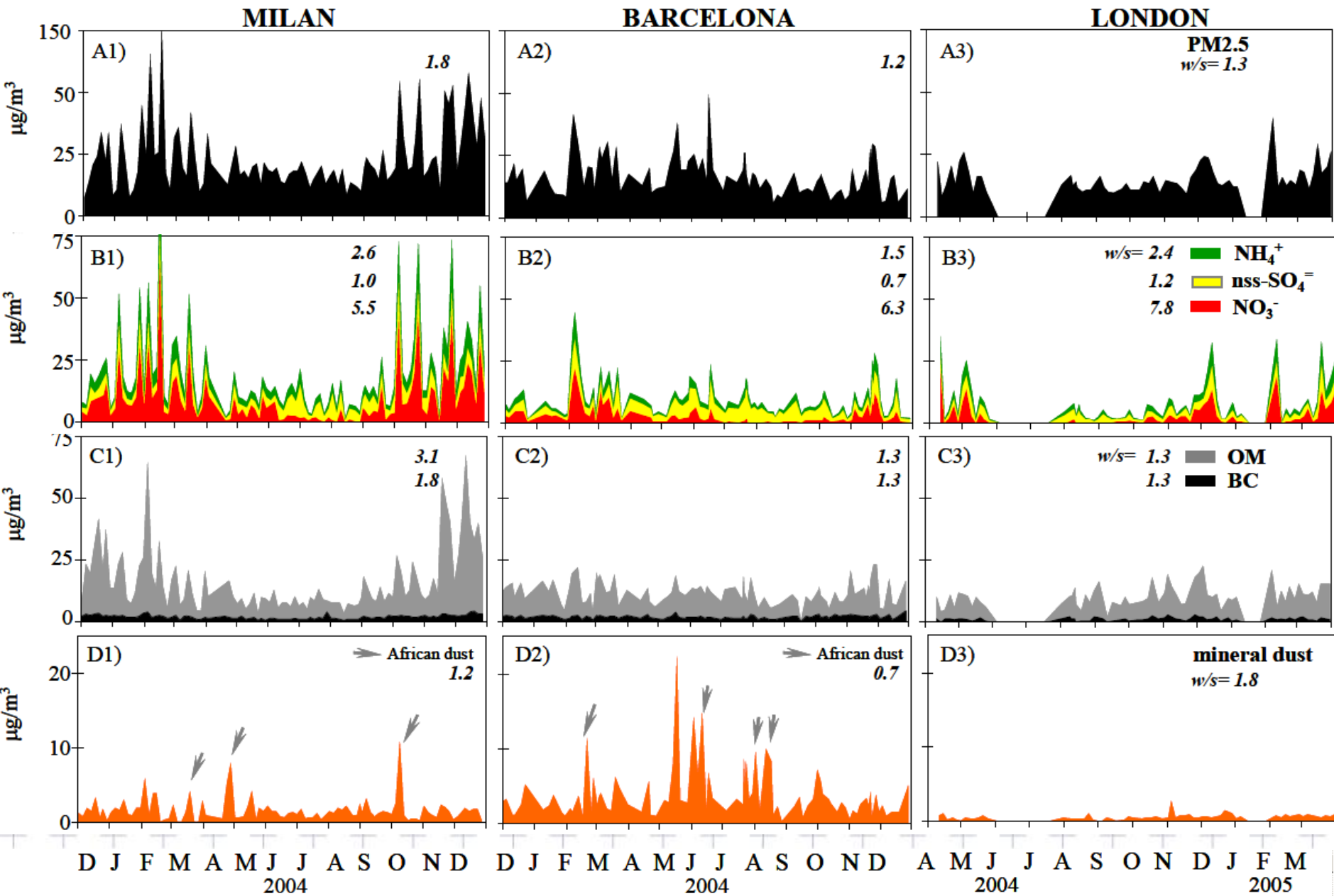
$PM_{2.5}$: dust + sea salt + vehicle exhaust + oil refining + power plants + ships +...



Urban particles

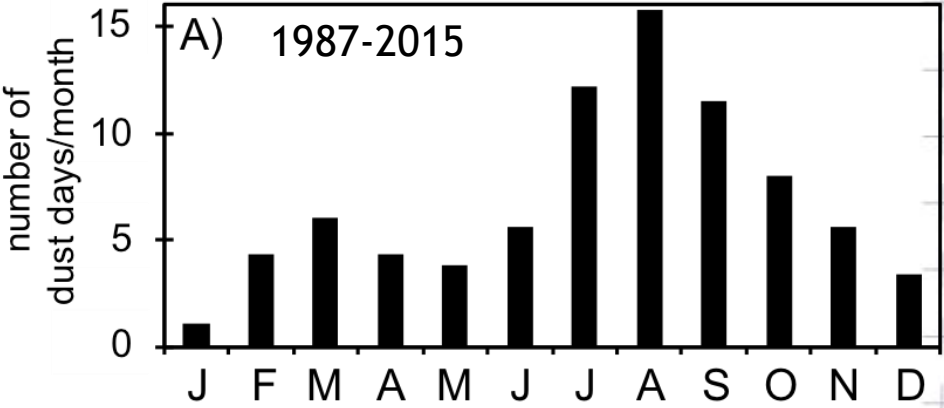


PM in urban areas

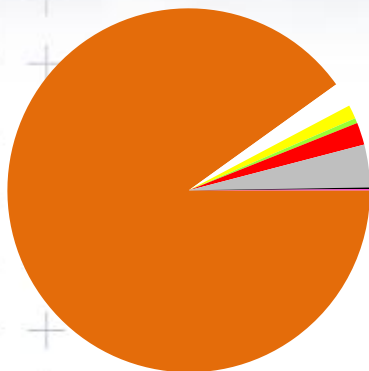


PM in remotes sites

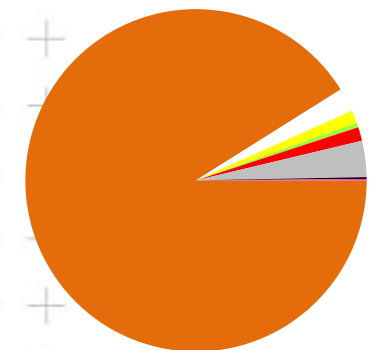
Summer Izaña is within the SAL



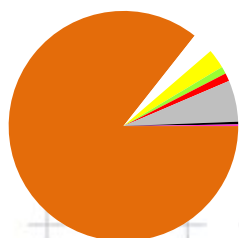
PM_x composition in the SAL



PM _T	47.3	μg/m ³
91%	42.6	dust (Al, Fe, Ca, Ti..)
2.2%	1.0	none ammonium-sulfate
1.2%	0.5	ammonium-sulfate
0.4%	0.2	ammonium
1.9%	0.9	nitrate
3.8%	1.8	organic matter
0.2%	0.07	elemental carbon



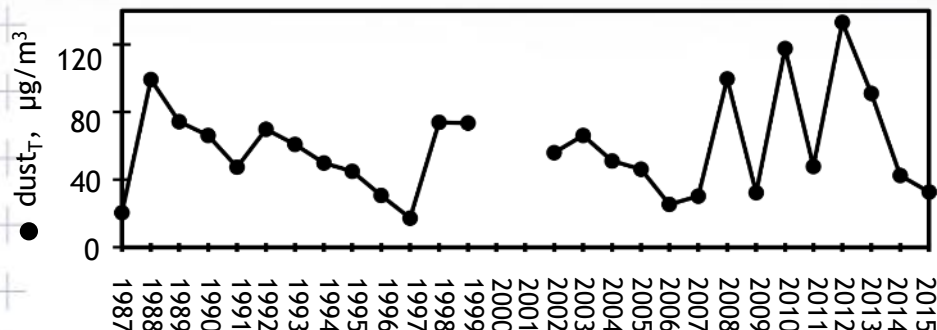
PM ₁₀	42.0	μg/m ³
91%	38.3	dust
2.2%	0.9	none ammonium-sulfate
1.2%	0.5	ammonium-sulfate
0.4%	0.2	ammonium
1.3%	0.6	nitrate
3.4%	1.4	organic matter
0.2%	0.07	elemental carbon



PM _{2.5}	18.5	μg/m ³
85%	15.8	dust
3.0%	0.6	none ammonium-sulfate
2.7%	0.5	ammonium-sulfate
1.0%	0.2	ammonium
1.1%	0.2	nitrate
5.8%	1.1	organic matter
0.4%	0.07	elemental carbon



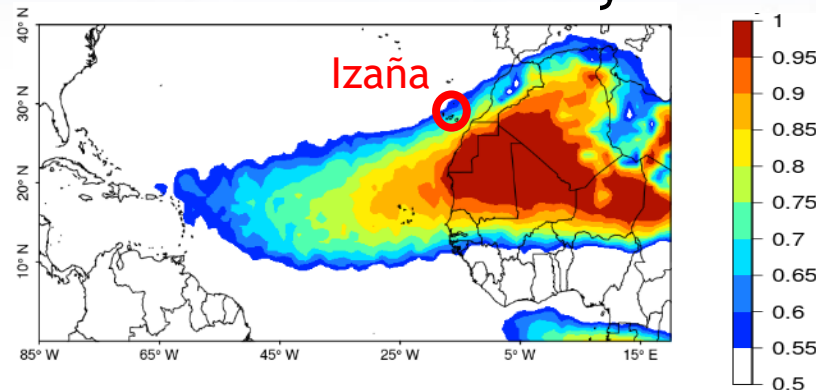
summer dust at Izaña: 1987 - 2015



Max: 133 µg/m³ 2012

Min: 17 µg/m³ 1997

Saharan Air Layer



M DFA: Major Dust Frequency Activity

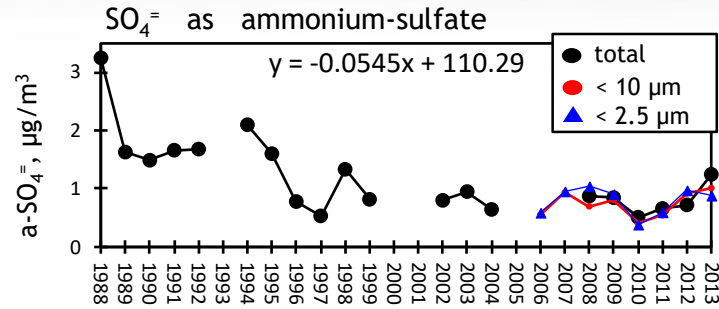
UV Absorbing Aerosol Index = sensitive to iron oxides in dust

$$\text{M DFA} = \frac{\text{number days UV Absorbing Aerosol Index} > 1}{\text{total number of days in the month}}$$

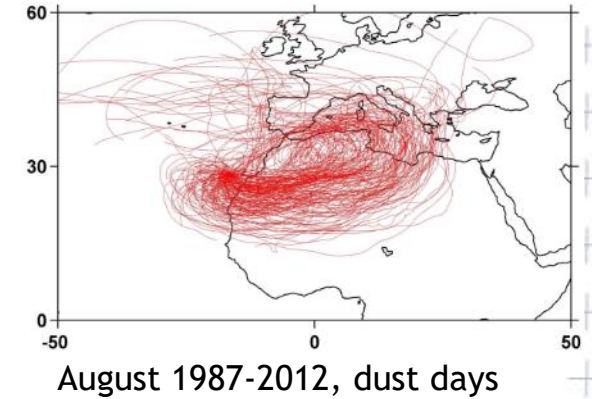
= fraction of summertime AI>1

Satellite (Earth Probe, Nimbus 7, Aura):
 Total Ozone Monitor Spectrometer (1987-2001)
 Ozone Monitor Instrument (2005-2012)

ammonium-sulfate in the Saharan Air Layer



(1) air laden in Saharan dust has previously passed over the Mediterranean and Europe



dust, aerosols and pollutants

in-situ observations

PM₁₀ and PM_{2.5} levels

PM₁₀ and PM_{2.5} composition

complementary observations

let's build our observation network !!!

in-situ observations



dust air quality



in-situ observations

PM₁₀ and PM_{2.5} levels

PM₁₀ and PM_{2.5} composition

complementary observations

in-situ

meteorology:

wind, temperature, relative humidity, pressure

gaseous pollutants (**reference methods**):

NO_x: vehicle exhausts, ships, oil refining, power plants..

SO₂., ships, oil refining, power plants

CO: vehicle exhausts



Examples of reference methods:

NO_x: chemiluminiscense. EN 14211: 2006

SO₂: fluorescense. EN 14212: 2006

CO: NDIR absorption. EN 14626: 2006

O₃: NDIR absorption. EN 14625: 2006

in-situ observations



dust air quality



in-situ observations

PM_{10} and $PM_{2.5}$ levels

PM_{10} and $PM_{2.5}$ composition

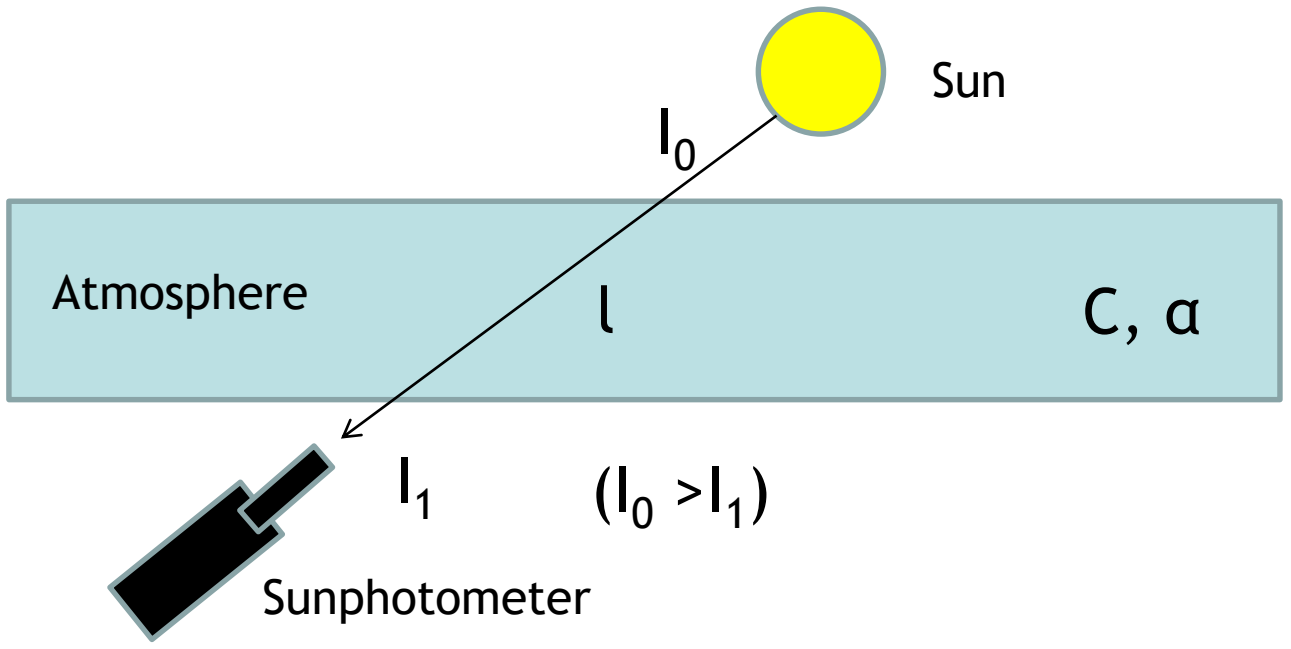
complementary observations
ground based remote sensing

column

vertical distribution

CONCEPTS:

Knowing the sunlight's energy at the top of the atmosphere, the thickness of the atmosphere, and the amount of sunlight transmitted to the earth's surface may allow us to **determine the amount of extinction**, and thus, the amount of **aerosols (dust)**.



Beer's Law

$$I = I_0 \cdot e^{-\sigma_{\text{ext}} \cdot L}$$

- Transmissivity (T)
- Extinction coefficient (σ_{ext}): ϵC
- path length (L)
- molar absorptivity of the absorber (ϵ)
- concentration of absorbing species in the material (C)

CONCEPTS:

Aerosol Extinction: A measure of attenuation of the light passing through the atmosphere due to scattering and absorption by aerosol particles.

Extinction coefficient (σ_{ext}) is the fractional depletion of radiance per unit path length (also called attenuation). It has units of km^{-1} .

Aerosol Mass Load: The columnar aerosol mass concentration ($\mu\text{g}/\text{cm}^2$) is the total aerosol mass in a vertical column of atmosphere.

CONCEPTS:

Aerosol Optical Depth (or Thickness)

"Aerosol Optical Depth" (AOD) is the degree to which aerosols prevent the transmission of light. The aerosol optical depth or optical thickness (τ) is defined as the integrated extinction coefficient over a vertical column of unit cross section.

$$AOD = \int_{z=0}^{z=toa} \sigma_{ext}(z) dz$$

Angstrom Exponent (α)

An exponent that expresses the spectral dependence of Aerosol Optical Depth (τ) with the wavelength of incident light (λ). The spectral dependence of aerosol optical thickness can be approximated (depending on size distribution) by:

$$AOD = \beta \lambda^{-\alpha}$$

$\alpha \gg 0.9$ FINE particles

$\alpha \ll 0.7$ COARSE particles

where α is the Angstrom exponent (β = aerosol optical depth at 1 μm)

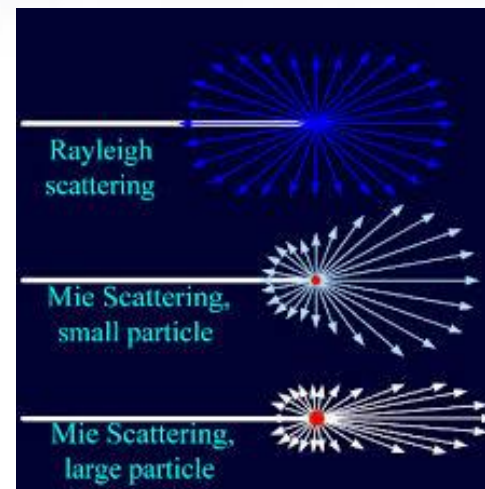
i.e. If AOD $>\sim 0.2$ and $\alpha < 0.7$ then we are observing dust (aprox.)

CONCEPTS:

Aerosol Asymmetry Factor A measure of the preferred scattering direction (forward or backward) for light encountering aerosol particles.

$$g = \frac{1}{2} \int_{-1}^{+1} \cos \Theta P(\cos \Theta) d \cos \Theta$$

$$P(\cos \Theta) = \frac{1 - g^2}{(1 + g^2 - 2g \cos \Theta)^{3/2}}$$



In general, **$g=0$ indicates scattering directions evenly distributed** between forward and backward directions, i.e. isotropic scattering (e.g. scattering from small particles)

$g < 0$ scattering in the backward direction (i.e scattering angle > 90 deg.), often referred to as backscattering, is scattering at 180 deg.

$g > 0$ scattering in the forward direction (i.e scattering angle < 90 deg.), often referred to as forward-scattering, is scattering at 0 deg. **For larger size or Mie particles, g is close to $+1$. Including DUST**

ASSESSMENT OF OBSERVATIONS CONSISTENCY

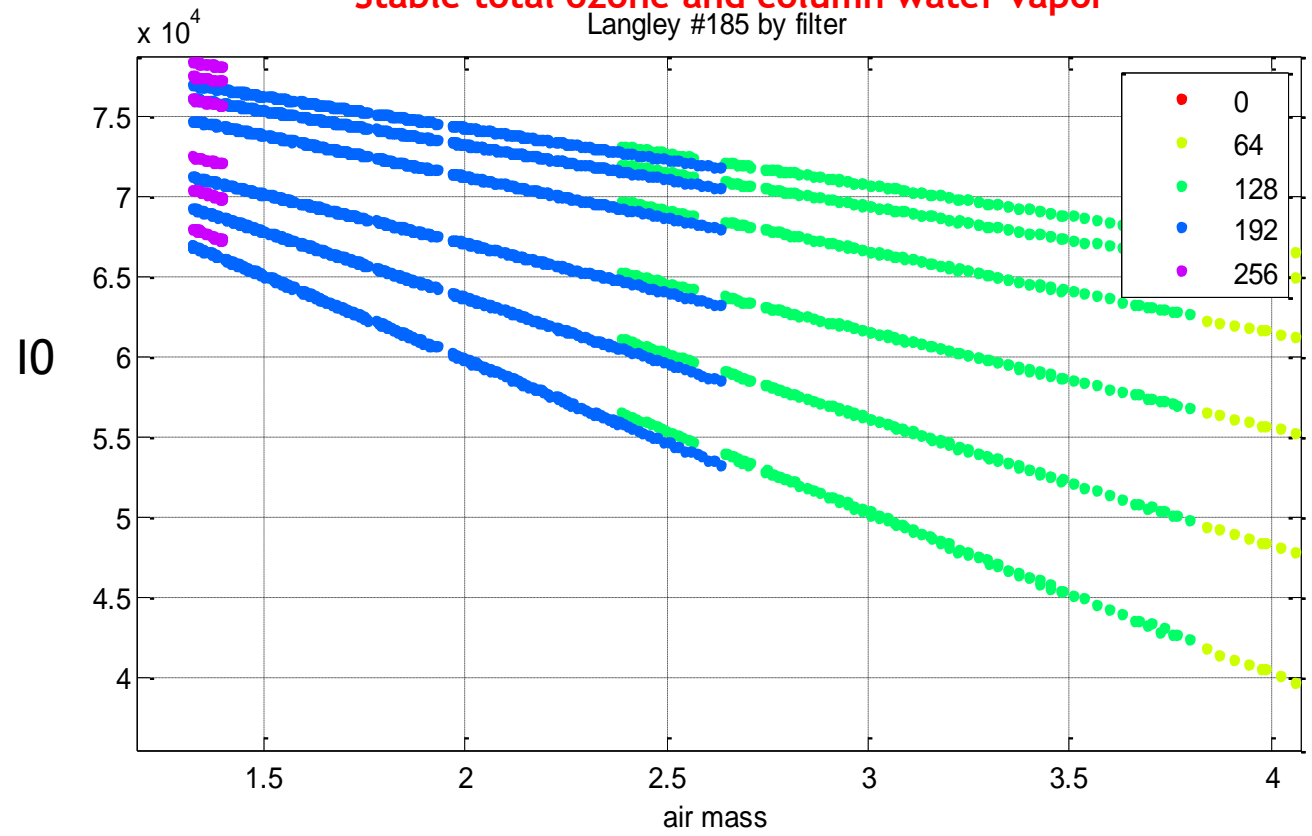
Langley plot calibration (100 determination for each wavelength):

$$I = I_0 \cdot e^{-\sigma_{\text{ext}} \cdot L}$$

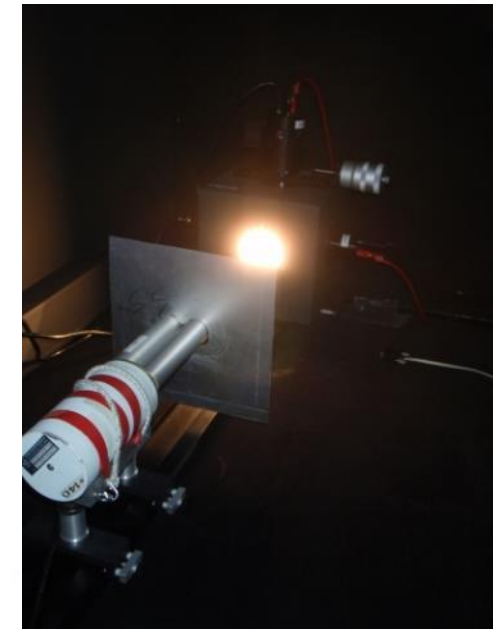
$$\ln I = \ln I_0 - \sigma_{\text{ext}} L$$

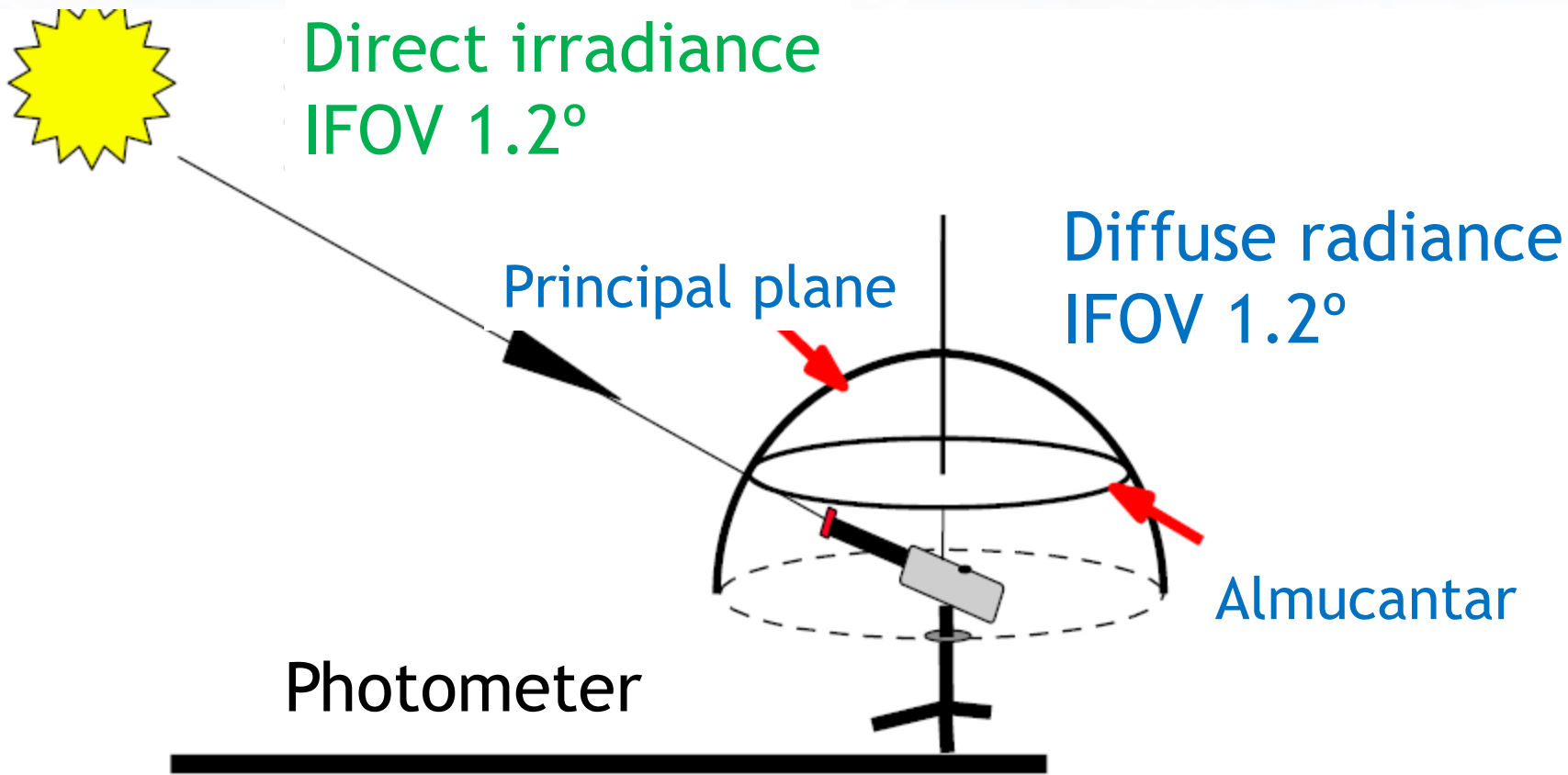
If σ_{ext} is constant during the observation  We can determine I_0

- Pristine conditions (very low and constant aerosol load)
- No clouds
- Stable total ozone and column water vapor



- The Cimel Electronique 318 spectral radiometer is a solar-powered, weather-hardy, robotically-pointed sun and sky spectral sun photometer.
- A sensor head points the sensor head at the sun according to a preprogrammed routine.
- The Cimel controller, batteries, and the optional Vitel satellite transmission equipment are usually deployed in a weatherproof plastic case.





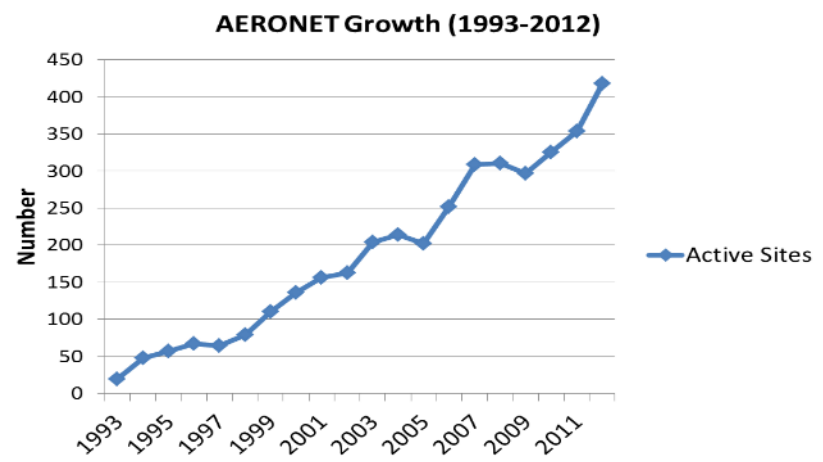
Sun measurements
Sky measurements

AERONET Aerosol Robotic Network-Twenty Years of Observations and Research

The **AERONET program** is a federation of ground-based remote sensing aerosol networks established by NASA and LOA-PHOTONS (CNRS) and has been expanded by collaborators from international agencies, institutes, universities, individual scientists and partners.



- >7000 citations
- >400 sites
- Over 80 countries
- <http://aeronet.gsfc.nasa.gov>



AERONET provides a long-term, continuous public database of aerosol optical, microphysical, and radiative properties for aerosol research and characterization, validation of satellite measurements, and synergism with other databases.

AERONET Data Flows

<http://aeronet.gsfc.nasa.gov>

Flux measurements

Direct - $\lambda=340, 380, 440, 500, 670, 870, 940, 1020$ nm

Diffuse - $\lambda=440, 670, 870, 1020$ nm (alm, pp, pol)

Calibration and processing information

Mauna-Loa and Izaña

CNRS-University of Lille and University of Valladolid

Aerosol optical depth and precipitable water computations

Cloud screening and quality control

Inversion products

Volume size distribution ($0.05 < \text{size} < 15 \mu\text{m}$), refractive index, single scattering albedo ($\lambda=440, 670, 870, 1020$ nm)

Holben et al.
RSE, 1998
Holben et al.
JGR, 2001

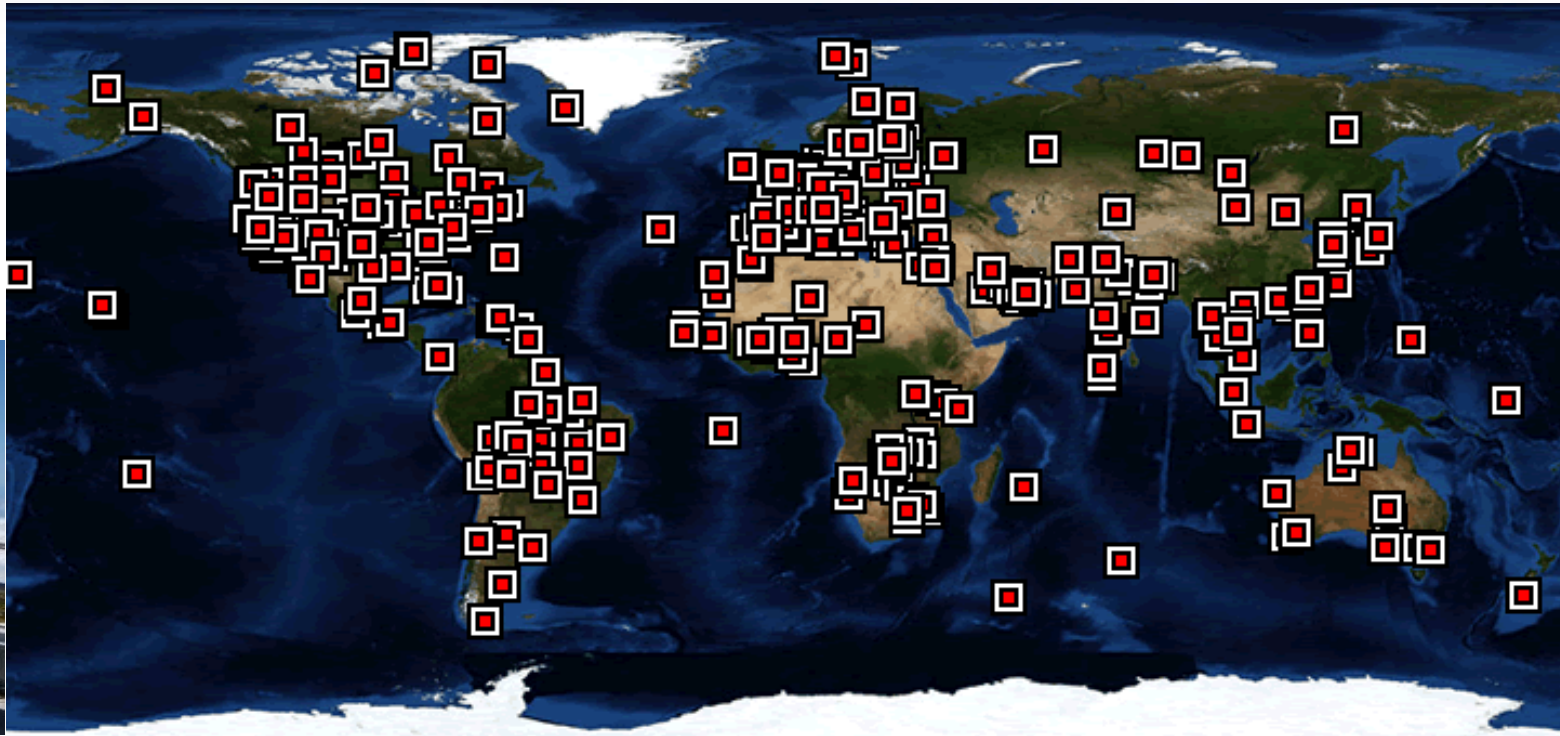
Eck et al.
JGR, 1999

Smirnov et al.
RSE, 2000

Dubovik and King
JGR, 2000
Dubovik et al.
JGR, 2000
GRL, 2002

AERONET (Aerosol RObotic NETwork)-

<http://aeronet.gsfc.nasa.gov>



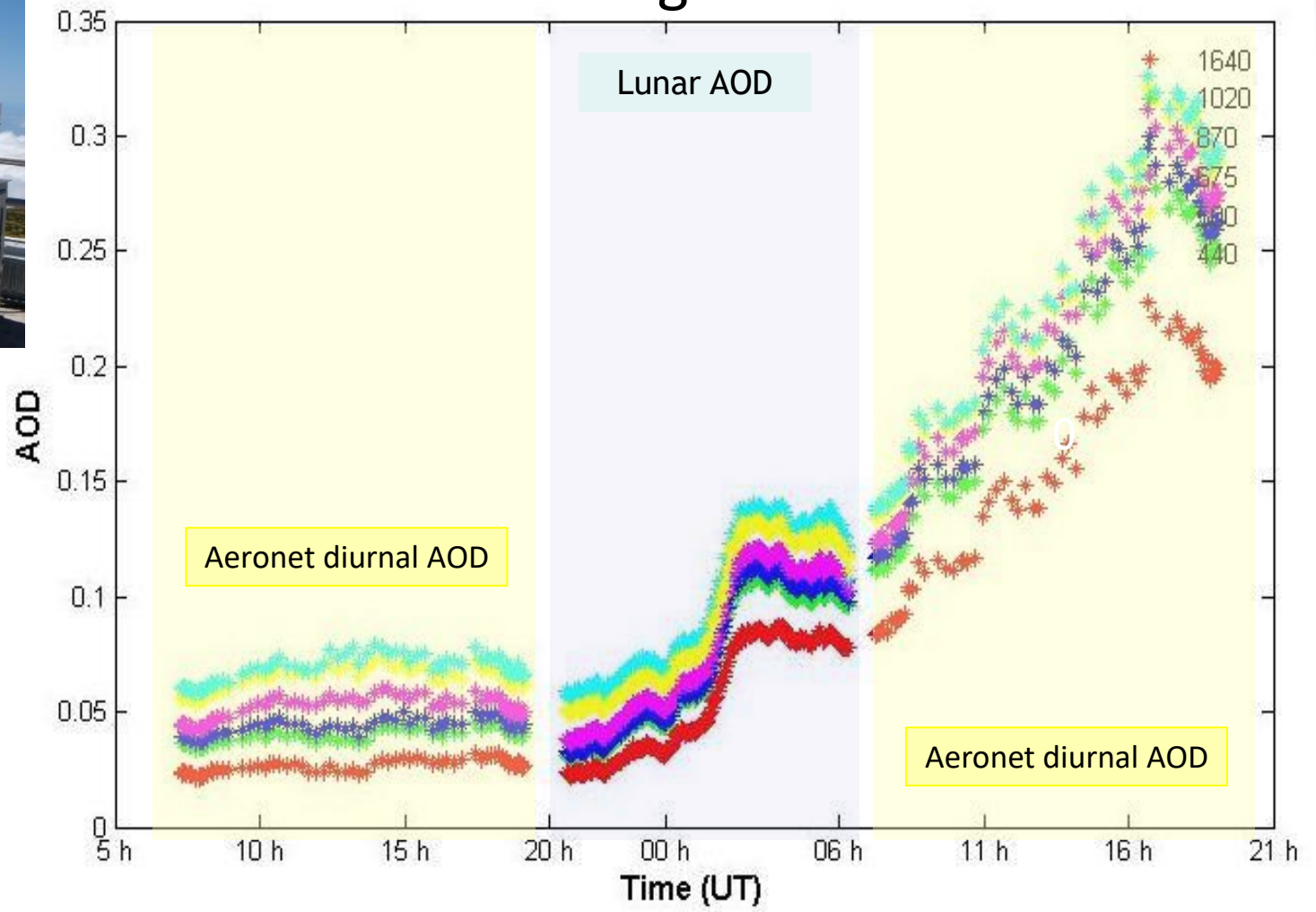
An internationally Federated Network

- Characterization of aerosol optical properties
- Validation of satellite aerosol retrieval
- Near real-time acquisition; long term measurements

AERONET provides:

- global Aerosol Optical Depth of Dust in near real-time
- robust optical properties of Dust: size distribution, ref. Index, etc. (e.g. Asian Dust has stronger and less spectral dependent absorption than Saharan Dust)
- climatological models that reproduce observed optical properties of aerosol (useful for satellite retrievals)

AOD 14 August 2011



August 13

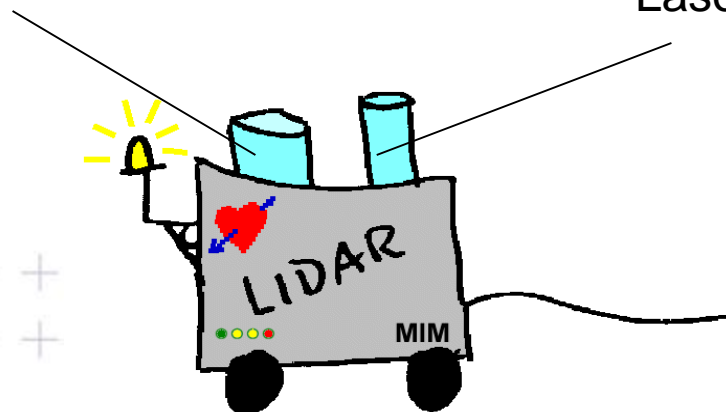
August 14

From total column observations...
to vertical resolved observations

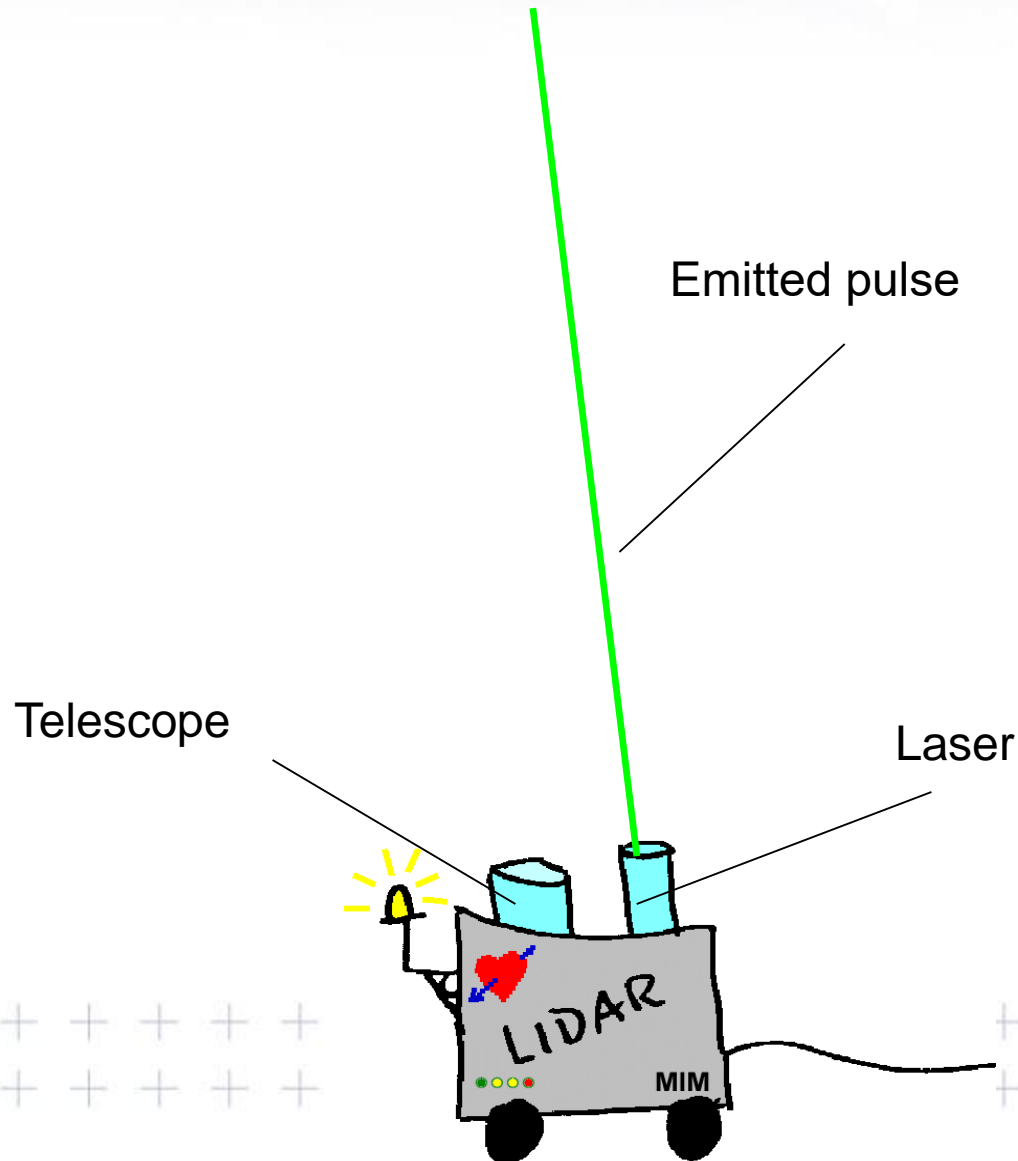
Lidars

Telescope

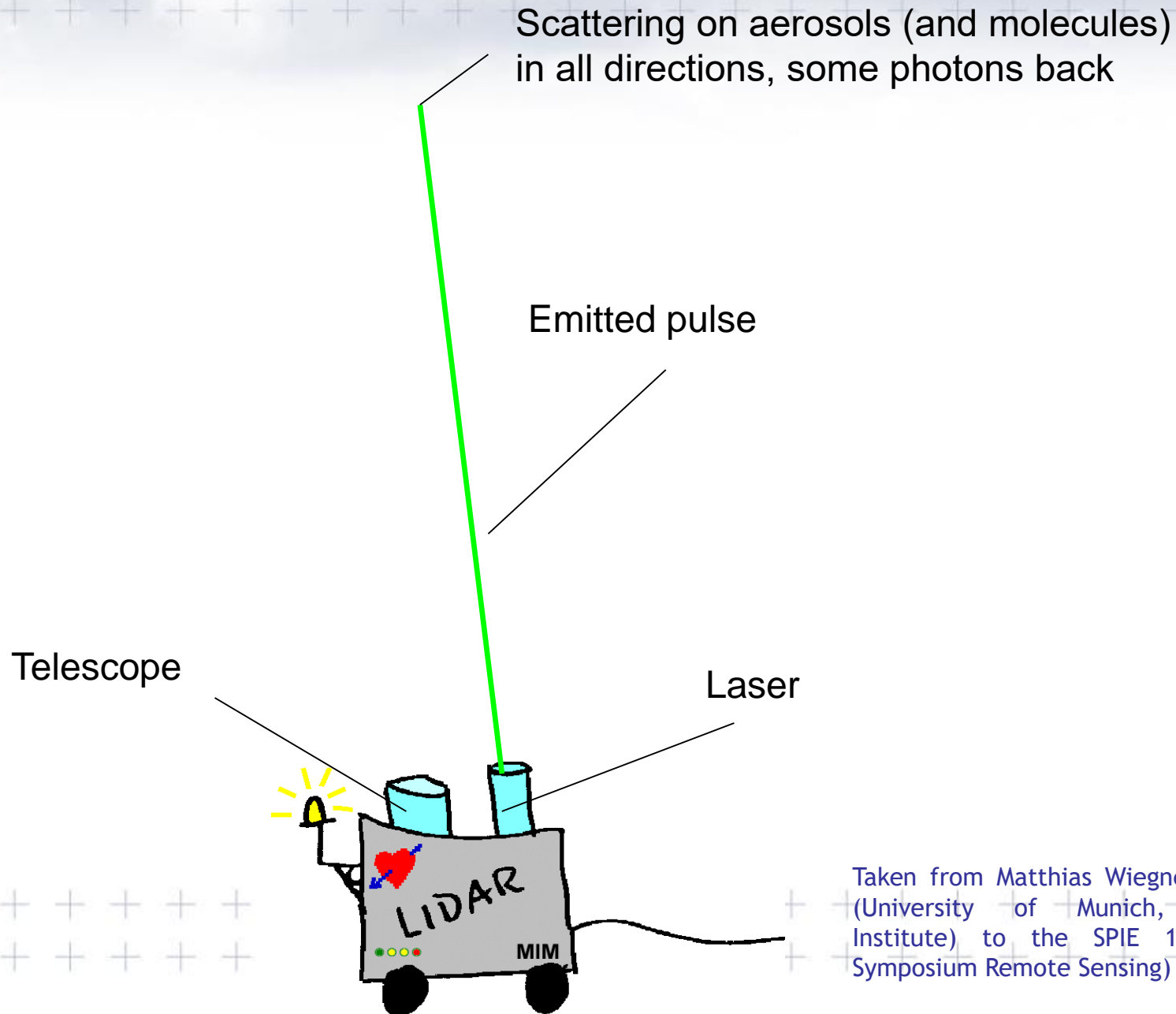
Laser



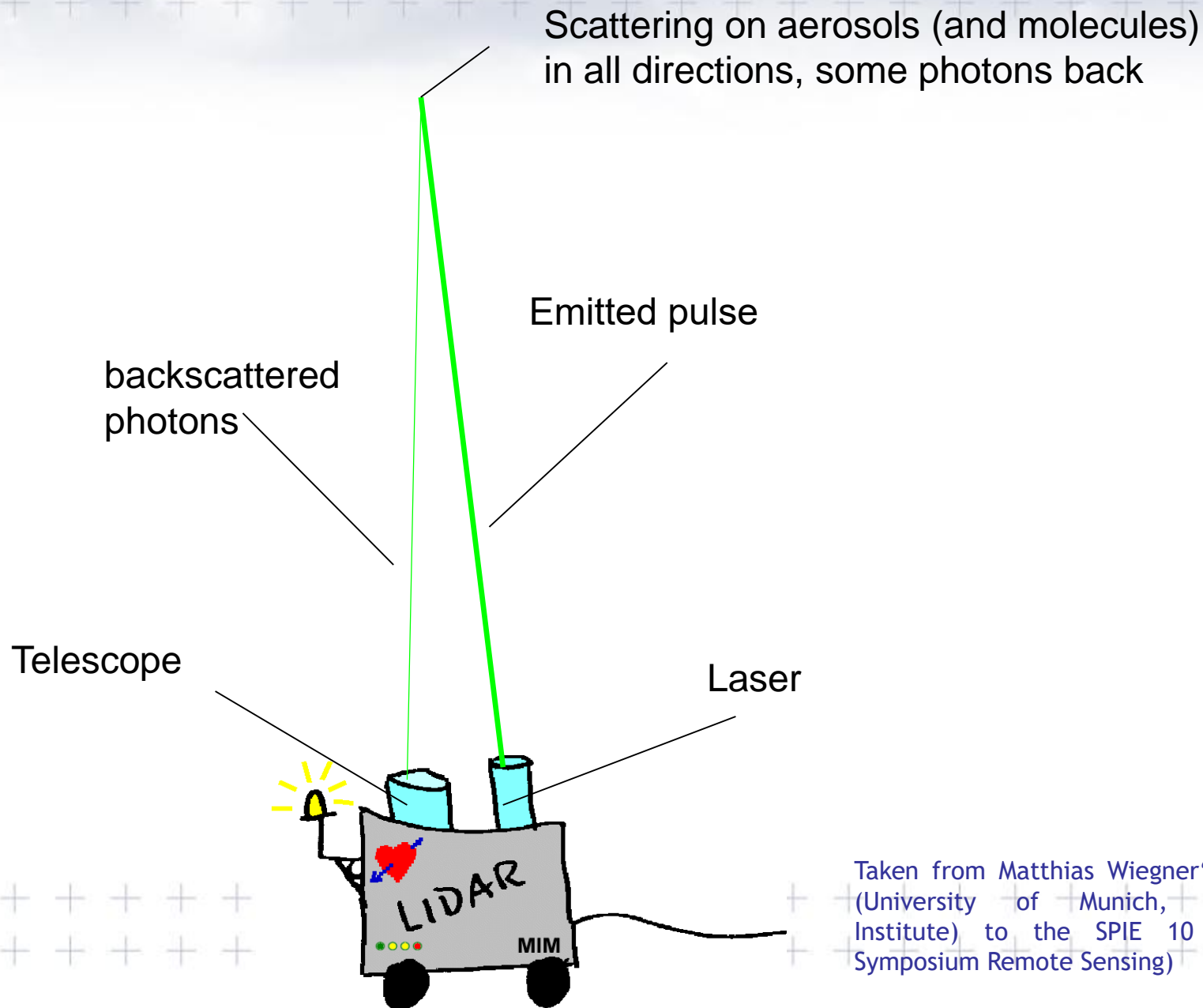
Taken from Matthias Wiegner's presentation (University of Munich, Meteorological Institute) to the SPIE 10 (International Symposium Remote Sensing)



Taken from Matthias Wiegner's presentation (University of Munich, Meteorological Institute) to the SPIE 10 (International Symposium Remote Sensing)



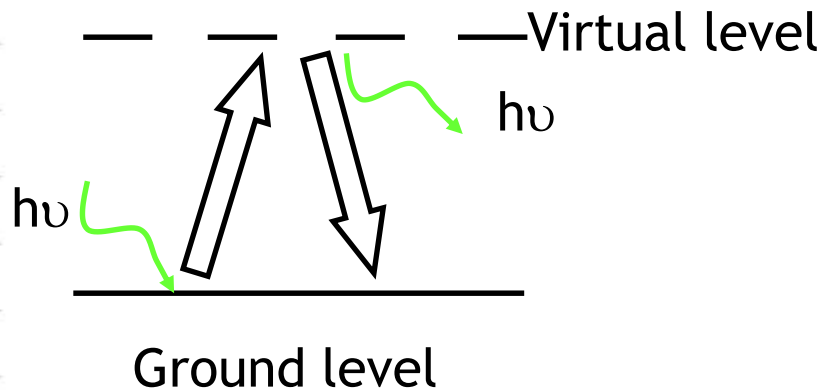
Taken from Matthias Wiegner's presentation (University of Munich, Meteorological Institute) to the SPIE 10 (International Symposium Remote Sensing)



Taken from Matthias Wiegner's presentation (University of Munich, Meteorological Institute) to the SPIE 10 (International Symposium Remote Sensing)

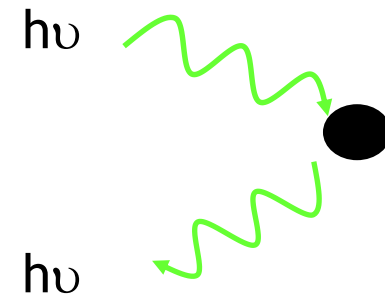
- Rayleigh Scattering

“Laser radiation elastically scattered from atoms or molecules is observed with no change of frequency”



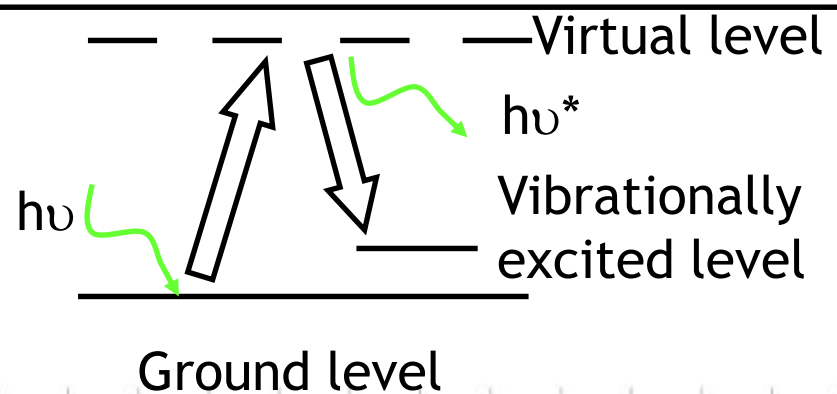
- Mie Scattering

“Laser radiation elastically scattered from small particulates or aerosols (of size comparable to wavelength of radiation) is observed with no change in frequency”



- Raman Scattering

“Laser radiation inelastically scattered from molecules is observed with a frequency shift characteristic of the molecule ($h\nu - h\nu^* = E$)”



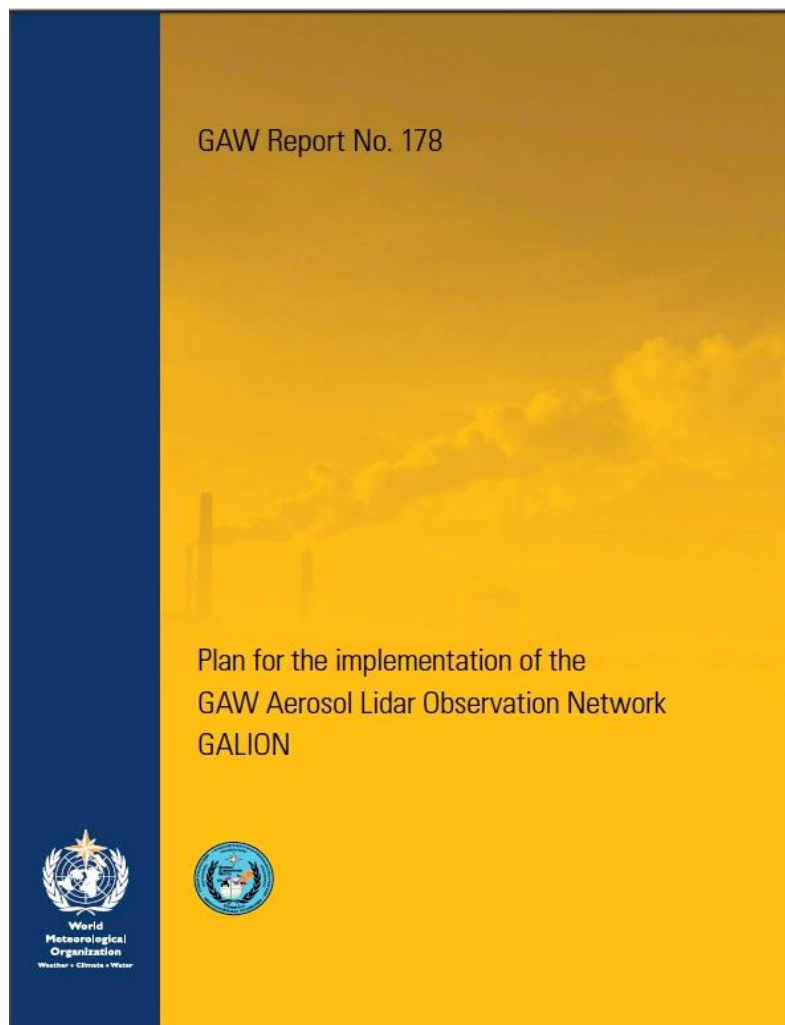


Lidar-Barcelona (UPC)
Raman Lidar
EARLINET-SPALINET



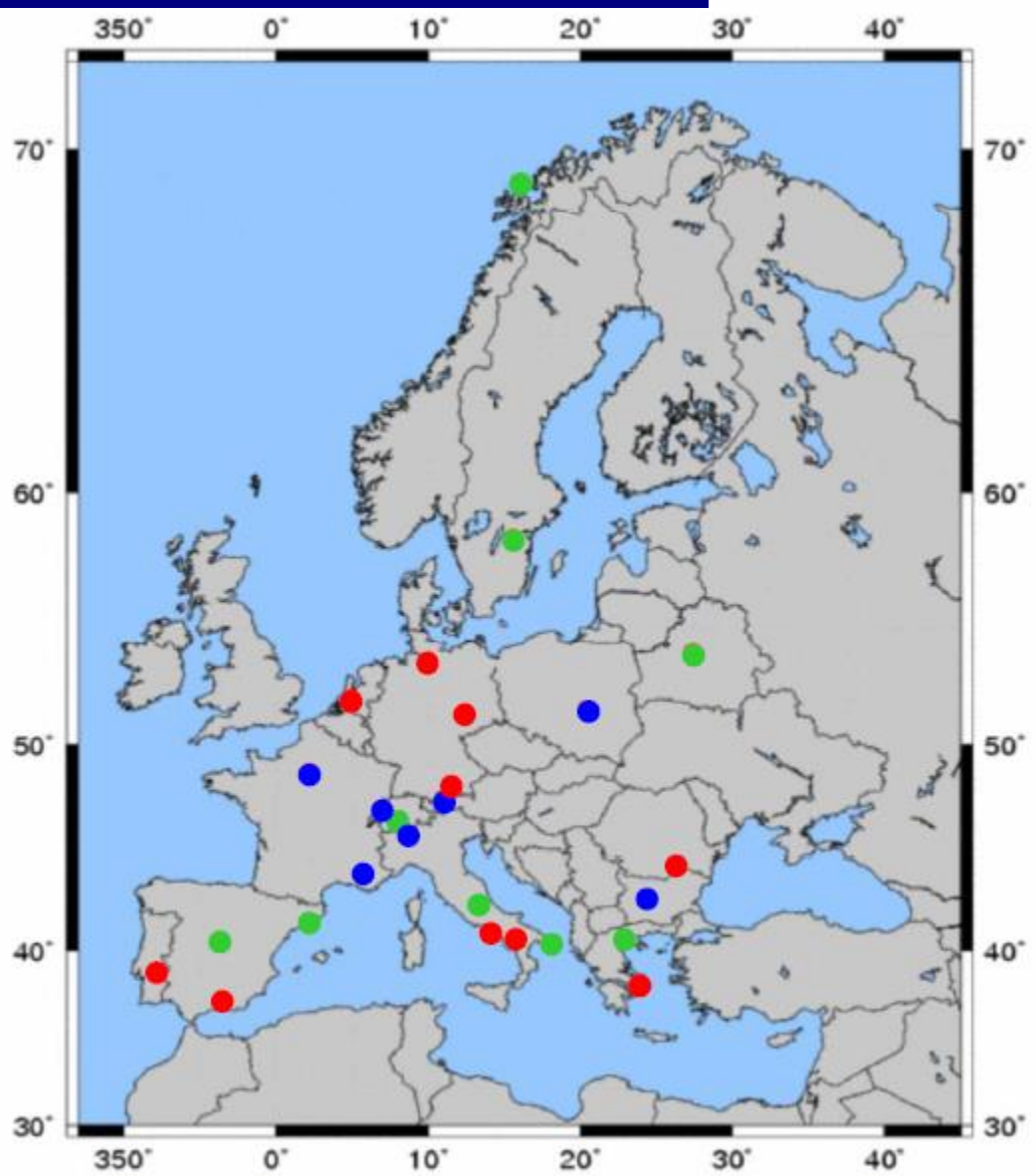
Lidar-Tenerife (INTA-AEMET); Elastic lidar
MPLNET

GAW Atmospheric Lidar Network (GALION)



<ftp://ftp.wmo.int/Documents/PublicWeb/arep/gaw/gaw178-galion-27-Oct.pdf>





EARLINET

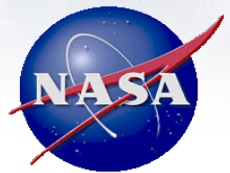
EARLINET (European Aerosol Research Lidar NETWORK) is a network of advanced lidar stations distributed over Europe with the main goal to provide a comprehensive, quantitative, and statistically significant data base for the aerosol distribution on a continental scale. EARLINET provides independent measurements of aerosol extinction and backscatter, and retrieval of aerosol microphysical properties.

10 EARLINET stations are equipped also with sunphotometers (they are part of AERONET).

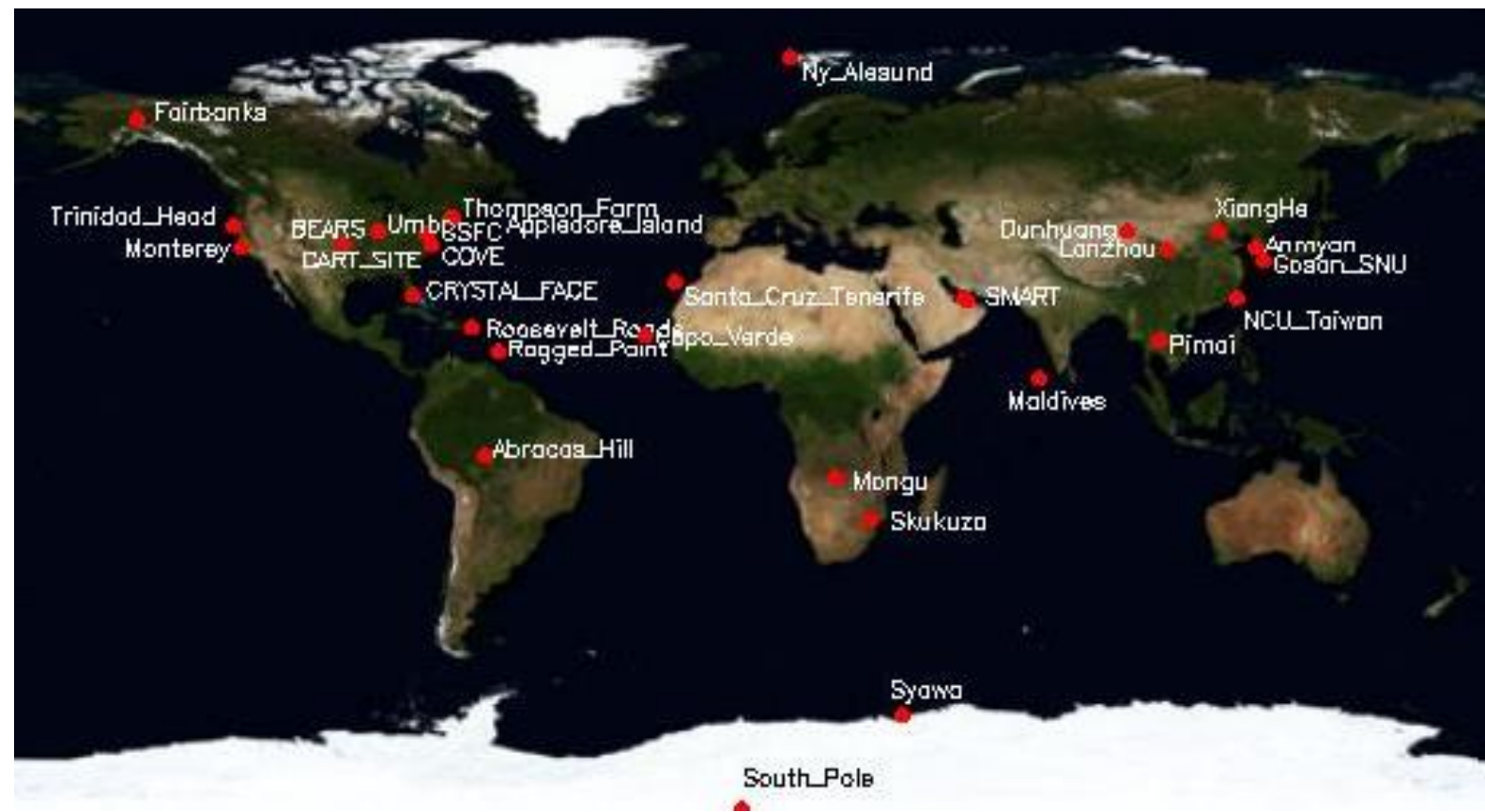
- 26 lidar stations**
 - 10 multiwavelength Raman lidar stations
 - backscatter (355, 532 and 1064 nm) + extinction (355 and 532 nm) + depol ratio (532 nm)
 - 9 Raman lidar stations
 - 7 single backscatter lidar stations

Aerosol lidar (MPLNet)

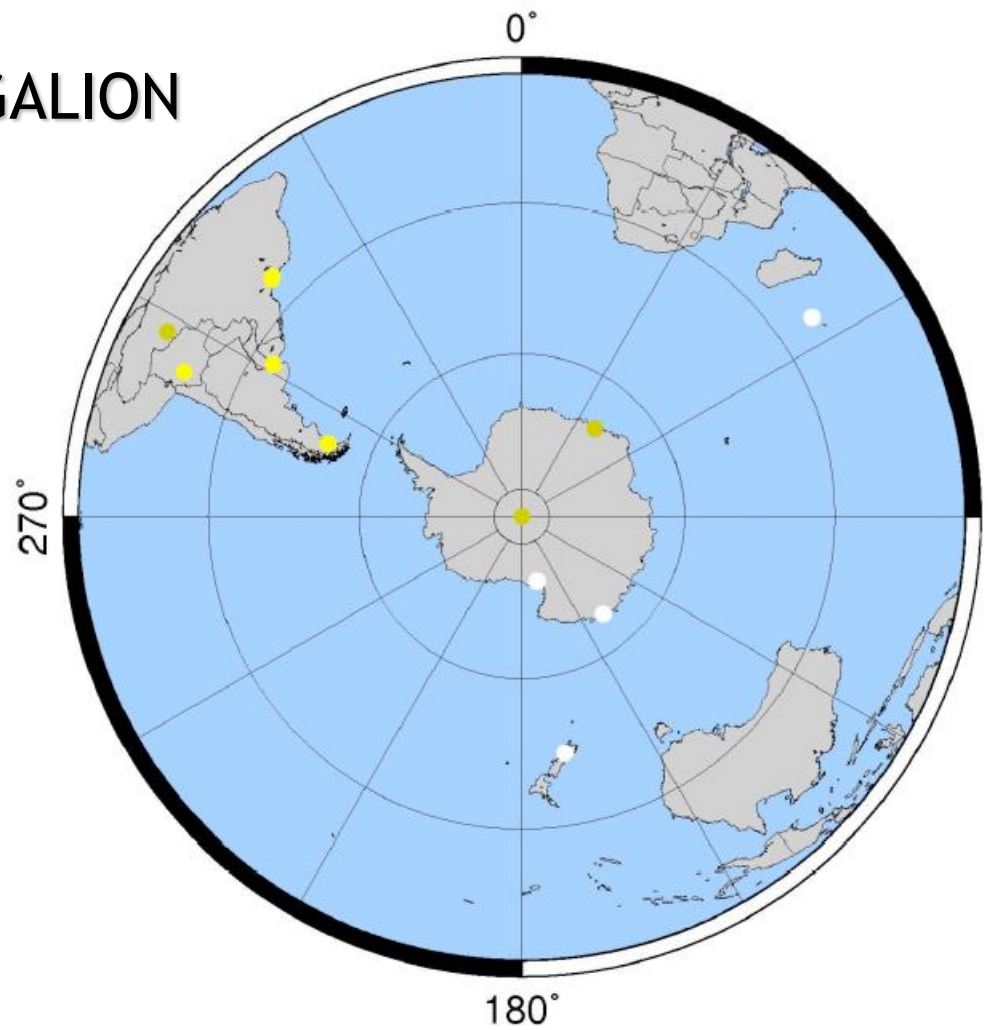
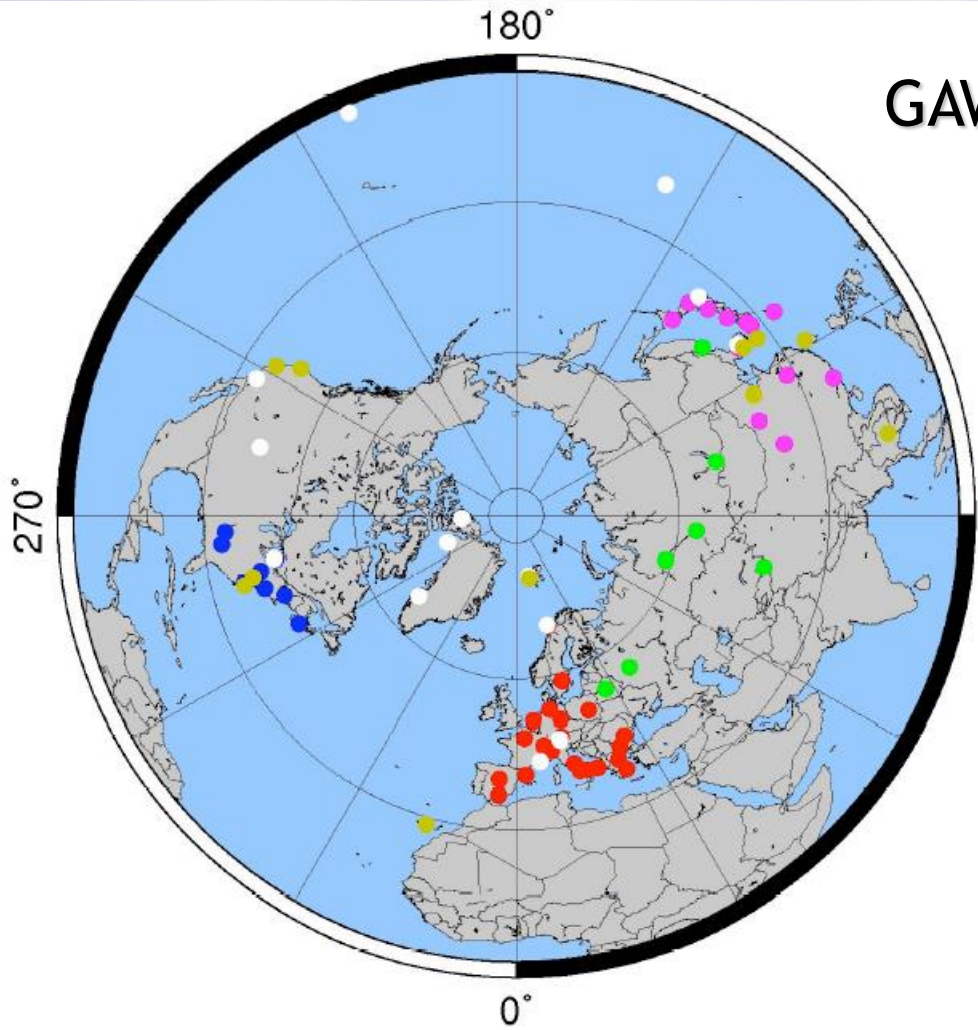
<http://mplnet.gsfc.nasa.gov/>



523 nm MPLNET
Automatized since July 2005

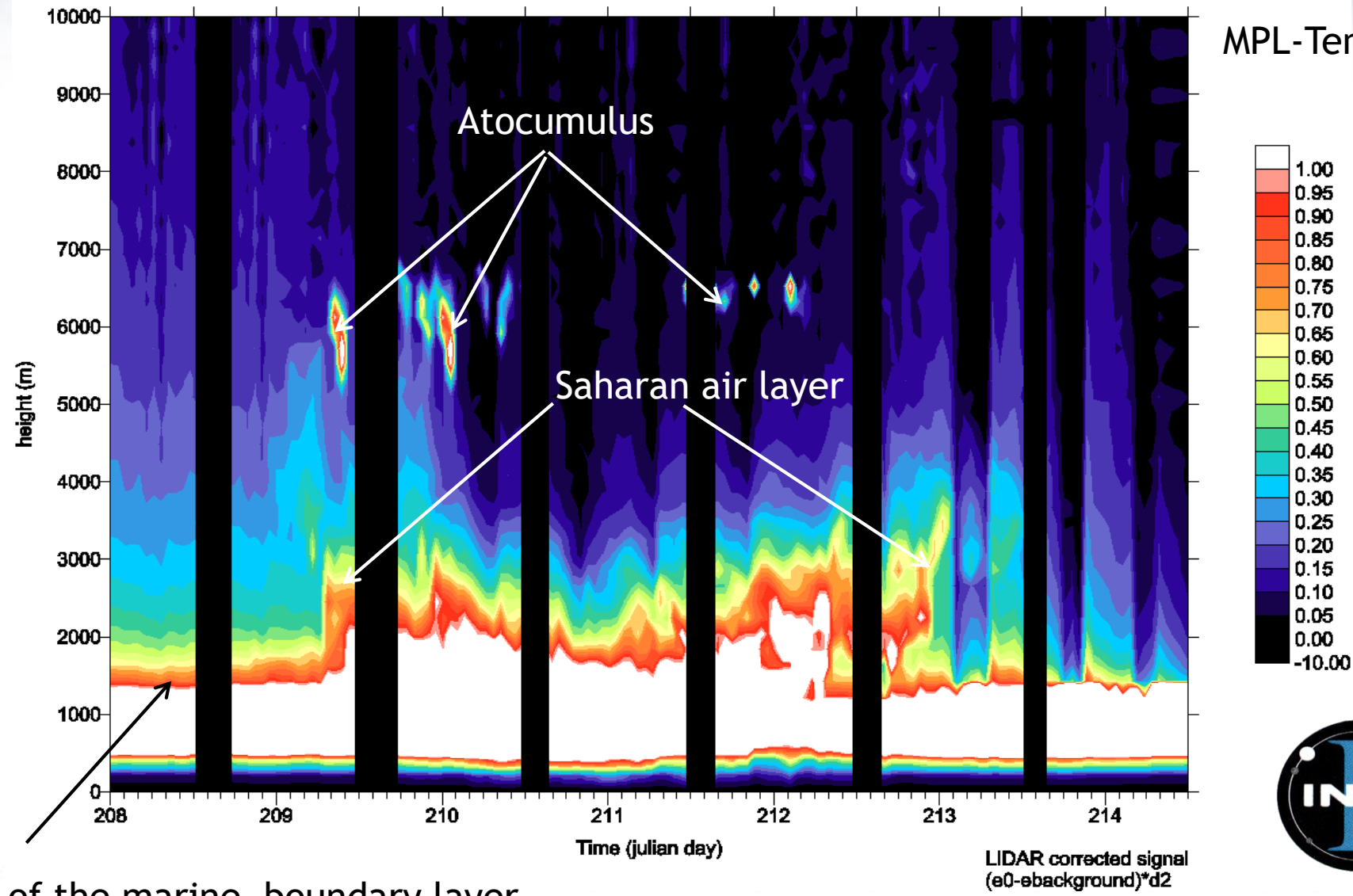


GAW-GALION



Distribution of stations as available through the cooperation between existing networks: **AD-NET** , **ALINE** , **CISLiNet** , **EARLINET** , **MPLNET** , **NDACC** , **REALM** .

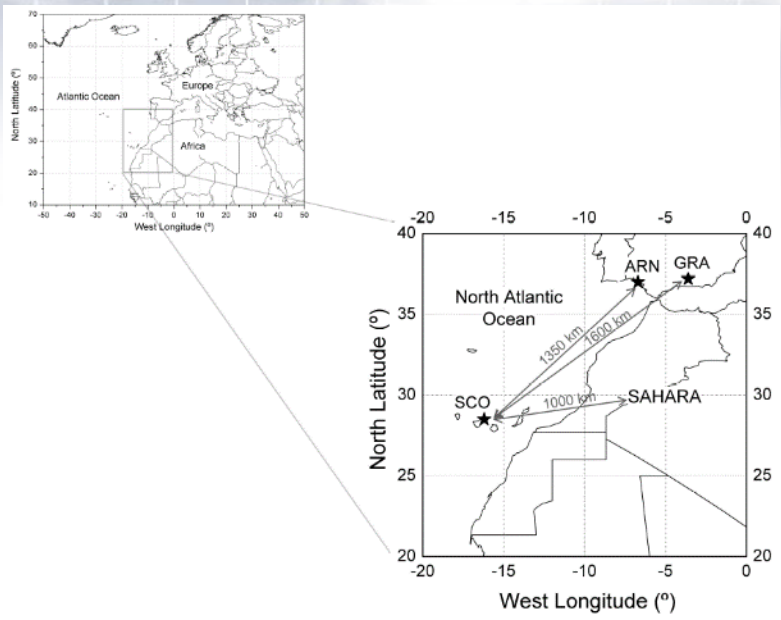
DUST EVENT 28 JULY - 2 AUGUST 2002



MPL-Tenerife

Top of the marine boundary layer

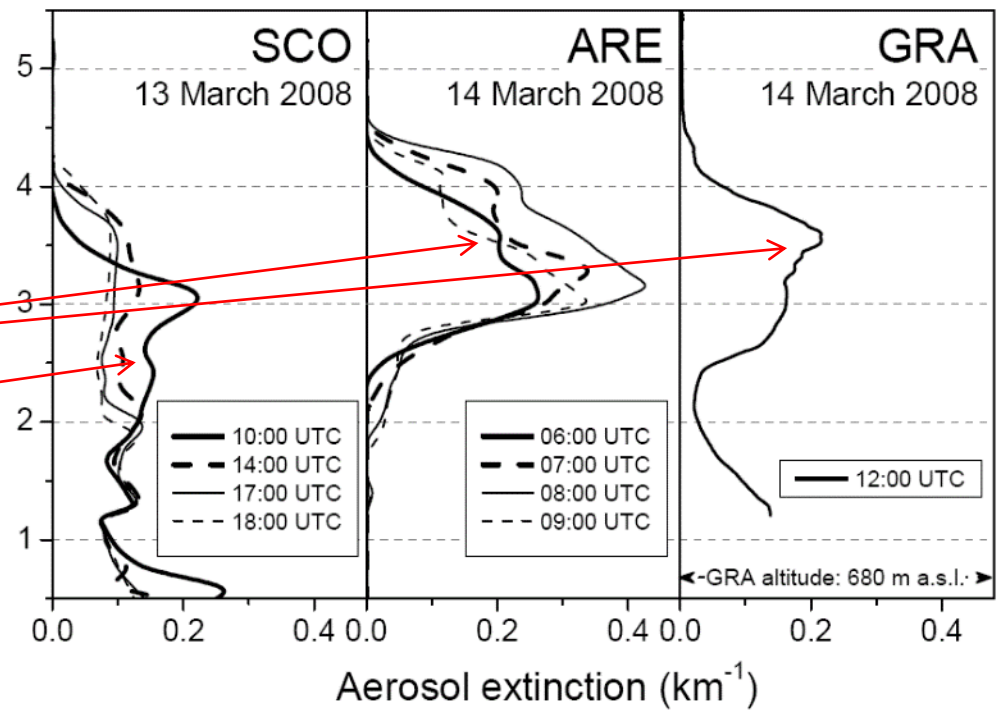
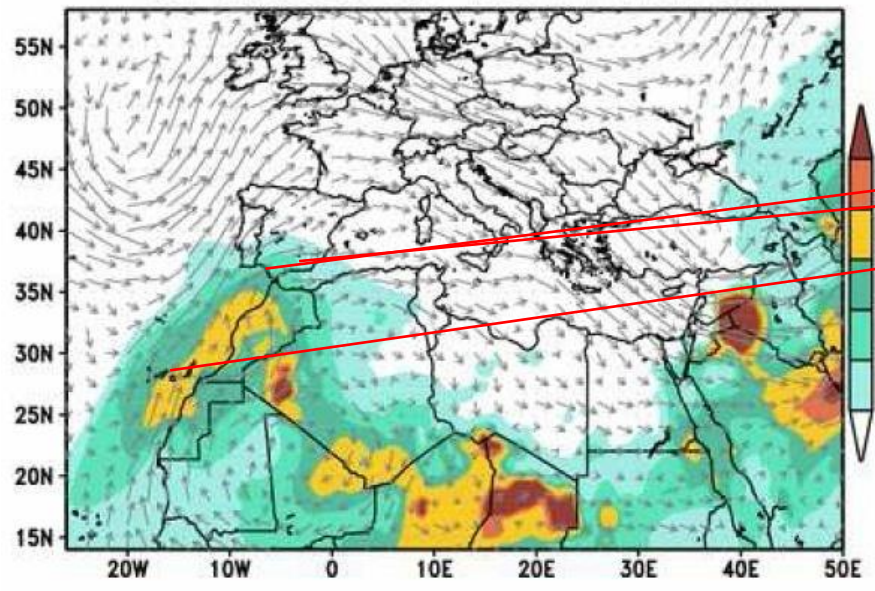




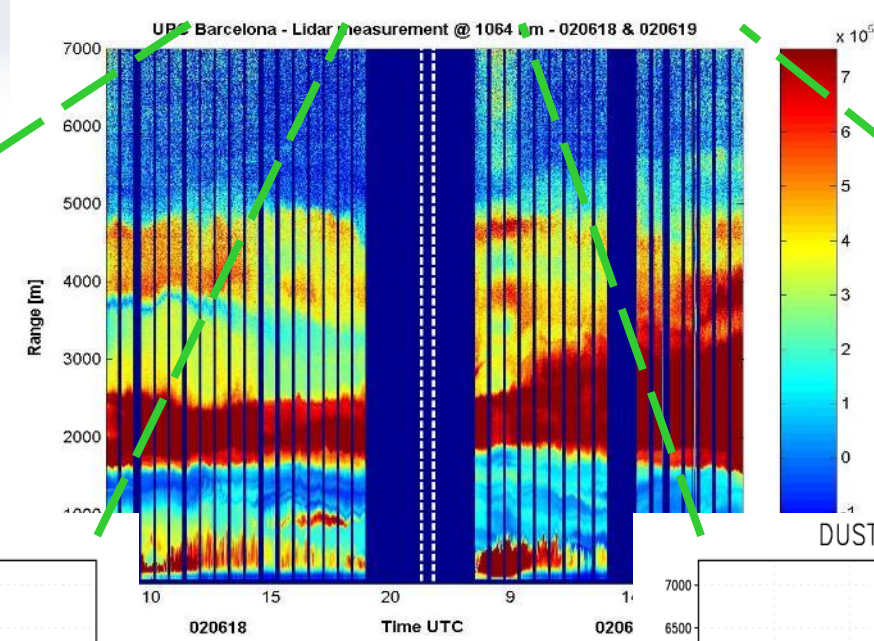
A case study of dust transport from Canary Islands to Iberian Peninsula

Córdoba-Jabonero et al., ACP Discuss., 2010

BSC/DREAM Dust Loading (g/m^2) and 3000m Wind Oh forecast for 12z 14 MAR 08



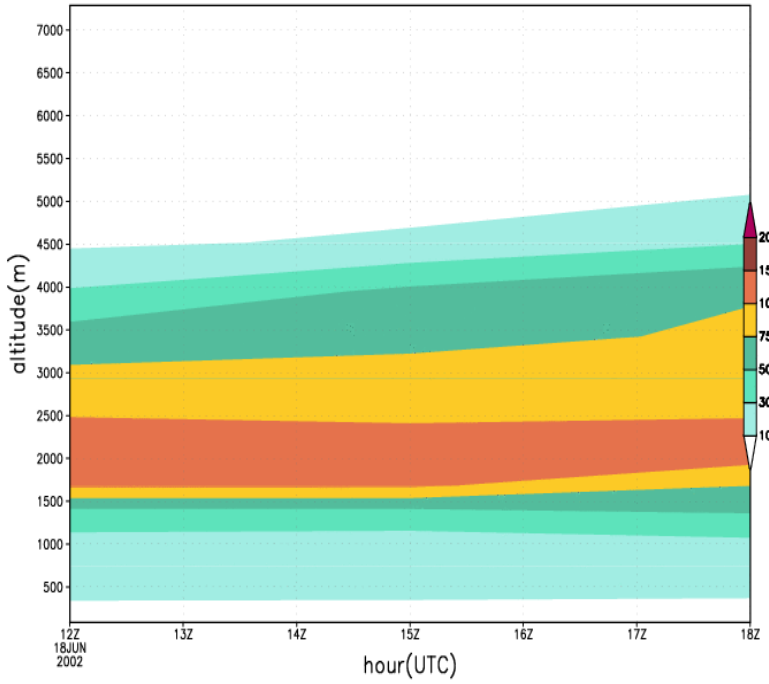
Barcelona lidar vs DREAM BSC



EARLINET: Lidar-UPC, Barcelona

18-19 June 2002

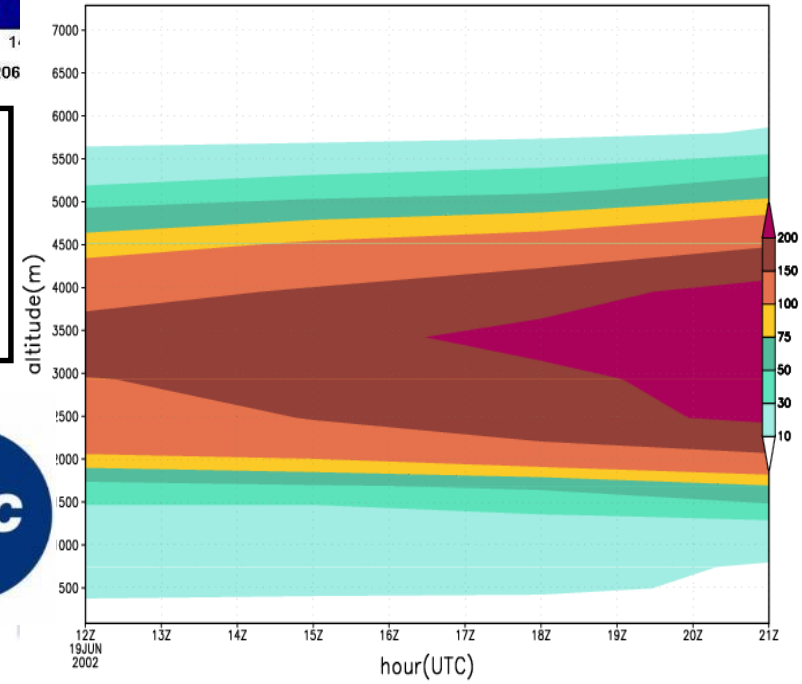
DUST CONC. ($\mu\text{g}/\text{m}^3$) 18 JUN 2002



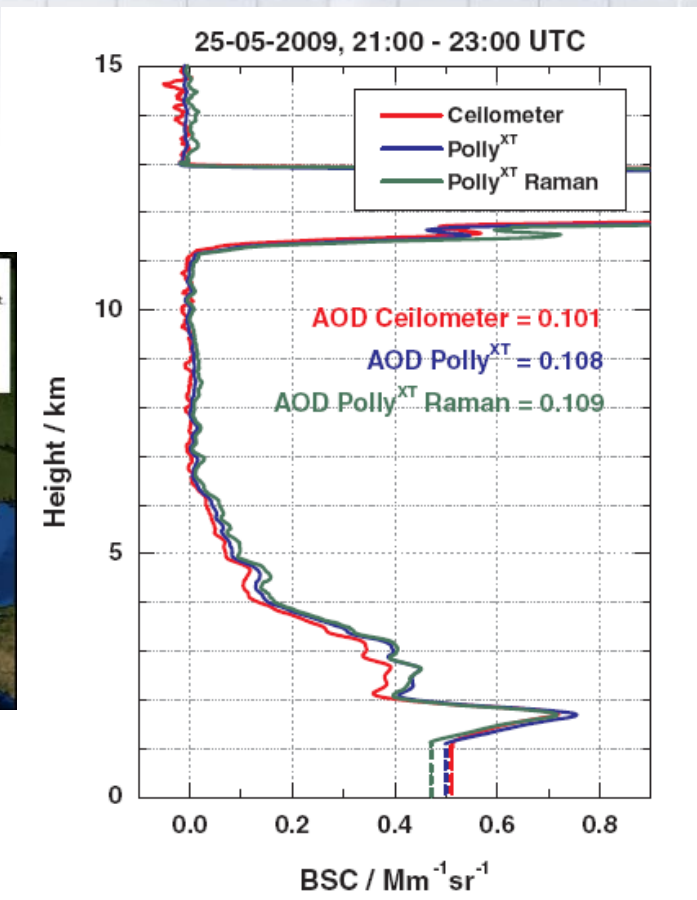
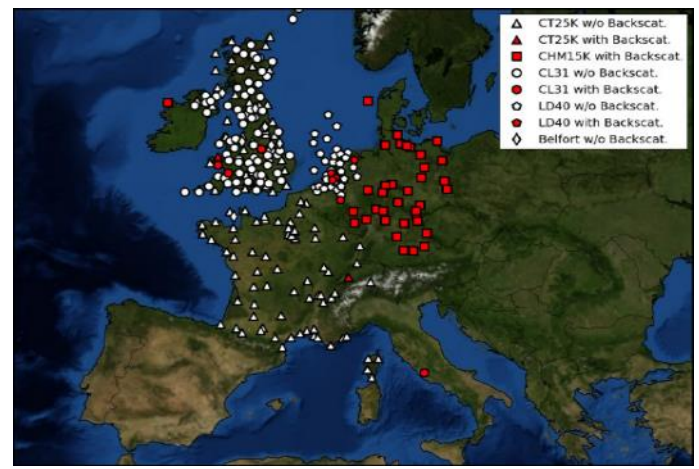
Vertical dust distribution validation: AIRLINET-DREAM



DUST CONC. ($\mu\text{g}/\text{m}^3$) 19 JUN 2002

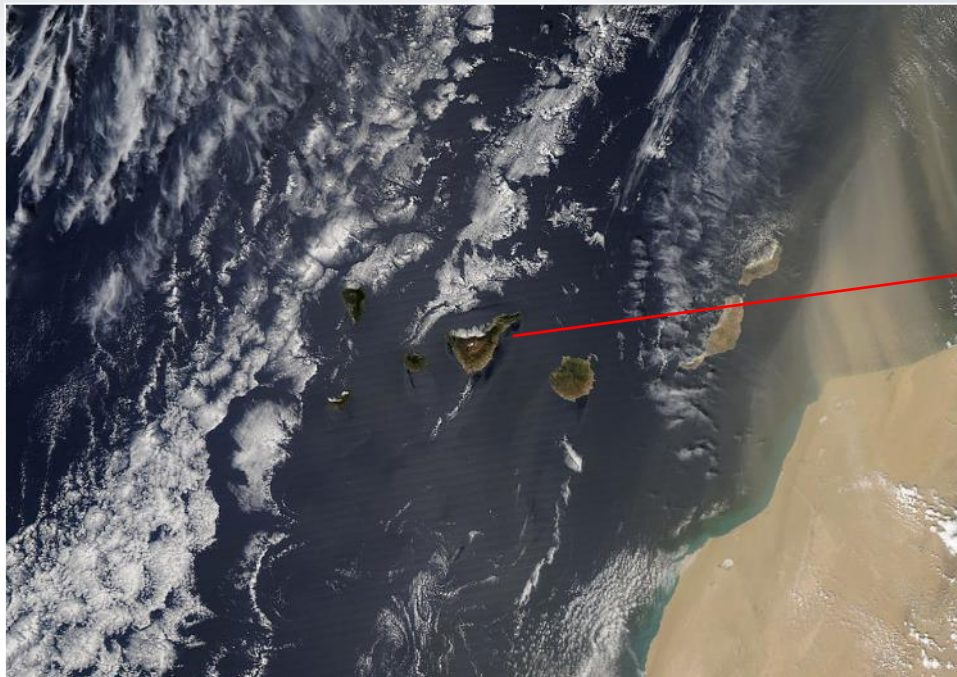


Met Services are replacing cloud-base ceilometer networks by aerosol backscatter profiling ceilometers (IR wavelength).
Objective: To monitor MLD (Mixing Layer Depth) based on several hundred profiling ceilometers (100km sampling)



Heese et al., Atmos. Mes. Tech. 2010, Ceilometer-lidar inter-comparison: backscatter coefficient retrieval and signal-to-noise ratio determination

Optimal for deserty areas !!



Viasala Ceilometer
CL-51

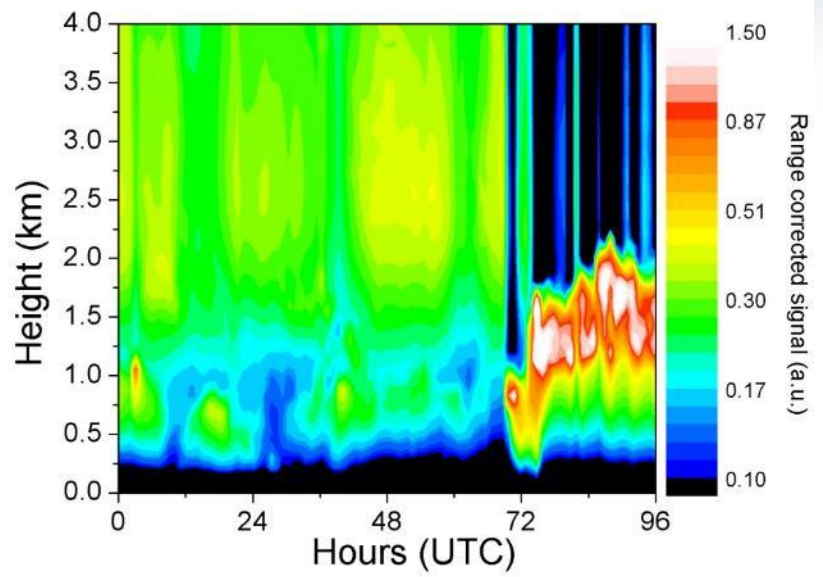
MicroPulse Lidar and Ceilometer inter-comparison during Saharan dust intrusions over the Canary Islands

Y. Hernández, S. Alonso-Pérez, E. Cuevas, C. Camino, R. Ramos, J. de Bustos, C. Marrero, C. Córdoba-Jabonero and M. Gil (2011)

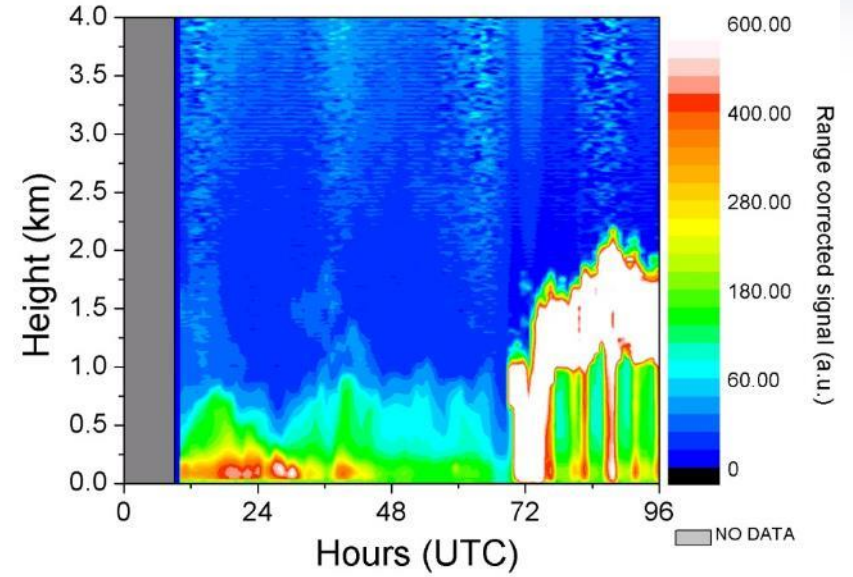
Campaign performed from January to March 2011 in Tenerife island

Ground-based remote sensing

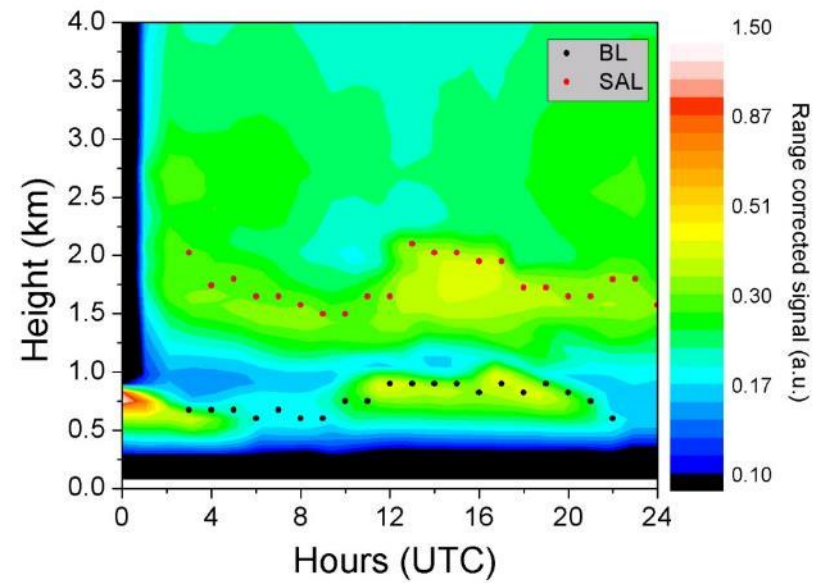
MPL-3 - Sta. Cruz de Tenerife. Mar 31- Apr 3, 2011



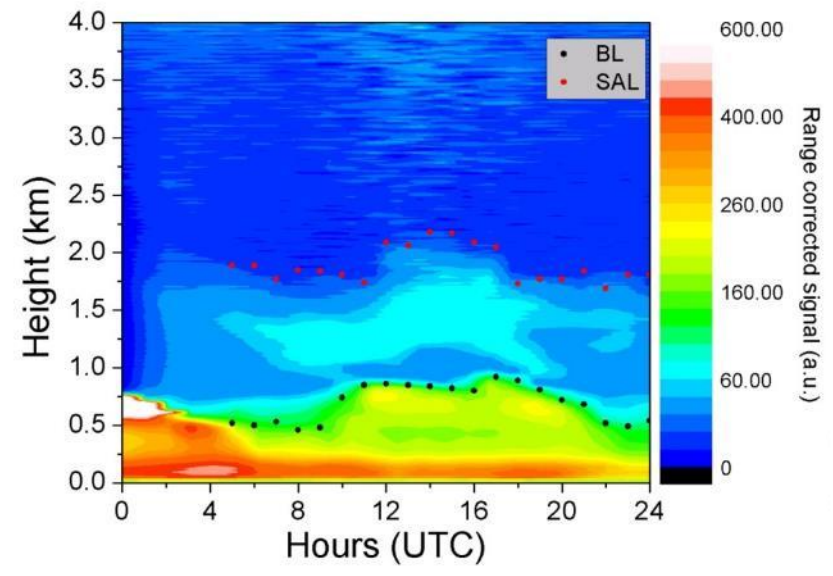
CL51 - Sta. Cruz de Tenerife. Mar 31- Apr 3, 2011



MPL-3 - Sta. Cruz de Tenerife. Feb 24, 2011



CL51 - Sta. Cruz de Tenerife. Feb 24, 2011



dust, aerosols and pollutants

in-situ observations

PM_{10} and $PM_{2.5}$ levels

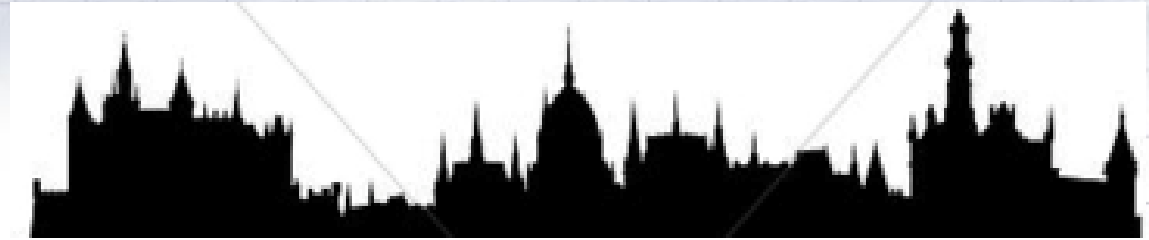
PM_{10} and $PM_{2.5}$ composition

complementary observations

observation network



dust air quality



Recommended priorities

Level 1 (max priority) - PM_{10} and $PM_{2.5}$ levels - automatic methods

Level 1 (max priority) - meteorology (wind, T, RH, P, rain)

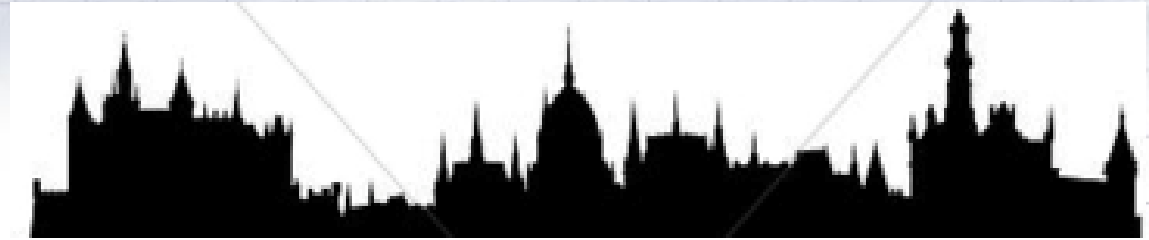
Level 2 - PM_{10} and $PM_{2.5}$ levels - complementary gravimetric method

Level 3 - gaseous pollutants: NO_x , SO_2 , CO,...

Level 4 - PM_{10} and $PM_{2.5}$ chemical composition



dust air quality



Recommended priorities

Level 1

- PM_{10} and $PM_{2.5}$ levels - automatic methods

Level 1

- meteorology (wind, T, RH, P, rain)

Level 2

- PM_{10} and $PM_{2.5}$ levels - complementary gravimetric method

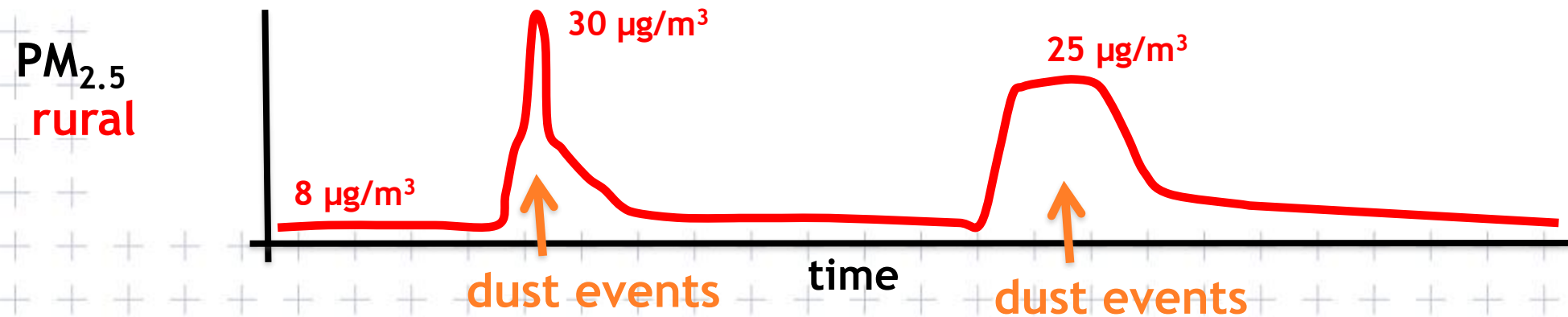
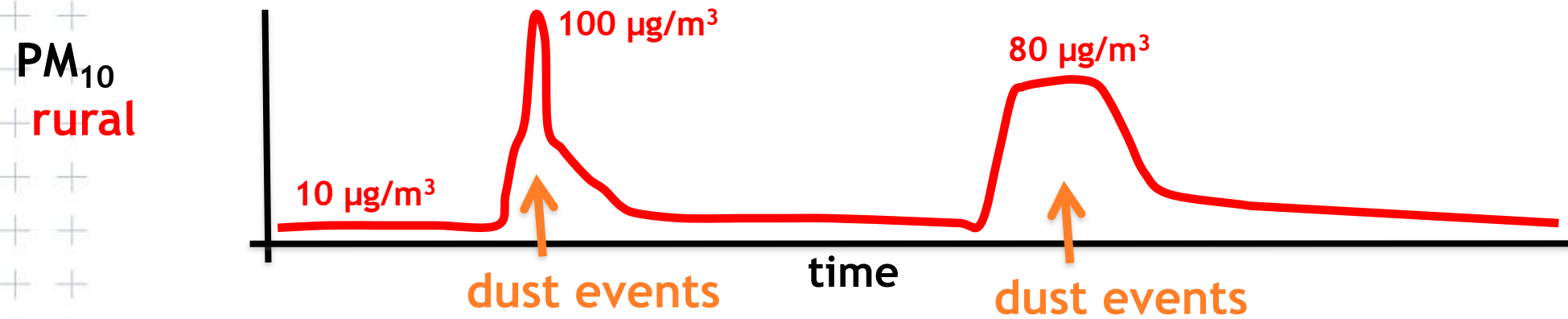
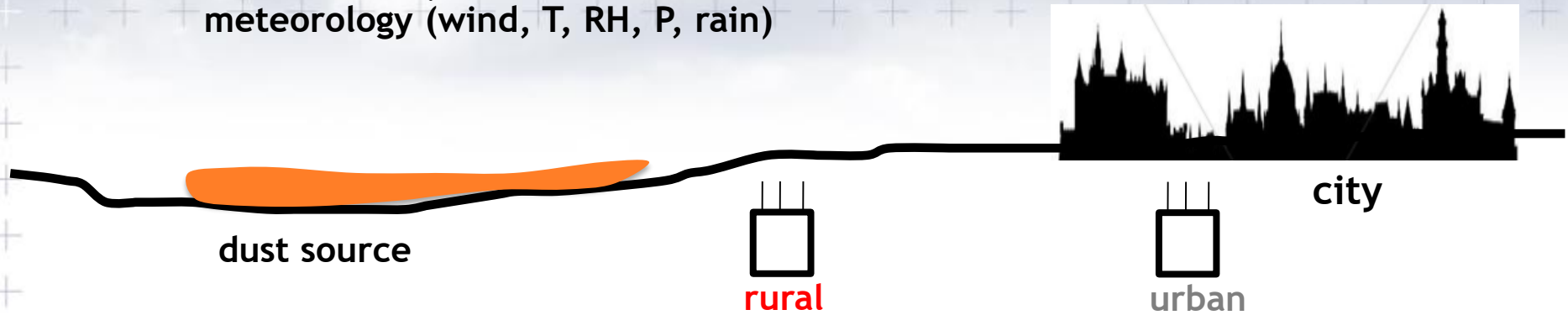
Level 3

- gaseous pollutants: NO_x , SO_2 , CO ,...

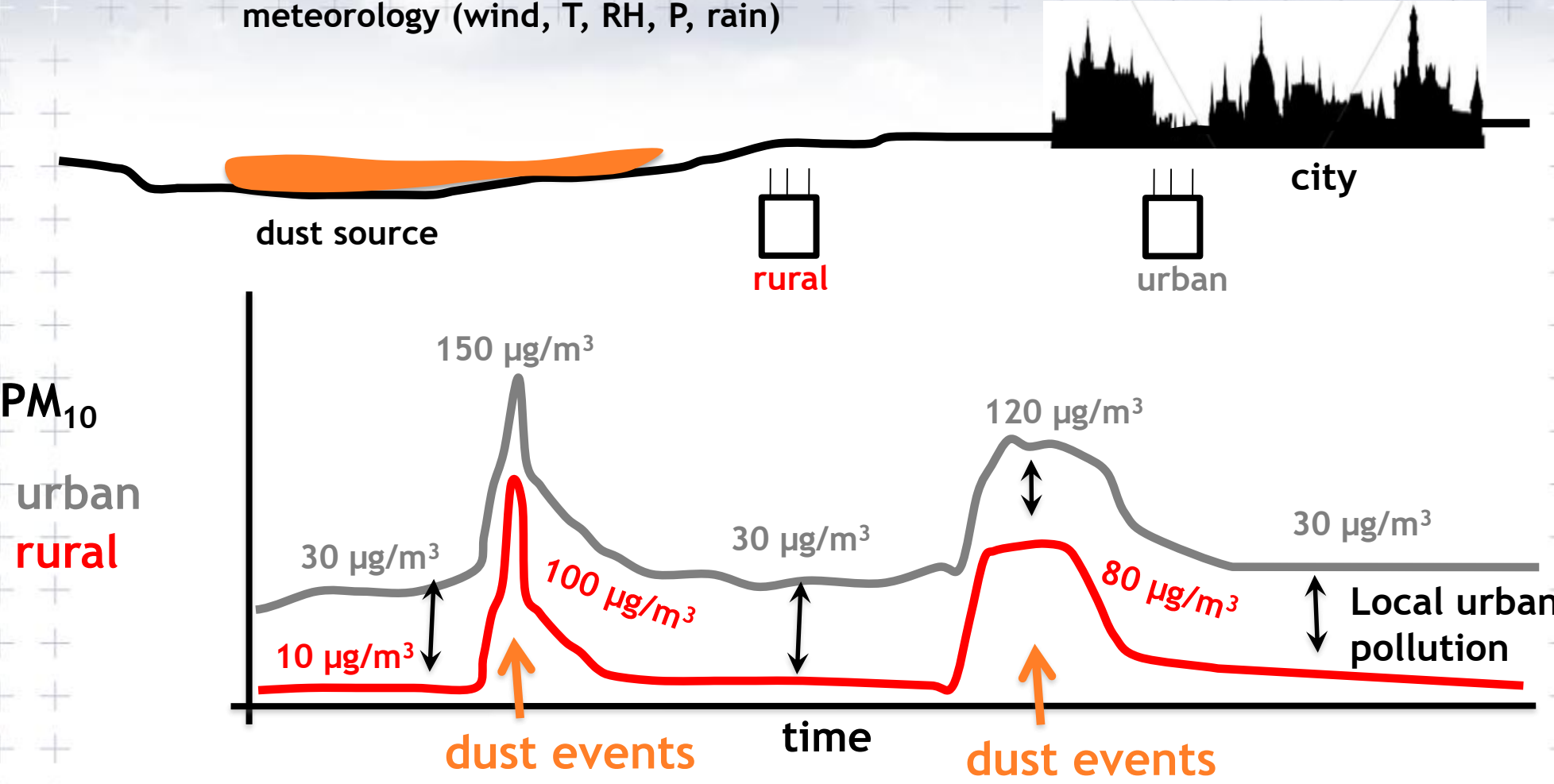
Level 4

- PM_{10} and $PM_{2.5}$ chemical composition

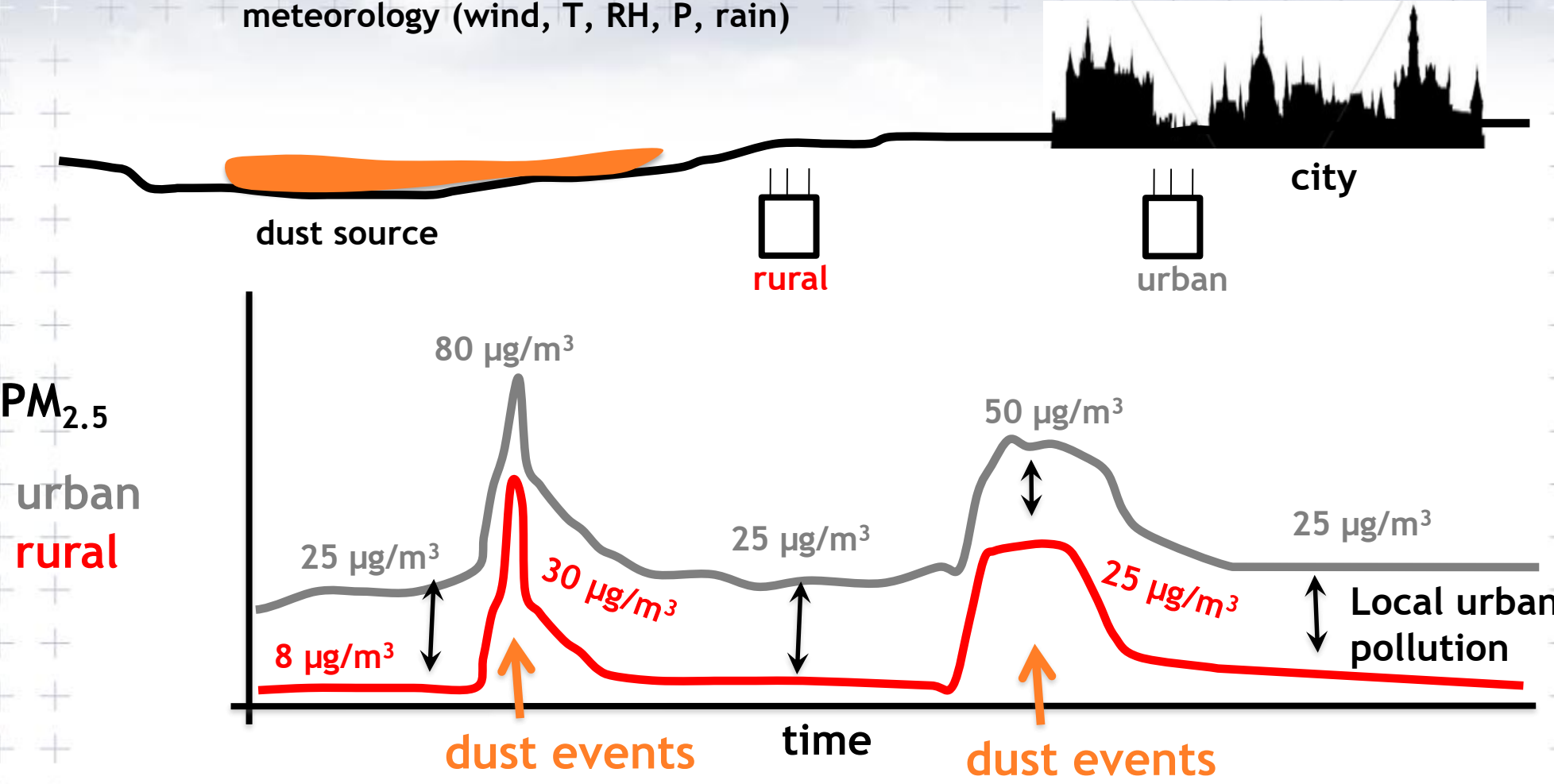
Level 1 PM_{10} and $PM_{2.5}$ - automatic methods meteorology (wind, T, RH, P, rain)



Level 1 PM_{10} and $PM_{2.5}$ - automatic methods meteorology (wind, T, RH, P, rain)

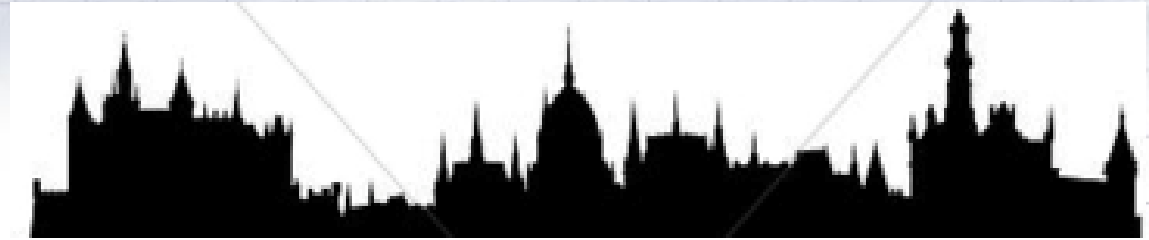


Level 1 PM_{10} and $PM_{2.5}$ - automatic methods meteorology (wind, T, RH, P, rain)





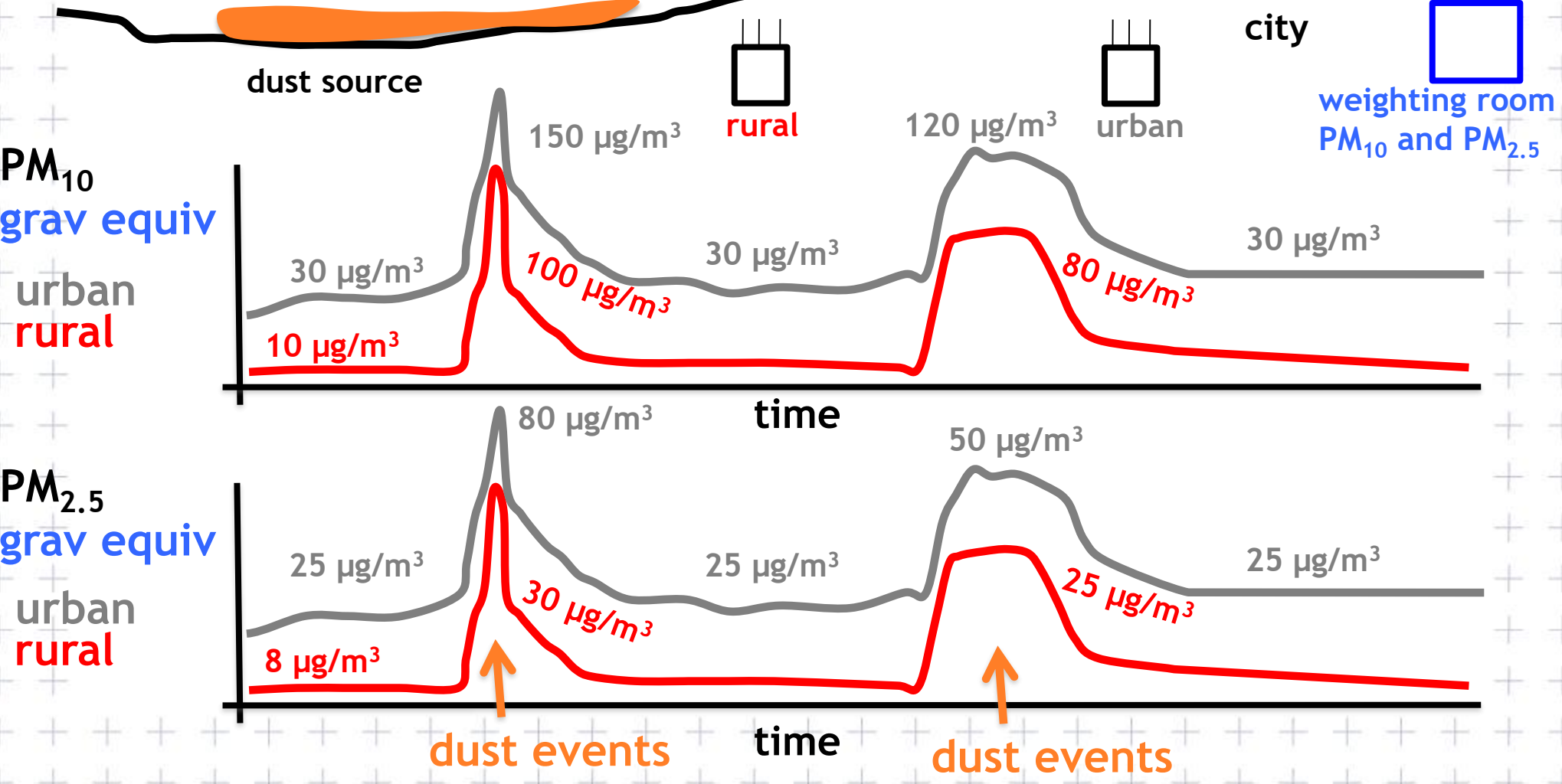
dust air quality

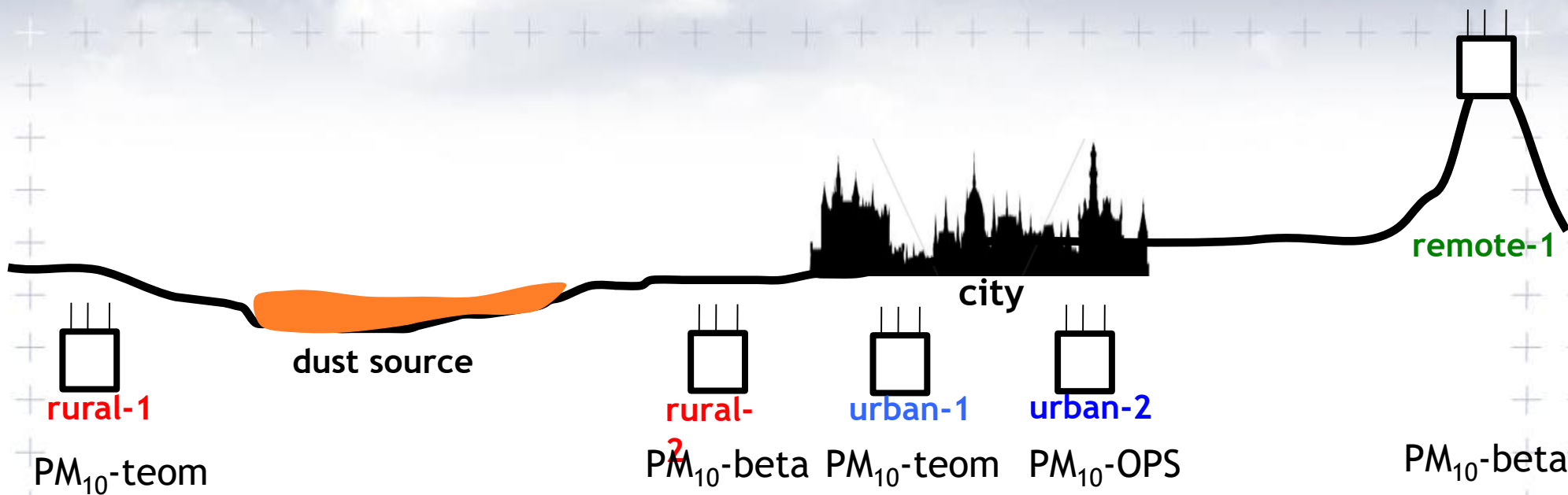


Recommended priorities

- Level 1 - PM_{10} and $PM_{2.5}$ levels - automatic methods
- Level 1 - meteorology (wind, T, RH, P, rain)
- Level 2** - PM_{10} and $PM_{2.5}$ levels - complementary gravimetric method
- Level 3 - gaseous pollutants: NO_x , SO_2 , CO ,...
- Level 4 - PM_{10} and $PM_{2.5}$ chemical composition

Level 2 PM_{10} and $PM_{2.5}$ - automatic methods
 meteorology (wind, T, RH, P, rain)
 PM_{10} and $PM_{2.5}$ - gravimetric reference method





Standardized data

raw data

rural-1

$$PM_{10} \text{ (grav equiv)} = 1.27 PM_{10} \text{ (TEOM)}$$

rural-2

$$PM_{10} \text{ (grav equiv)} = 0.71 PM_{10} \text{ (BETA)}$$

urban-1

$$PM_{10} \text{ (grav equiv)} = 0.95 PM_{10} \text{ (TEOM)}$$

urban-2

$$PM_{10} \text{ (grav equiv)} = 0.79 PM_{10} \text{ (OPS)}$$

remote-1

$$PM_{10} \text{ (grav equiv)} = 0.67 PM_{10} \text{ (BETA)}$$



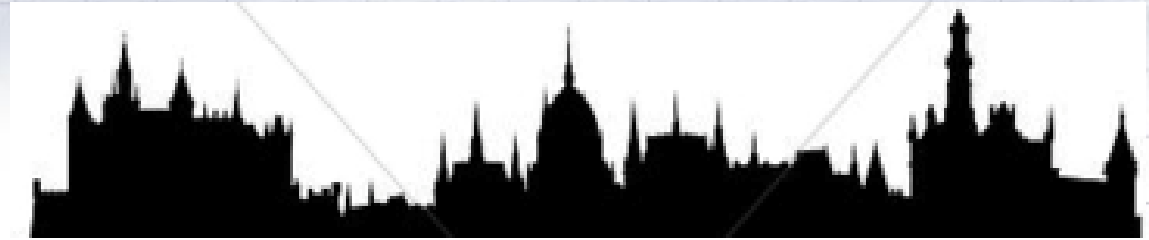
samplers of PM_{10} and $PM_{2.5}$

room of conditioning and weighting filters

1 month in summer (30 days) sampling
1 month in winter (30 days) sampling
at each station



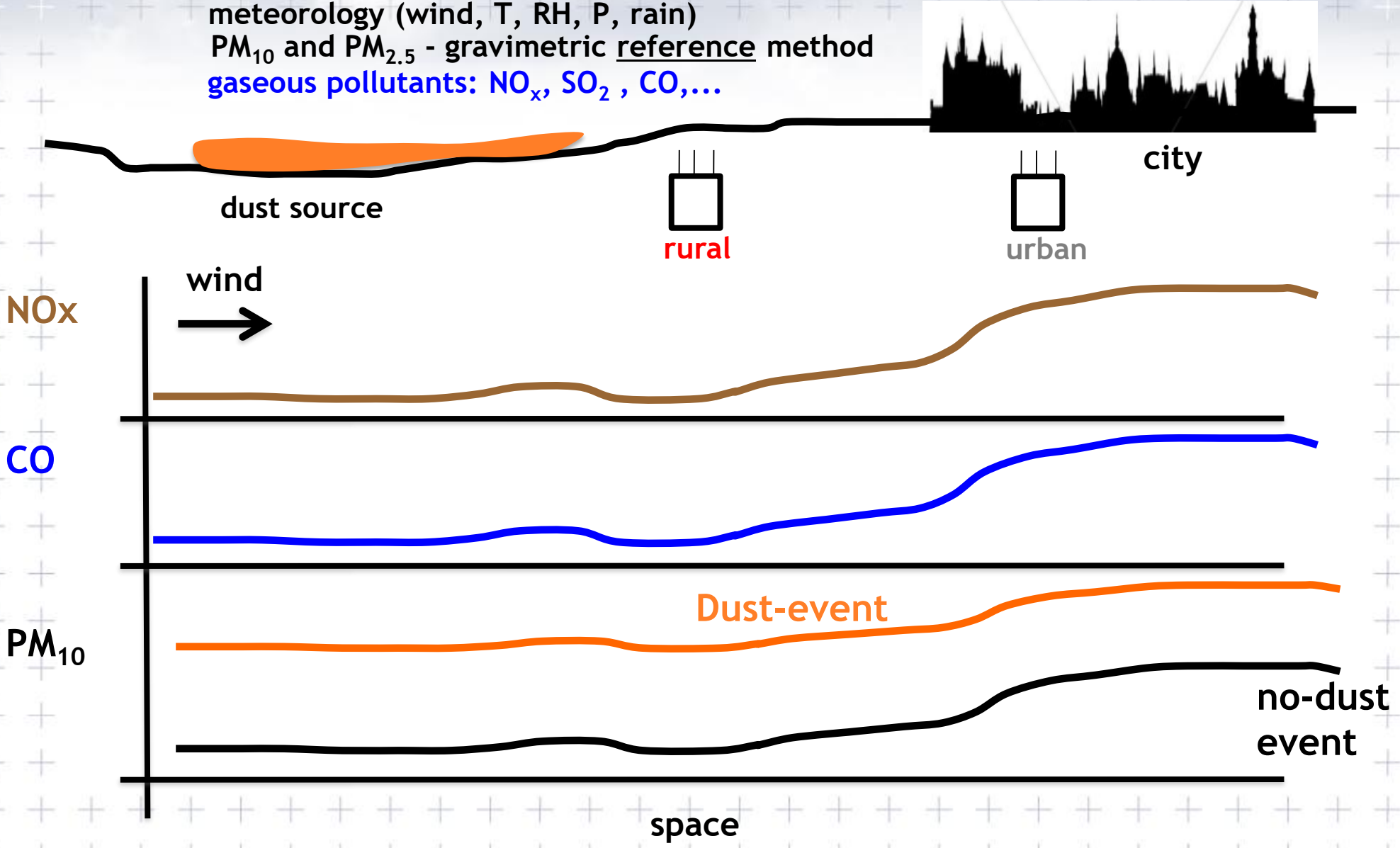
dust air quality



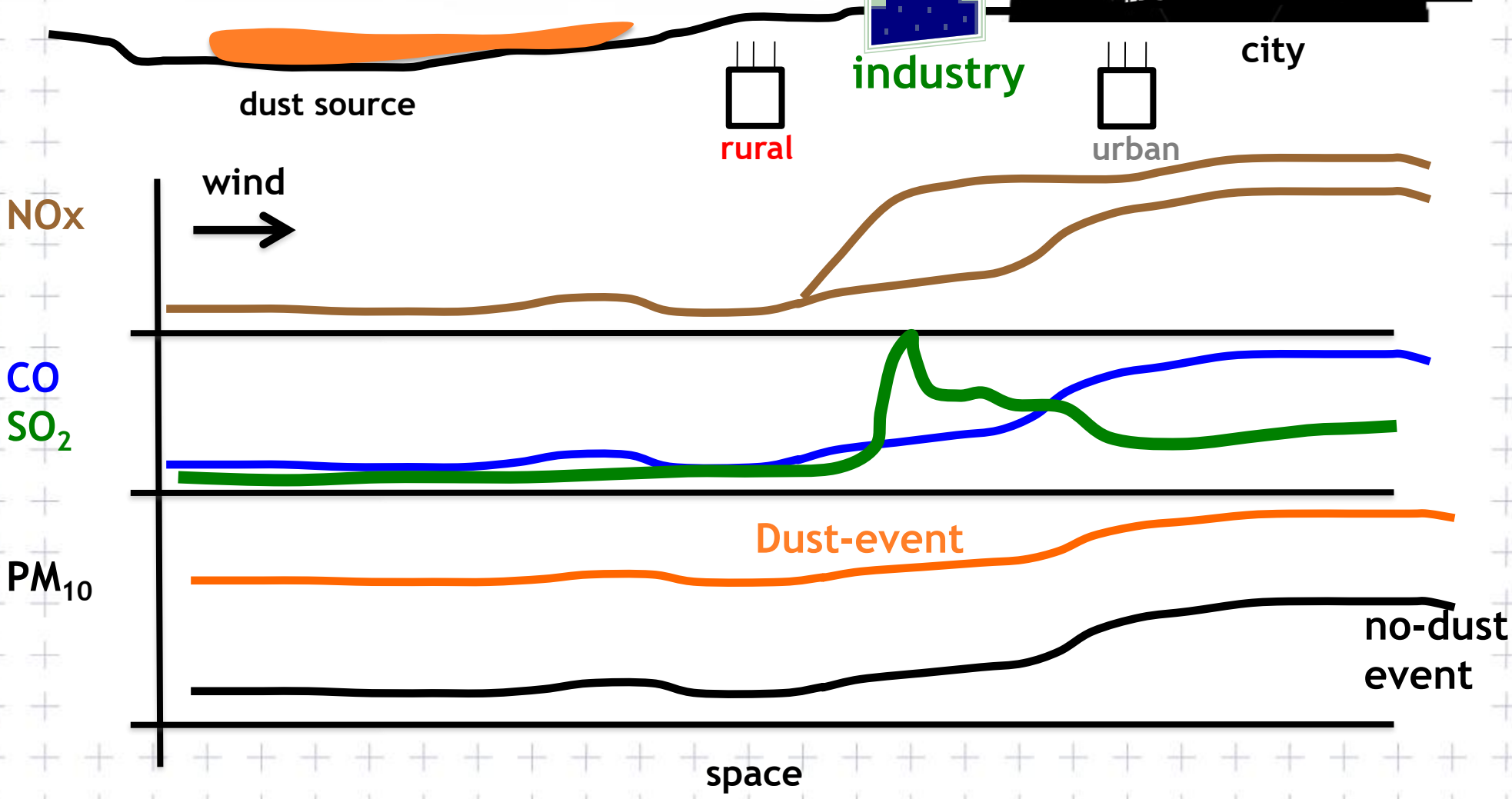
Recommended priorities

- Level 1 (max) - PM_{10} and $PM_{2.5}$ levels - automatic methods
- Level 1 (max) - meteorology (wind, T, RH, P, rain)
- Level 2 - PM_{10} and $PM_{2.5}$ levels - complementary gravimetric method
- Level 3 - gaseous pollutants: NO_x , SO_2 , CO ,...**
- Level 4 - PM_{10} and $PM_{2.5}$ chemical composition

Level 3 PM_{10} and $PM_{2.5}$ - automatic methods
meteorology (wind, T, RH, P, rain)
 PM_{10} and $PM_{2.5}$ - gravimetric reference method
gaseous pollutants: NO_x , SO_2 , CO ,...

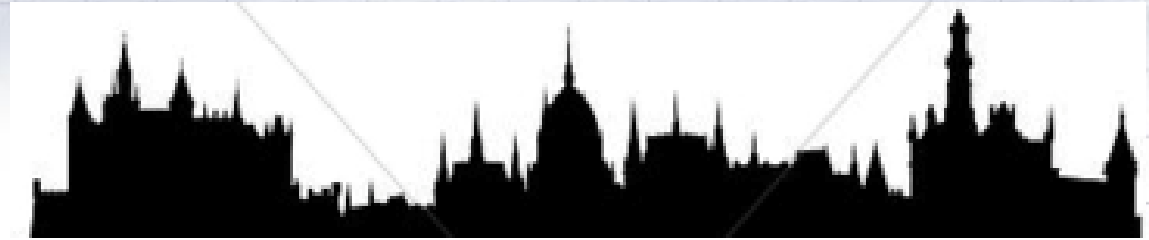


Level 3 PM_{10} and $PM_{2.5}$ - automatic methods
meteorology (wind, T, RH, P, rain)
 PM_{10} and $PM_{2.5}$ - gravimetric reference method
gaseous pollutants: NO_x , SO_2 , CO, ...





dust air quality



Recommended priorities

Level 1 (max) - PM_{10} and $PM_{2.5}$ levels - automatic methods

Level 1 (max) - meteorology (wind, T, RH, P, rain)

Level 2 - PM_{10} and $PM_{2.5}$ levels - complementary gravimetric method

Level 3 - gaseous pollutants: NO_x , SO_2 , CO ,...

Level 4 - PM_{10} and $PM_{2.5}$ chemical composition

Level 4 PM_{10} and $PM_{2.5}$ - automatic methods
 meteorology (wind, T, RH, P, rain)
 PM_{10} and $PM_{2.5}$ - gravimetric reference methods
 gaseous pollutants: NO_x , SO_2 , CO, ...
 PM_{10} and $PM_{2.5}$ - chemical composition

